STRUCTURAL INTERACTIONS AND FERMENTATION OF LENTIL PROTEIN WITH OTHER PROTEINS TO IMPROVE THE FUNCTIONAL AND NUTRITIONAL PROPERTIES

ALROSAN MOHAMMAD HANI AHMAD

UNIVERSITI SAINS MALAYSIA

2022

STRUCTURAL INTERACTIONS AND FERMENTATION OF LENTIL PROTEIN WITH OTHER PROTEINS TO IMPROVE THE FUNCTIONAL AND NUTRITIONAL PROPERTIES

by

ALROSAN MOHAMMAD HANI AHMAD

Thesis submitted in fulfilment of the requirements for the degree of Doctor of Philosophy

February 2022

ACKNOWLEDGEMENT

Thanks to Allah, who gave me the strength to face all challenges to attain my achievements

My deep gratitude to my advisor **Dr. Tan Thuan Chew**, I was honoured to be his student in USM a long academic life full of success and achievements. I was blessed to learn from his wisdom and patience as well as from his enormous knowledge.

I am very grateful to my dear co-advisor **Prof. Dato. Dr. Azhar Mat Easa**, who had a great impact on accomplishing this work and attaining its objectives through supporting me in organising this research in a dynamic and effective design and with the necessary technical skills.

I would like to express my sincere gratitude to **Prof. Dr. Muhammad H. Alu'datt** and **Dr. Sana Gammoh** for the continuous support of my Ph.D study and research.

My sincere acknowledgement to my father and mother, who have been a source of support and encouragement to me throughout my life, their prayer for me was what sustained me thus far, my mother **Alia T. Hatamleh**, who has been an inspiration in my life, and also for the myriad of ways in which, throughout my life, has activity supported me in my determination to find and realize my potential, and to make this contribution to our world.

To my brothers, Yazan, Hazem, Abdullah and Ahmad.

To my sisters, Shereen, Isra'a and Asma'a, whose, constant support and encouragements never cease to sustain me.

To my all friends, Ahmad H. Salal, Hussain Nemreny, Mahmood Subhi Jameel and Mohammad Ali

My special thanks to my wife, Shahid alkhateeb for support and encouragement to be throughout my study time

I also thank my friend in JUST, Ashraf Mutlaq

TABLE OF CONTENTS

ACK	NOWL	EDGEMI	ENT	ii
TABI	LE OF	CONTEN	TS	. iii
LIST	OF TA	BLES		X
LIST	OF FI	GURES		xiv
LIST	OF AE	BREVIA	TIONS	XX
ABST	RAK.			ciii
ABST	FRACT		X	XV
CHA	PTER 1	I INT	RODUCTION	1
1.1	Backg	round of S	tudy	1
1.2.	Ration	ale of Stud	ły	7
1.2.1	Aim a	nd Objecti	ves	7
1.2.2	Proble	m Stateme	nts	7
1.2.3	Hypot	heses		8
1.3	Frame	work of St	udy	9
CHA	PTER 2	2 LIT	ERATURE REVIEW	10
2.1	Plant-	Based Prot	eins	10
	2.1.1	Lentil Pro	oteins (LPs)	14
		2.1.1(a)	Chemistry and Structure of LPs	14
		2.1.1(b)	Quality and Health Benefits of LPs	15
		2.1.1(c)	Functional Properties of LPs	15
	2.1.2	Quinoa P	roteins (QPs)	22
		2.1.2(a)	Chemistry and Structure of QPs	22
		2.1.2(b)	Quality and Health Benefits of QPs	23
		2.1.2(c)	Functional Properties of QPs	23
2.2	Dairy-	Based Pro	teins	25
	2.2.1	Whey Pro	otein Isolates (WPIs)	25

		2.2.1(a)	Chemistry and Structure of WPIs	25
		2.2.1(b)	Quality and Health Benefits of WPIs	26
		2.2.1(c)	Functional Properties of WPIs	26
	2.2.2	Casein Pr	roteins (CPs)	28
		2.2.2(a)	Chemistry and Structure of CPs	28
		2.2.2(b)	Quality and Health Benefits of CPs	29
		2.2.2(c)	Functional Properties of CPs	29
2.3	Protei	n-Protein I	nteraction (PPI)	30
	2.3.2	Molecula	r Forces involved in Protein Structure	37
		2.3.2(a)	Molecular Force Binding based on Protein Interaction in Solutions	39
		2.3.2(b)	Molecular Forces Binding based on Interaction at the Interface	43
	2.3.3	Structura Result of	l Alterations of Protein Complex Formation as a Molecular Forces	45
	2.3.4	Interfacia Interactio	al Properties of Protein Complexes as a Result of an	47
	2.3.5	Inhibition Interaction	n of Structure Folding of Protein through Protein	49
	2.3.6	Character	risation of Protein Structure	49
		2.3.6(a)	X-ray Diffraction (XRD)	50
		2.3.6(b)	Fourier Transform Infrared Spectroscopy (FTIR) and Circular Dichroism (CD)	54
		2.3.6(c)	Microscopy	54
	2.3.7	Determin	nation the Molecular Forces Governing in PPI	56
	2.3.8	Effect of	PPI on Functional Properties of Protein	60
		2.3.8(a)	Gelation	60
		2.3.8(b)	Emulsification	62
		2.3.8(c)	Solubility	64
2.4	Ferme	entation		66

	2.4.1	Water Ke	fir Seeds (WKS)	70
	2.4.2	Effect of	Fermentation on Protein Quality	73
	2.4.3	Mechanis	sms of Degradation of Proteins	75
	2.4.4	Digestibil	lity	75
	2.4.5	Effect Fer	rmentation on Carbohydrates	76
	2.4.6	Effect Fer	rmentation on Non-nutritive Compounds	77
		2.4.6(a)	Phenolics Compounds	79
			2.4.6(a)(i) Phenolic Covalent Bonds	81
		2.4.6(b)	Saponins	82
			2.4.6(b)(i) Structural Aspects of Saponins	85
			2.4.6(b)(ii) Effect of Processing on Saponins in Pulses	86
CHA	APTER	3 PAR PRO ISOI	TICULAR CHARACTERISTICS OF LENTIL TEINS, QUINOA PROTEINS, WHEY PROTEIN LATE AND CASEIN PROTEINS	87
3.1	Introd	uction		87
3.2	Mater	ials and Me	ethods	88
	3.2.1	Materials	and Chemicals	88
	3.2.2	Preparatio	on of LPs	89
	3.2.3	Preparatio	on of QPs	90
	3.2.4	Determin	ation of Protein Solubility	90
	3.2.5	Determin	ation of Total Crude Protein	91
	3.2.6	Spectral A	Acquirements	92
		3.2.6(a)	UV Spectrophotometry	92
		3.2.6(b)	Fluorescence spectrometry	93
		3.2.6(c)	Hydrophobicity (<i>H</i> ₀)	93
		3.2.6(d)	Fourier Transform Infrared Spectroscopy (FTIR)	94
	3.2.7	Zeta Pote	ntial	95
	3.2.8	Surface N	Aorphology	96

	3.2.9	Amino Acid Analysis	96
	3.2.10	Statistical Analysis	97
3.3	Result	s and Discussion	98
	3.3.1	Solubility in Water	98
	3.3.2	Surface Morphology of LPs, QPs, WPIs and CPs	100
	3.3.3	Protein Structure	102
	3.3.4	Interpretation of Self-Repellent Surfaces	107
		3.3.4(a) Surface Hydrophobicity	107
		3.3.4(b) Zeta Potential	108
	3.3.5	Amino Acids	110
3.4	Conclu	usion	112
		PROTEINS ON THE SOLUBILITY, PROTEIN STRUCTURE, INTERFACIAL PROPERTIES AND QUALITY OF LENTIL PROTEIN BASED ON THE PH-RECYCLING	113
4.1	Introdu	action	113
4.1 4.2	Introdu Materi	als and Methods	113 116
4.1 4.2	Introdu Materi 4.2.1	als and Methods	113 116 116
4.1 4.2	Introdu Materi 4.2.1 4.2.2	als and Methods Materials and chemicals Complexation LPs with WPIs, QPs, and CPs	113 116 116 116
4.1 4.2	Introdu Materi 4.2.1 4.2.2 4.2.3	als and Methods Materials and chemicals Complexation LPs with WPIs, QPs, and CPs Determination of Protein Solubility in Water	113 116 116 116 119
4.1 4.2	Introdu Materi 4.2.1 4.2.2 4.2.3 4.2.4	als and Methods Materials and chemicals Complexation LPs with WPIs, QPs, and CPs Determination of Protein Solubility in Water Spectral Acquirements	113 116 116 116 119 119
4.1 4.2	Introdu Materi 4.2.1 4.2.2 4.2.3 4.2.4	als and Methods Materials and chemicals Complexation LPs with WPIs, QPs, and CPs Determination of Protein Solubility in Water Spectral Acquirements	113 116 116 116 119 119 120
4.1 4.2	Introdu Materi 4.2.1 4.2.2 4.2.3 4.2.4	als and Methods Materials and chemicals Complexation LPs with WPIs, QPs, and CPs Determination of Protein Solubility in Water Spectral Acquirements 4.2.4(a) UV Spectrophotometry 4.2.4(b) Fluorescence Spectrometry	113 116 116 116 119 119 120 120
4.1 4.2	Introdu Materi 4.2.1 4.2.2 4.2.3 4.2.4	als and Methods Materials and chemicals Complexation LPs with WPIs, QPs, and CPs Determination of Protein Solubility in Water Spectral Acquirements 4.2.4(a) UV Spectrophotometry 4.2.4(b) Fluorescence Spectrometry 4.2.4(c) Hydrophobicity (<i>H</i> ₀)	113 116 116 116 119 119 120 121
4.1 4.2	Introdu Materi 4.2.1 4.2.2 4.2.3 4.2.4	actionals and MethodsMaterials and chemicalsComplexation LPs with WPIs, QPs, and CPsDetermination of Protein Solubility in WaterSpectral Acquirements4.2.4(a)UV Spectrophotometry4.2.4(b)Fluorescence Spectrometry4.2.4(c)Hydrophobicity (H_0)4.2.4(d)Molecular Forces	113 116 116 116 119 119 120 120 121
4.1 4.2	Introdu Materi 4.2.1 4.2.2 4.2.3 4.2.4	actionals and MethodsMaterials and chemicalsComplexation LPs with WPIs, QPs, and CPsDetermination of Protein Solubility in WaterSpectral Acquirements4.2.4(a)UV Spectrophotometry4.2.4(b)Fluorescence Spectrometry4.2.4(c)Hydrophobicity (H_0)4.2.4(d)Molecular Forces4.2.4(e)Fourier Transform Infrared Spectroscopy (FTIR)	113 116 116 116 119 119 120 121 121 122
4.1 4.2	Introdu Materi 4.2.1 4.2.2 4.2.3 4.2.4	action als and Methods Materials and chemicals Complexation LPs with WPIs, QPs, and CPs Determination of Protein Solubility in Water Spectral Acquirements 4.2.4(a) UV Spectrophotometry 4.2.4(b) Fluorescence Spectrometry 4.2.4(c) Hydrophobicity (H_0) 4.2.4(d) Molecular Forces 4.2.4(e) Fourier Transform Infrared Spectroscopy (FTIR) Zeta Potential	113 116 116 116 119 119 120 120 121 121 122 123

	4.2.7	Amino Acid Analysis	123
	4.2.8	Statistical Analysis	124
4.3	Result	s and Discussion	124
	4.3.1	Solubility in Water	124
	4.3.2	Surface Morphology Alteration Induced by Protein-Protein Interactions	130
	4.3.3	Protein-Protein Interaction	135
	4.3.4	Molecular Forces Governing Protein-Protein Interaction	137
	4.3.5	Structural Changes Attributable to pH Recycling	139
	4.3.6	Interfacial Characteristics of Interaction-Affected Proteins	150
	4.3.7	Analysis of Amino acids	154
4.4	Conclu	ision	159
СНА	PTER 5	5 EFFECTS OF FERMENTATION ON THE QUALITY, STRUCTURE, AND NON-NUTRIENT CONTENTS OF LENTH OUNOA CASEIN AND WHEY	
		PROTEINS	160
5.1	Introdu	PROTEINS	 160 160
5.1 5.2	Introdu Materi	als and Methods	 160 160 162
5.1 5.2	Introdu Materi 5.2.1	Proparation of WKS	 160 160 162 162
5.1 5.2	Introdu Materi 5.2.1 5.2.2	PROTEINS	 160 160 162 162 163
5.1 5.2	Introdu Materi 5.2.1 5.2.2 5.2.3	PROTEINS	 160 160 162 162 163 164
5.1 5.2	Introdu Materi 5.2.1 5.2.2 5.2.3 5.2.4	PROTEINS	160 160 162 162 163 164 164
5.1 5.2	Introdu Materi 5.2.1 5.2.2 5.2.3 5.2.4 5.2.5	PROTEINS	160 160 162 163 163 164 164 165
5.1 5.2	Introdu Materi 5.2.1 5.2.2 5.2.3 5.2.3 5.2.4 5.2.5 5.2.6	PROTEINS	160 160 162 163 163 164 165 165
5.1 5.2	Introdu Materi 5.2.1 5.2.2 5.2.3 5.2.4 5.2.5 5.2.6 5.2.7	PROTEINS	160 162 162 163 164 164 165 165 166
5.1 5.2	Introdu Materi 5.2.1 5.2.2 5.2.3 5.2.3 5.2.4 5.2.5 5.2.6 5.2.7 5.2.8	PROTEINS als and Methods Preparation of WKS Design of LP-WKS Preparation and Fermentation Assessment of pH and Total Soluble Solids (TSS) Protein Digestibility Determination of Total Phenolic Content (TPC) Determination of Sugars Determination of Phenolic Compounds Fourier Transform Infrared Spectroscopy (FTIR)	160 162 162 163 163 164 165 165 166 168
5.1 5.2	Introdu Materi 5.2.1 5.2.2 5.2.3 5.2.4 5.2.5 5.2.6 5.2.7 5.2.8 5.2.9	PROTEINS als and Methods Preparation of WKS Design of LP-WKS Preparation and Fermentation Assessment of pH and Total Soluble Solids (TSS) Protein Digestibility Determination of Total Phenolic Content (TPC) Determination of Sugars Determination of Phenolic Compounds Fourier Transform Infrared Spectroscopy (FTIR) Total Saponins	160 162 162 163 163 164 165 165 166 168 168
5.1 5.2	Introdu Materi 5.2.1 5.2.2 5.2.3 5.2.4 5.2.5 5.2.6 5.2.7 5.2.8 5.2.9 5.2.10	PROTEINS	160 162 162 163 163 164 165 165 166 168 168 168

		5.2.10(b) Preparation of Culture Media	169
		5.2.10(c) Incubation Condition	169
		5.2.10(d) Enumeration of Microorganisms by Surface Plate Technology	170
	5.2.11	Statistical Analysis	171
5.3	Result	s and Discussion	171
	5.3.1	Effects of WKS Fermentation on pH and TSS of Proteins	171
	5.3.2	Effect of WKS Fermentation on the Quality of Proteins	174
	5.3.3	Effect of WKS Fermentation on the Sugars Profile of Proteins	176
	5.3.4	Influence of WKS Fermentation on the Secondary Protein Structure of Proteins	179
	5.3.5	Effect of WKS Fermentation on the Phenolic Compounds of Proteins	186
	5.3.6	The Dynamics of Bacterial Growth throughout Protein Fermentation	195
	5.3.7	Effect Fermentation on the Saponin Content	198
_ .			
5.4	Concl	usion	201
5.4 CHA	Concl	usion 6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS OF LENTIL-BASED PROTEIN COMPLEXES CONTENT DURING FERMENTATION	201 203
5.4 CHA 6.1	Concl APTER	usion	201 203 203
 5.4 CHA 6.1 6.2 	Concl APTER Introd Mater	 analysis of QUALITY, STRUCTURE, AND NON- NUTRIENTS OF LENTIL-BASED PROTEIN COMPLEXES CONTENT DURING FERMENTATION uction and Methods 	201 203 203 205
5.4CHA6.16.2	Concl APTER Introd Mater 6.2.1	 analysis of quality, structure, and non- nutrients of lentil-based protein complexes content during FERMENTATION uction preparation of WKS 	201 203 203 205 206
5.4CHA6.16.2	Concl APTER Introd Mater 6.2.1 6.2.2	6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 0 LENTIL-BASED PROTEIN COMPLEXES COMPLEXES CONTENT During FERMENTATION uction Image: State Sta	201 203 203 205 206 206
5.4CHA6.16.2	Concl APTER Introd Mater 6.2.1 6.2.2 6.2.3	usion 6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 6 Interview COMPLEXES CONTENT COMPLEXES CONTENT During FERMENTATION uction ials and Methods Preparation of WKS Design of LP-based Protein Complexes WKS Fermentation Assessment of pH and Total Soluble Solids (TSS)	201 203 203 205 206 206 206
5.4CHA6.16.2	Concl APTER (Introd Mater 6.2.1 6.2.2 6.2.3 6.2.4	6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 0 LENTIL-BASED PROTEIN COMPLEXES COMPLEXES CONTENT During FERMENTATION uction Image: State Sta	201 203 203 205 206 206 206 207
5.4CH46.16.2	Concl APTER (Introd Mater 6.2.1 6.2.2 6.2.3 6.2.4 6.2.5	6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 7 DESIGN OF LENTIL-BASED 9 PROTEIN COMPLEXES CONTENT 9 DURING FERMENTATION FERMENTATION 9 Indext (Content (Cont	201 203 203 205 206 206 206 207 207
5.4CH46.16.2	Concl APTER (Introd Mater 6.2.1 6.2.2 6.2.3 6.2.4 6.2.5 6.2.6	6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 6 ANALYSIS OF QUALITY, STRUCTURE, AND NON- NUTRIENTS 0 PROTEIN COMPLEXES COMPLEXES CONTENT DURING FERMENTATION uction ials and Methods Preparation of WKS Design of LP-based Protein Complexes WKS Fermentation Assessment of pH and Total Soluble Solids (TSS) Protein Digestibility Determination of Total Phenolic Content (TPC) Determination of Sugars	201 203 203 205 206 206 206 207 207 207

	6.2.8	Fourier Transform Infrared Spectroscopy (FTIR)	208
	6.2.9	Total Saponins Content (TSC)	208
	6.2.10	Microbiological Quality	209
	6.2.11	Statistical Analysis	209
6.3	Result	s and Discussion	209
	6.3.1	Effects of WKS Fermentation on pH and TSS of LP-Based Protein Complexes	209
	6.3.2	Effect of WKS Fermentation on the Protein Quality of LP- Based Protein Complexes	212
	6.3.3	Effect of WKS Fermentation on the Sugars Profile of LP- Based Protein Complexes	215
	6.3.4	Influence of WKS Fermentation on the Secondary Protein Structure of LP-based Protein Complexes	219
	6.3.5	Effect of WKS Fermentation on the Phenolic Compounds of LP-based Protein Complexes	226
	6.3.6	The Dynamics of Bacterial Growth throughout Protein Fermentation	232
	6.3.7	Effect Fermentation on the Total Saponins Content	235
6.4	Conclu	usion	239
CHA	PTER 7	7 GENERAL CONCLUSION	240
7.1	Genera	al conclusion	240
7.2	Recon	nmendations	241
REFI	ERENC	'ES	242

LIST OF PUBLICATIONS

LIST OF TABLES

		Page
Table 2.1	Summary of food proteins sources commonly used in the food industry	11
Table 2.2	Summary outcomes of functional properties of lentil proteins (LPs), quinoa proteins (QPs), whey protein isolates (WPIs), and casein proteins (CPs)	17
Table 2.3	Summary of interaction types between proteins and their driving molecular forces	32
Table 2.4	Methods to identify and characterize the protein structure	50
Table 2.5	Methods to identify molecular forces of protein	56
Table 2.6	Summary of results of studies on the influence of natural/microorganism fermentation on fermented foods	68
Table 2.7	Summary of results from research on the fermentation of water kefir seed on fermented foods	71
Table 2.8	Summary of saponins and phenolic content in the lentil proteins	83
Table 3.1	LC-MS/MS gradient profile for determination of amino acid.	97
Table 3.2	Percentage of β -sheet, random coil (RC), α -helix, and β - turn (% of secondary protein structure) as results by FTIR between 1,600 to 1,700 1/cm of lentil proteins (LPs), whey protein isolates (WPIs), casein proteins (CPs), and quinoa proteins (QPs).	106
Table 3.3	Amino acids composition (%) in lentil proteins (LPs), whey protein isolates (WPIs), casein proteins (CPs) and quinoa proteins (QPs).	111
Table 4.1	Preparation of protein complex dispersions 1% (w/v) of lentil protein (LPs), whey protein isolates (WPIs), quinoa protein (QPs) and casein protein (CPs).	117
Table 4.2	Percentage of β -sheet, random coil (RC) α -helix, and β -turn (% of secondary protein structure) as results by FTIR between 1,600 to 1,700 1/cm of lentil proteins (LPs), whey protein isolates (WPIs), and LP-WPI protein complexes	142
Table 4.3	Percentage of β -sheet, random coil (RC), α -helix, and β -turn (% of secondary protein structure) as results by FT-IR	

	between 1,600 to 1,700 1/cm of lentil proteins (LPs), casein proteins (CPs), and LP-CP protein complexes.	143
Table 4.4	Percentage of β -sheet, random coil (RC), α -helix, and β -turn (% of secondary protein structure) based on FTIR spectra between 1,600 and 1,700 1/cm of lentil proteins (LPs), quinoa proteins (QPs), and LP-QP protein complexes.	144
Table 4.5	Analysis of amino acids (%) of lentil proteins (LPs), whey protein isolates (WPIs), and LP-WPI protein complexes	156
Table 4.6	Analysis of amino acids (%) of lentil proteins (LPs), quinoa proteins (QPs), and LP-QP protein complexes	157
Table 4.7	Analysis of amino acids (%) of lentil proteins (LPs), casein proteins (CPs), and LP-CP protein complexes.	158
Table 5.1	Retention time (t _R) for fructose, glucose, and sucrose	166
Table 5.2	HPLC gradient profile for determination of phenolic compounds.	167
Table 5.3	Retention time (t _R) and wavelength for phenolic compounds.	167
Table 5.4	Changes in the pH, total soluble solids (TSS) and protein digestibility of unfermented lentil proteins (LPs), quinoa proteins (QPs), whey protein isolate (WPIs) and casein protein (CPs) (Day 0) and water kefir seed (WKS) fermented LPs, QPs, WPIs, and CPs (Day 1 to 5)	173
Table 5.5	Changes in the sugar profile of unfermented lentil proteins (LPs), quinoa proteins (QPs), whey protein isolate (WPIs) and casein protein (CPs) (Day 0) and water kefir seed (WKS) fermented LPs, QPs, WPIs, and CPs (Day 1 to 5)	178
Table 5.6	Changes in the percentage of secondary protein components (β -sheet, random coil (RC), α -helix, and β -turn) of unfermented (Day 0) and water kefir seed fermented (from Day 1 to 5) lentil proteins (LPs) based on FTIR measurements.	181
Table 5.7	Changes in the percentage of secondary protein components (β -sheet, random coil (RC), α -helix, and β -turn) of unfermented (Day 0) and water kefir seed fermented (from Day 1 to 5) quinoa proteins (QPs) based on FTIR measurements.	182
Table 5.8	Changes in the percentage of secondary protein components (β -sheet, random coil (RC), α -helix, and β -turn) of unfermented (Day 0) and water kefir seed	

	fermented (from Day 1 to 5) whey protein isolates (WPIs) based on FTIR measurements
Table 5.9	Changes in the percentage of secondary protein components (β -sheet, random coil (RC), α -helix, and β - turn) of unfermented (Day 0) and water kefir seed fermented (from Day 1 to 5) casein proteins (CPs) based on FTIR measurements
Table 5.10	Changes in the total phenolic content (TPC, expressed in mg GAE/100 g) and phenolic compounds (mg/100 g) of unfermented (Day 0) and water kefir seed fermented (from Day 1 to 5) lentil proteins (LPs)
Table 5.11	Changes in the total phenolic content (TPC, expressed in mg GAE/100 g) and phenolic compounds (mg/100 g) of unfermented (Day 0) and water kefir seed fermented (from Day 1 to 5) quinoa proteins (QPs)
Table 5.12	Changes in the total phenolic content (TPC, expressed in mg GAE/100 g) and phenolic compounds (mg/100 g) of unfermented (Day 0) and water kefir seed fermented (from Day 1 to 5) whey protein isolates (WPIs)
Table 5.13	Changes in the total phenolic content (TPC, expressed in mg GAE/100 g) and phenolic compounds (mg/100 g) of unfermented (Day 0) and water kefir seed fermented (from Day 1 to 5) casein proteins (CPs)
Table 5.14	Changes in the total saponins content (TSC) (mg OAE/100 g) of unfermented (Day 0) lentil proteins (LPs), quinoa proteins (QPs), whey protein isolates (WPIs) and casein protein (CPs) and water kefir seed fermented (from Day 1 to 5) LPs, QPs, WPIs, and CPs
Table 6.1	Changes in the pH and total soluble solids (TSS), profile of unfermented LP-QP, LP-WPI, and LP-CP protein complexes (Day 0) and water kefir seed fermented LP-QP, LP-WPI, and LP-CP protein complexes (Day 1 to 5)
Table 6.2	Changes in protein digestibility of unfermented LP-QP, LP-WPI, and LP-CP protein complexes (Day 0) and water kefir seed fermented LP-QP, LP-WPI, and LP-CP protein complexes (Day 1 to 5)
Table 6.3	Sugars profile of unfermented LP-QP, LP-WPI, and LP-CP protein complexes (Day 0) and water kefir seed fermented LP-QP, LP-WPI, and LP-CP protein complexes (Day 1 to 5)
Table 6.4	Changes in the percentage of secondary protein components (β -sheet, random coil (RC), α -helix, and β -

	turn) of unfermented LP-QP protein complexes (Day 0) and fermented LP-QP protein complexes (Day 1 to 5) based on FTIR measurements.	221
Table 6.5	Changes in the percentage of secondary protein components (β -sheet, random coil (RC), α -helix, and β -turn) of unfermented LP-WP complexes (Day 0) and fermented LP-WP complexes (Day 1 to 5) based on FTIR measurements.	222
Table 6.6	Changes in the percentage of secondary protein components (β -sheet, random coil (RC), α -helix, and β -turn) of unfermented LP-CP protein complexes (Day 0) and fermented LP-CP protein complexes (Day 1 to 5) based on FTIR measurements.	223
Table 6.7	Changes in the total phenolic content (TPC, mg GAE/100 g) and phenolic compounds (mg/100 g) of unfermented LP-QP protein complexes (Day 0) and water kefir seed fermented LP-QP protein complexes (Day 1 to 5).	227
Table 6.8	Changes in the total phenolic content (TPC, mg GAE/100 g) and phenolic compounds (mg/100 g) of unfermented LP-WP protein complexes (Day 0) and water kefir seed fermented LP-WP protein complexes (Day 1 to 5).	228
Table 6.9	Changes in the total phenolic content (TPC, mg GAE/100 g) and phenolic compounds (mg/100 g) of unfermented LP-CP protein complexes (Day 0) and water kefir seed fermented LP-CP protein complexes (Day 1 to 5).	229
Table 6.10	Changes in the total saponins content (TSC) (mg OAE/100 g) of LP-QP, LP-WP, and LP-CP protein complexes (Day 0) and water kefir seed fermented LP-QP, LP-WPI, and LP-CP protein complexes (Day 1 to 5)	238

LIST OF FIGURES

		Page
Figure 1.1.	Schematic diagram of the framework of the study: Phase I, Phase II, and Phase III	9
Figure 2.1	Types of molecular forces involve protein structure and protein-protein interaction.	38
Figure 2.2	Schematic illustration of the interactions of hetero-proteins. When macroions of lactoferrins and soy protein isolates are close enough by electrostatic interaction, water molecules will be expelled.	43
Figure 2.3	Schematic illustration of the soy protein isolates (SPIs) and sodium caseinate (SC) as stabilized droplets with the emulsion at pH 6.8 laden with vitamin A (VA)	45
Figure 2.4	Schematic illustration of the protein-protein interaction mechanism between rice proteins and casein proteins to form protein hybrids.	46
Figure 2.5	Schematic illustration of the gelation mechanism of a binary mixture of globular protein formed as a result of protein-protein interaction. Gelation could be derived from crosslinking of (A) finely stranded networks or (B) microgels.	61
Figure 2.6	Schematic illustration of the food matrix of different components including protein, carbohydrate and non-nutritive compounds, and enzymes breakdown the complex matrix.	74
Figure 2.7	A schematic diagram of the mechanisms of the fermentation process of the protein matrix. (1) Protein matrix. (2) The metabolism active by microorganisms and their enzymes. (3) During fermentation, the metabolism is active by microorganisms and their enzymes, including oligopeptidase, amylases, lipases, esterases, peroxidases, peptide hydrolases, and iminopeptidase. (4) Degradation of the matrix protein and hydrolysis of the bonds between different molecules. (5) Degradation of the saponins content, decreasing the total saponins content might be attributed to the degradation of saponins connected with protein and carbohydrates through microbial fermentation. (6) Degradation of protein hydrolysis by peptidase hydrolysis to make partial-digestion protein and increase the protein quality of QPs. (7) Total phenolic compounds were increased in the first days of fermentation and then decreased during the last days because of the hydrolysis of	

	the phenolic compounds by Lactobacillus strains. (8) The microbial community may use sucrose, glucose, and fructose as carbon source.	78
Figure 2.8	Schematic illustration of the primary cell wall structure of plant with structural components and phenolic compounds and description insert table the covalent bounds of phenolic compounds to cell wall structural components	80
Figure 3.1	The solubility of lentil proteins (LPs), whey protein isolates (WPIs), casein proteins (CPs) and quinoa proteins (QPs) at pH 7.0.	98
Figure 3.2	The surface morphologies of (A) lentil proteins (LPs), (B) whey protein isolates (WPIs), (C) quinoa proteins (QPs), and (D) casein proteins (CPs).	102
Figure 3.3	The fluorescence intensity of lentil proteins (LPs), whey protein isolates (WPIs), casein proteins (CPs), and quinoa proteins (QPs) at pH 7.0.	104
Figure 3.4	The absorbance of lentil proteins (LPs), whey protein isolates (WPIs), casein proteins (CPs) and quinoa proteins (QPs) at pH 7.0.	105
Figure 3.5	The surface hydrophobicity of lentil proteins (LPs), whey protein isolates (WPIs), casein proteins (CPs), and quinoa proteins (QPs) at pH 7.0.	108
Figure 3.6	The surface charge of lentil proteins (LPs), whey protein isolates (WPIs), casein proteins (CPs), and quinoa proteins (QPs) at pH 7.0.	109
Figure 4.1	Schematic diagram illustrated steps of prepare protein complexes powder Preparation of protein complex dispersions (1%, w/v) of lentil proteins (LPs) with other proteins; whey protein isolates (WPIs), quinoa proteins (QPs), and casein protein (CPs), based on the pH-recycling.	118
Figure 4.2	The solubility of control (LPs subjected to pH-recycling but without the addition of WPIs), whey protein isolates (WPIs), and LP-WPI protein complexes at a different ratio of 1:1.2, 1:1.5, and 1:2. Values with different superscripts within the columns are statistically significant from each other (P <0.05)	125
Figure 4.3	The solubility of control (LPs subjected to pH-recycling but without the addition of CPs), casein proteins (CPs), and LP-CP protein complexes at a different ratio of 1:0.1, 1:0.5, and 1:1. Values with different superscripts within the column are statistically significant from each other (P <0.05)	127

Figure 4.4	The solubility of control (LPs subjected to pH-recycling but without the addition of QPs), quinoa proteins, and LP-QP complexes at different ratios of 1:1,1:1.2, 1:1.5, and 1:2. Values with different superscripts within the column are statistically significant from each other (P <0.05)
Figure 4.5	Surface morphology of (A) lentil proteins (LPs), (B) whey protein isolates (WPIs), and LP-WPI protein complexes; (C) LP-WPI-1:1.2, (D) LP-WPI-1:1.5, and (E) LP-WPI- 1:2. LP-WPI-1:1.2, LP-WPI-1:1.5, and LP-WPI-1:2 represent LP-WPI protein complexes at ratios of 1:1.2, 1:1.5, and 1:2. 131
Figure 4.6	Surface morphology of (A) lentil proteins (LPs), (B) quinoa protein (QPs), and LP-QP protein complexes; (C) LP-QP- 1:1, (D) LP-QP-1:1.5, and (E) LP-WPI-1:2. LP-QP-1:1, LP-QP-1:1.5, and LP-QP-1:2 represent LP-QP protein complexes at ratios of 1:1, 1:1.5, and 1:2
Figure 4.7	Surface morphology of (A) lentil proteins (LPs), (B) casein protein (CPs), and LP-CP complexes; (C) LP-CP-1:0.1, (D) LP-CP-1:0.5, and (E) LP-CP-1:1. LP-CP-1:0.1, LP-CP- 1:0.5, and LP-CP-1:1 represent LP-CP protein complexes at ratio of 1:0.1, 1:0.5, and 1:1
Figure 4.8	(A) Fluorescence emission of lentil proteins (LPs) and different ratios of LP-WPI-1:1.2, LP-WPI-1:1.5, and LP-WPI-1:2 represent LP-WPI protein complexes at ratios of 1:1.2, 1:1.5, and 1:2, respectively. The inset chart is the zoom in profile of the LP-WP complexes at different ratio at pH 7:0. (B) Fluorescence emission of LPs and different ratios of LP-QP-1:1, LP-QP-1:1.5, and LP-QP-1:2 represent LP-QP protein complexes at ratios of 1:1, 1:1.5, and 1:2, respectively. The inset chart is the zoom in profile of the LP-QP-1:1.5, and LP-QP-1:2 represent LP-QP protein complexes at ratios of 1:1, 1:1.5, and 1:2, respectively. The inset chart is the zoom in profile of the LP-QP complexes at different ratios at pH 7:0. (C) Fluorescence emission of LPs and different ratios of LP-CP-1:0.1, LP-CP-1:0.5, and LP-CP-1:1 represent LP-CP complexes at ratio of 1:0.1, 1:0.5, and 1:1, respectively
Figure 4.9	(A) Fluorescence emission profile of (A) LP-WP-1:1.2, (B) LP-CP-1:0.1, and (C) LP-QP-1:1 with foreign substances: NaCl, thiourea, and sodium dodecyl sulphate represent electrostatic interactions, hydrogen bonding, and hydrophobic interactions, respectively. LP-WP-1:1.2 represents LP-WP protein complexes at a ratio of 1:1.2. LP-QP-1:1 represents LP-QP protein complexes at a ratio of 1:1.2. LP-QP-1:1.0.1 represents LP-CP protein complexes at a ratio of 1:0.1
Figure 4.10	Fluorescence emission profile of (A) LP-WP-1:1.2, (B) LP-QP-1:1, and (C) LP-CP-1:0.1 at different pH (pH 7 to 12).

	LP-WP-1:1.2 represents LP-WPI protein complex at ratio of 1:1.2. LP-QP-1:1 represents LP-QP protein complex at ratio of 1:1. LP-CP-1:0.1 represents LP-CP protein complex at ratio of 1:0.1.	. 140
Figure 4.11	Ultraviolet-visible spectrum profile of (A) LP-WPI-1:1.2, (B) LP-QP-1:1, and (C) LP-CP-1:0.1 at a range of pH 7 to 12. LP-WP-1:1.2 represents LP-WP protein complex at ratio of 1:1.2. LP-QP 1:1 represents LP-QP protein complex at ratio of 1:1. LP-CP-1:0.1 represents LP-CP protein complex at ratio of 1:0.1.	. 146
Figure 4.12	(A) Ultraviolet-visible spectrum profile of different ratio of (A) LP-WPI, (B) LP-CP, and (C) LP-CP protein complexes at pH 7. LP-WPI-1:1.2, LP-WPI-1:1.5, and LP-WPI-1:2 represent LP-WPI protein complexes at ratios of 1:1.2, 1:1.5, and 1:2, respectively. LP-QP-1:1, LP-QP-1:1.5, and LP-QP-1:2 represent LP-QP protein complexes at ratios of 1:1, 1:1.5, and 1:2, respectively. LP-CP-1:0.1, LP-CP- 1:0.5, and LP-CP-1:1 represent LP-CP protein complexes at ratio of 1:0.1, 1:0.5, and 1:1, respectively.	. 147
Figure 4.13	Schematic diagram of the interactions between lentil proteins (LPs) and whey protein isolates (WPIs). (1) Protein dispersion (1%) contains LPs 45% and WPIs 55% in distilled water. (2) The protein dispersion was adjusted to 12 and stirred for 1 h at room temperature (21°C). (3) Removal of protons on proteins during complexation at pH 12 resulted in proteins interact electrostatically. (4) The protein dispersion was readjusted to 7.0 during the neutralization process, leading to the contribution of ionic interaction, hydrogen bonds, and hydrophobic interaction in the creation of LP-WPI protein complexes. (5) LP-WPI protein complexes formed by structural interaction driving molecular forces.	. 148
Figure 4.14	Schematic diagram of interactions between lentil proteins (LPs) and Casein proteins (CPs). (1) protein dispersion (1%) contains LPs 90% and CP 10% in distilled water. (2) The protein dispersion was adjusted to 12 and stirred for 1 h at room temperature (21 °C). (3) Completely remove proton on proteins during complexation at pH 12, resulting proteins interacts electrostatically. (4) The protein dispersion was readjusted to 7.0, during the neutralization process, ionic interaction, hydrogen bonds and hydrophobic interaction contribute in creation of protein complexes of LPs and CPs. (5) LP-CP protein complexes formed by structural interaction driving molecular forces.	. 149

Figure 4.15	Schematic diagram of interactions between lentil proteins (LPs) and quinoa proteins (QPs) driven by ionic interaction, hydrogen bonds, and hydrophobic interaction	150
Figure 4.16	Surface hydrophobicity of (A) LP-WPI-1:1.2, (B) LP-QP- 1:1, and (C) LP-CP-1:0.1 at a range of pH 7 to 12. Zeta potential of (D) LP-WPI-1:1.2, (E) LP-QP-1:1, and (F) LP- CP-1:0.1 at a range of pH 7 to 12. The data presented as the mean \pm standard deviation (n=3). Values with different superscripts within the same figure are statistically significant from each other (<i>P</i> <0.05). LP-WPI-1:1.2 represents the LP-WPI protein complex at the ratio of 1:1.2. LP-QP-1:1.1 represents the LP-QP protein complex at the ratio of 1:1.1. LP-CP-1:0.1 represents the LP-CP protein complex at the ratio of 1:0.1.	152
Figure 4.17	(A) Surface hydrophobicity of lentil proteins (LPs), whey protein isolates (WPIs) and different ratio of LP-WP protein complexes at pH 7. (B) Surface hydrophobicity of LPs, quinoa proteins (QPs) and different ratios of LP-QP protein complexes at pH 7. (C) Surface hydrophobicity of LPs, casein proteins (CPs) and different ratio of LP-CP protein complexes at pH 7. (D) Zeta potential of LPs, WPI, and LP-WP protein complexes at different ratios. (E) Zeta potential of LPs, QPs, and LP-QP protein complexes at different ratios. (F) Zeta potential of LPs, CPs, and LP-CP protein complexes at different ratios. The data presented as the mean \pm standard deviation (n=3). Values with different superscripts within the figure are statistically significant from each other (<i>P</i> <0.05)	153
Figure 5.1	Schematic diagram of the fermentation process steps. (1) Preparation of water kefir seeds (WKS). (2) Preparation of protein dispersion (1%, w/v) of lentil protein (LPs), quinoa protein (QPs), whey protein isolates (WPIs) or casein proteins (CPs) and 5% (v/v) WKS in distilled water at 25 °C for 5 days of fermentation.	163
Figure 5.2	Schematic diagram of the fermentation process steps. Protein with phenolic compounds and carbohydrates connected via ester linkages and breakdown of the ester linkages by microorganisms and their enzymes, resulting in the release of protein, phenolic compounds, and carbohydrates.	193
Figure 5.3	Microbial activity of lactic acid bacteria (LAB), acetic acid bacteria (AAB), and yeast throughout the lentil proteins (LPs) fermentation with water kefir seeds (WKS). The data presented in the chart represent the mean \pm standard deviation (n=2). Values with different superscripts within	

	the same colour column are statistically significant from each other (P <0.05)
Figure 5.4	Microbial activity of lactic acid bacteria (LAB), acetic acid bacteria (AAB), and yeast throughout the quinoa proteins (QPs) fermentation with water kefir seeds (WKS). Values with different superscripts within the same colour column are statistically significant from each other (P <0.05)
Figure 5.5	Microbial activity of lactic acid bacteria (LAB), acetic acid bacteria (AAB), and yeast throughout the whey protein isolates (WPIs) fermentation with water kefir seeds. Values with different superscripts within the same colour column are statistically significant from each other (P <0.05)
Figure 5.6	Microbial activity of lactic acid bacteria (LAB), acetic acid bacteria (AAB), and yeast throughout the casein proteins (CPs) fermentation with water kefir seeds. Values with different superscripts within the same colour column are statistically significant from each other (P <0.05)
Figure 6.1	Microbial activity of lactic acid bacteria (LAB), acetic acid bacteria (AAB), and yeast throughout LP-QP protein complexes fermentation with water kefir seeds. Values with different superscripts within the same colour column are statistically significant from each other ($P < 0.05$). LP- QP represents LP-QP protein complex at a ratio of 1:1
Figure 6.2	Microbial activity of lactic acid bacteria (LAB), acetic acid bacteria (AAB), and yeast throughout LP-WP protein complexes fermentation with water kefir seeds. Values with different superscripts within the same colour column are statistically significant from each other (P <0.05). LP- WP represents LP-WP protein complex at a ratio of 1:1.2234
Figure 6.3	Microbial activity of lactic acid bacteria (LAB), acetic acid bacteria (AAB), and yeast throughout LP-CP protein complexes fermentation with water kefir seeds. Values with different superscripts within the same colour column are statistically significant from each other (P <0.05). LP- CP represents LP-CP protein complex at a ratio of 1:0.1

LIST OF ABBREVIATIONS

A230 Wavelength 230 nm AAB Acetic Acid Bacteria AFM Atomic Force Microscopy ANS 8-anilino-1-naphthalenesulfonic Acid **BPPs Blood Plasma Proteins** CD Circular Dichroism **CFPs** Cod Fish Protein CFU colony-forming units **CPHs** Corn Protein Hydrolysates CPs **Casein Proteins** EAI **Emulsion Activity Index** ESI **Electrospray Ionization** ESI **Emulsion Stability Index** ESI+ Electrospray Interface FAO Food and Agriculture Organisation FTIR Fourier-Transform Infrared Spectroscopy HPLC High-performance Liquid Chromatography ITC Isothermal Titration Calorimetry LAB Lactic Acid Bacteria LFs Lactoferrins LPs Lentil Proteins MRM Multiple Reaction-Monitoring ND Not Detection OAC Oil Absorption Capacity

nľ	Isoelectric Point
pr	Isociecule i onit
PPI	Protein-Protein Interaction
PPs	Peanut Proteins
QPs	Quinoa Proteins
RGs	Rice Glutelin
RPs	Rice Proteins
SC	Sodium Caseinate
SDS	Sodium Dodecyl Sulphate
SEM	Scanning Electron Microscopy
SF	Soy Flour
SGs	Soy Globulins
SL	Soybean Lecithin
SLS	Static Light Scattering
SPIs	Soy Protein Isolates
SPIs	Soy Protein Isolates
SRM	Super-Resolution Microscopy
TEM	Transmission Electron Microscopy
TPC	Total Phenolic Compounds
TPC	Total Phenolic Content
t _R	Retention Time
TSC	Total Saponins Content
TSS	Total Soluble Solids
VA	Vitamin A
WAC	Water Absorption Capacity
WGPs	Wheat Gluten Proteins

- WHO World Health Organisation
- WKS Water Kefir Seeds
- WPIs Whey Protein Isolates
- XPS X-ray Photoelectron Spectroscopy
- XRD X-ray Diffraction
- β -lg β -lactoglobulin
- λF_{max} Maximum Emission

INTERAKSI STRUKTUR DAN FERMENTASI PROTEIN LENTIL DENGAN PROTEIN LAIN UNTUK PENAMBAHBAIKAN SIFAT FUNGSIAN DAN NUTRISI

ABSTRAK

Dalam beberapa tahun kebelakangan ini, ada peningkatan minat terhadap protein lentil (LP) disebabkan oleh ketersediaannya yang baik, kos rendah, dan pengeluaran global. Faktor utama yang membatasi penggunaan LP dalam industri makanan adalah kelarutannya yang rendah, kekurangan pencernaan, dan jumlah saponin yang tinggi, yang menyebabkan kepahitan. Interaksi protein-protein (PPI) dan fermentasi telah digunakan pada masa lalu untuk mengatasi beberapa batasan ini dalam protein lentil. Pengkompleksan protein dan prosedur fermentasi WKS berjaya digabungkan untuk meningkatkan aspek fungsian dan pemakanan kompleks protein berasaskan LP. Tujuan kajian ini adalah untuk meningkatkan kelarutan dan pencernaan protein lentil, serta penurunan kandungan saponinnya melalui PPI dan fermentasi. Untuk mencapai tujuan ini, kajian ini dipisahkan kepada 3 fasa. Objektif Fasa 1 adalah untuk membandingkan struktur protein dan kelarutan dan sifat antara 4 jenis protein yang berlainan: LP, protein quinoa (QP), protein kasein (CP), dan protein whey (WP). Kelarutan WP, QP, dan CP lebih tinggi daripada LP, masing-masing pada 92.7, 80.1, 89.7, dan 58.0% (P<0.05). Menurut penemuan kami, WPI, QP, dan CP mempunyai struktur protein dan sifat antara muka yang lebih unik daripada LP. Manakala hidrofobisiti permukaan LP lebih tinggi daripada WP, CP dan QP, sekitar 618, 24, 236, dan 404 a.u. (P<0.05), masing-masing. Sebaliknya, WPI, CP dan QP mempunyai cas permukaan yang lebih tinggi lebih berbanding LP, masing-masing kira-kira -33.2, -30.9, -32.9, dan -22.9 (P<0.05). Menurut penemuan kami, kadar

struktur tertiari LP adalah lebih tinggi daripada WPI, CP, dan QP yang dikaji menggunakan perolehan spektrum. Untuk Fasa 2, LP dikomplekskan dengan WP, CP, atau QP pada pelbagai nisbah untuk mengkaji pengaruh nisbah dan jenis protein pada struktur, kelarutan molekul pengikatan air dan sifat antaramuka kompleks protein. Melalui nisbah terbaik untuk kompleks protein, kelarutan LP meningkat dengan ketara (P < 0.05) hingga mencapai 92.7, 76.8, dan 86.5% setelah kompleks pada nisbah 1: 1.2 (untuk LP-WP), 1:1 (untuk LP-QP), dan 1:0.1 (untuk LP-CP), masing-masing. Berdasarkan keputusan kami melalui analisis morfologi, pendarfluor dan spektrum UV, dan FTIR, LP berinteraksi dengan WP, CP, atau QP untuk membentuk struktur protein baru. Akhirnya, dalam Fasa 3, LP, QP, WP, CP dan kompleks protein berasaskan LP yang dipilih diperam menggunakan biji kefir air selama 5 hari. Dengan menggunakan proses fermentasi ini, peningkatan yang ketara (P<0.05) dalam pencernaan kompleks protein LP, LP-QP-1:1, LP-WP-1:1.2, dan LP-CP-1:0.1 meningkat dari 76.42 ke 84.1%, dari 76.46 ke 87.2%, dari 81.69 ke 93.0%, dan dari 79.53 ke 86.7%, masing-masing, selama tempoh penapaian. Jumlah kandungan saponin (TSC) kompleks protein LP, LP-WP-1:1.2, LP-QP-1:1, dan LP-CP-1:0.1 dikurangkan dengan ketara (P<0.05) pada nisbah 24.0, 16.5, 18.8, 30.0, dan 19.6%, masing-masing, semasa tempoh penapaian. Kompleks protein berasaskan LP yang ditapai berkembang menjadi kompleks protein baharu dengan struktur tersusun semula, menghasilkan kebolehcernaan yang lebih tinggi dan menurunkan TSC. Kesimpulannya, gabungan pengkompleksan protein dan teknik penapaian WKS berjaya meningkatkan sifat fungsian dan pemakanan untuk LP, QP, dan kompleks protein berasaskan LP.

STRUCTURAL INTERACTIONS AND FERMENTATION OF LENTIL PROTEIN WITH OTHER PROTEINS TO IMPROVE THE FUNCTIONAL AND NUTRITIONAL PROPERTIES

ABSTRACT

In recent years, there has been increasing interest in plant-based proteins such as lentil proteins (LPs) due to their good availability, low cost, and global production. The main factors limiting the use of lentil proteins in the food industry are its low water solubility, poor digestibility, and high amount of saponins, which leads to bitterness. Protein-protein interactions (PPI) and fermentation have been used in the past to address some of these limitations in plant-based proteins. Protein complexation and WKS fermentation procedures were combined successfully to improve the functional and nutritional aspects of LP-based protein complexes. The aim of this study is to enhance the solubility and digestibility of LPs, as well as the degradation of its saponins content through PPI and fermentation. In order to achieve this aim, this study is separated into 3 phases. For Phase 1, the objective was to compare the protein structural and solubility and interfacial properties of 4 different types of protein: LPs, quinoa proteins (QPs), casein proteins (CPs), and whey proteins (WPs). Solubility of WPs, QPs, and CPs was higher than that of LPs, at 92.7, 80.1, 89.7, and 58% (P<0.05), respectively. According to our findings, WPIs, QPs, and CPs have more different protein structures and interfacial properties than LPs. While the surface hydrophobicity of LPs was higher more than WPs, CPs and QPs, around 618, 24, 236, and 404 a.u. (P<0.05), respectively. On other hand, the WPIs, CPs, and QPs have higher surface charges than LPs, approximately -33.2, -30.9, -32.9, and -22.9 (P < 0.05), respectively. According to our findings, the rate tertiary structure of LPs

was higher than that of WPIs, CPs, and QPs studied using spectral acquirements. For Phase 2, LPs were complexed with WPs, CPs, or QPs at various ratios in order to study the effect of ratio and types of protein on the structural, molecular binding water solubility and interfacial properties of the protein complexes. Through the best ratio of protein complexes, the solubility of protein complexes increased significantly (P < 0.05) to reach 92.7, 76.8, and 86.5% after complexation at ratio 1:1.2 (for LP-WP), 1:1 (for LP-QP), and 1:0.1 (for LP-CP), respectively. Based on our findings using morphological analysis, fluorescence and UV spectra, and FTIR, the LPs interacted with WP, CP, or QP to form novel protein structures. Finally, in Phase 3, the LPs and selected LP-based protein complexes were fermented using water kefir seeds for 5 days. Using this fermentation process, a significant ($P \le 0.05$) increase in the digestibility of LPs, LP-QP-1:1, LP-WP-1:1.2, and LP-CP-1:0.1 protein complexes increased from 76.42 to 84.1%, from 76.46 to 87.2%, from 81.69 to 93.0%, and from 79.53 to 86.7%, respectively during the fermentation period. Total saponins content (TSC) of LPs, LP-WP-1:1.2, LP-QP-1:1, and LP-CP-1:0.1 protein complexes was reduced significantly (P<0.05) at ratio 24.0, 18.8, 30.0, and 19.6%, respectively, during the fermentation period. Fermented LP-based protein complexes developed into new protein complexes with reorganised structures, resulting in higher digestibility and decreased TSC. In conclusion, the combination of proteins complexation and WKS fermentation techniques were successful in improving the functional and nutritional properties of LPs, QPs, and LP-based protein complexes.

CHAPTER 1

INTRODUCTION

1.1 Background of Study

In the last decade, demands for high nutritional value-added food products and plant-based protein have increased in line with the growth of the human population and consumers' health awareness (IFT, 2020). Many efforts have been focused on replacing animal-based proteins with plant-based proteins (Sha & Xiong, 2020; Wild et al., 2014). Protein is an essential component of human nutrition because they are highly important to the recovery of damaged cells, immune response, cell signals, and the maintenance of muscle mass (Gombart et al., 2020) Plant- and animal-based proteins are the two important dietary sources of protein. However, a diet rich in animal-based proteins is also associated with higher saturated fat intake (Segovia-Siapco et al., 2020) and salt (Petit et al., 2019). Overconsumption of meat, especially processed meat, can lead to health problems (Bujnowski et al., 2011; Kafatos, 2009), coronary heart disease (Clifton, 2011; Richi et al., 2015), high blood pressure (Elliott et al., 2006; Richi et al., 2015; Tuso et al., 2013), and increased serum and urine uric acid (Tracy et al., 2014). On contrary, various studies have shown plant proteins have had several beneficial health effects, such as anti-cancer activities (Bingham, 1999; James et al., 2017) enhancing metabolic dysfunction due to obesity (Dinu et al., 2017; Tonstad et al., 2009; Tuso et al., 2013; Wanezaki et al., 2015), preventive effects against cardiovascular diseases (Kim et al., 2019; Kwok et al., 2014; Marsh et al., 2012; Tuso et al., 2013) and nutritional support for cirrhotic patients (Kim et al., 2019; Tuso et al., 2013; Bianchi et al., 1993). The global production of plant-based protein in 2013 was about 6,452 thousand tonnes, and in 2018 it was approximately 8,541.2 thousand tonnes, with the production of plant-based protein forecasted to be around 10,555 thousand tonnes by 2023 (Kerry, 2021). The demand for plant-based protein in the United States of America (USA) increased, whereby the USA consumers' appetite for plant-based protein e.g., lentil and quinoa, increased around 44 and 36% in 2020, respectively (IFT, 2020). Lentil (*Lens culinaris*) is an edible legume with a high amount of protein content, between 20.6 and 31.4 g/100 g (Jarpa-Parra et al., 2014), as well as possessing favourable ratios of leucine to isoleucine (1.24–1.98) and leucine to lysine (1.08–2.03) (Stevenson et al., 2007). It is also reported to be high in fibre and low in fat (Reif et al., 2021). However, lentil proteins (LPs) have poor solubility in the water, with their solubility reported to be approximately 55% (Jarpa-Parra et al., 2014).

Solubility is considered one of the most important functional properties on which other functional properties rely upon. Thus, considerable efforts have focused on improving the solubility of proteins by preparing soluble biopolymer composites (He et al., 2020b; Wang et al., 2018a). Whey protein isolates (WPIs) and casein proteins (CPs) have high solubility and nutritional value, they have been a popular use in food industries. Many researchers have revealed that WPIs and CPs can interact with other plant-based proteins, such as rice proteins (Lin et al., 2017; Wang et al., 2019a; Wang et al., 2018b). In addition, quinoa proteins (QPs) are a new trend in food industries because of their high solubility and an excellent source of amino acids, as well as having a high level of solvent-protein interaction capabilities (Dakhili et al., 2019; Elsohaimy et al., 2015; Suárez-Estrella et al., 2021). Hence, it will be interesting to investigate how the functional properties of LPs upon protein complexation with QPs. The interaction between animal- and plant-based proteins is being comprehensively investigated to develop balanced mixtures of protein complexes. The protein structure is a feature of the interaction of protein components (Mezzenga et al.,

2005; Ubbink et al., 2008). Molecular forces that are involved in protein-protein interaction (PPI) include non-covalent ones, such as electrostatic, hydrogen and hydrophobic forces (He et al., 2020a; Nicolai, 2019; Sun et al., 2018), van der Waals forces (Karplus & Šali, 1995; Howell, 1992; Barclay & Ottewill, 1970), steric and hydration repulsive forces (Howell, 1992), and covalent disulphide linkages (Lin et al., 2019; Wang et al., 2017). These forces have a major role in modifying the protein microstructure and functional properties, especially on the solubility of the protein (He et al., 2020a; Wang et al., 2019a).

PPI is considered one of the best techniques to enhance the functional properties, such as solubility, emulsion, and foaming, of plant protein (He et al., 2020b; Sun et al., 2018; Wang et al., 2019a; Wang et al., 2018a). The functional properties of a protein are influenced by surface hydrophobicity more than hydrophobicity because of the polymeric aspect of the protein (Chandrapala et al., 2011; Poncin-Epaillard et al., 2012). The hydrophobicity of the protein surface plays a major role in solubility, physical stability, adsorption behaviour, and the aggregating tendency of the protein (He et al., 2020a; Nicolai, 2019; Wang et al., 2019a). The surface hydrophobicity of a protein is determined by the length of the chain in the protein through hydrogen bonding (Narbutt, 2020). Whereby the PPI effect on the surface charge and surface hydrophobicity of protein via electrostatic and hydrophobic forces and hydrogen bonds (He et al., 2020a; Nicolai, 2019; Wang et al., 2019a).

The solubility of 7S and 11S globulins of LPs depend on the degree of dissociation, aggregation, denaturation, composition, and structure (Tezuka et al., 2004; Renkema et al., 2000; Puppo & Afñón, 1999). The pH of an aqueous solution is also a major factor in the solubility of the protein. Non-covalent forces, such as hydrophobic and electrostatic interactions, govern the solubility of protein between

molecules. The solubility of protein is improved when the electrostatic repulsion between molecules is greater than hydrophobic interactions (Wang et al., 2019a; Mills et al., 2001; Utsumi, 1992). Hydrophobic interactions in an aqueous solution can also enhance PPI between LPs and WPIs, QPs, or CPs that could lead to an increase in the solubility of LPs after PPI.

Another key limitation of the use of plant protein in the food industry is its low digestibility resulting from its densely packed rigid system that is mainly due to hydrophobic interactions and disulphide bonds (Elsohaimy et al., 2015; Singh & Singh, 2016; Utsumi, 1992). Digestibility is often used as an indicator to evaluate the proteolysis capability and availability of a protein. Compared with proteins with low digestibility, proteins with high digestibility are considerably more suitable because of their high bioavailability, thus higher nutritional benefit. Fermentation is a suitable technology for biochemical changes, especially on the bio-accessibility and bioavailability of the food structure, as a result of acetic acid and lactic acid bacteria, such as Streptococcus, Leuconostoc, Lactococcus, and Lactobacillus spp., as well as their enzymatic actions (Azi et al., 2020; Tu et al., 2019; Nkhata et al., 2018; Randazzo et al., 2016; Hotz & Gibson, 2007). Fermentation can lead to a decrease in nonnutritive compounds, including trypsin, and promotes the cross-linking of proteins, as well as promoting proteolysis and release of the proteins from the matrix, thus improving the digestibility of plant-based proteins (Chandra-Hioe et al., 2016; Reyes-Moreno et al., 2004; Shekib, 1994). Çabuk et al. (2018) investigated the effect of fermentation on the digestibility of pea protein and found out that the fermentation increased the protein digestibility from 80.0 to 83.2% after 11 h with Lactobacillus plantarum fermentation.

Furthermore, fermentation can also enhance the nutritional value (increased total phenolic content) and decrease anti-nutrients (compounds that inhibit digestibility, such as trypsin and phytate). Both studies by Azi et al. (2020) and Tu et al. (2019) showed that kefir fermentation decreases the trypsin and phytate of soybean and then improves the digestibility of soybean protein. Kefir has become a very popular microbial consortium because of its excellent composition of microorganisms with probiotic function (Fiorda et al., 2017). This consortium is composed mainly of lactic acid bacteria, acetic acid bacteria, and yeast (Azi et al., 2020). Recently, Yang et al. (2020) reported enhancement of the digestibility of pea protein after 24 h fermentation with *Aspergillus oryzae*, *Bacillus licheniformis*, *Bacillus subtilis*, *Lactobacillus casei*, and *Lactobacillus plantarum*. Furthermore, fermentation could also degrade saponins, which are responsible for the bitterness in plant-based proteins (Suárez-Estrella et al., 2021). According to Lai et al. (2013), a 24 h lactic acid fermentation can reduce the total saponins content in soy milk.

Microorganisms that produce enzymes responsible for the cleavage of ester bonds, resulting in the release of bound phenolics. Most of the non-nutritive compounds, such as phenolic compounds, build ether and ester linkages with lignin through their hydroxyl groups in the aromatic ring, and ester linkages with structural carbohydrates and proteins via their carboxylic group (Bhanja et al., 2009; Liyana-Pathirana & Shahidi, 2006; Bartolomé et al., 1997). The cleavage of these ester bonds leads to an increase in protein digestibility and releases of free-phenol compounds that are water-soluble.

Therefore, PPI may improve the solubility of LPs and can be used to prepare novel complex protein composites of LPs with WPIs, QPs or CPs. Subsequently, these protein complexes can be subjected to a fermentation process using water kefir seeds to enhance the digestibility and reduce the bitterness of these complexes. Various protein-protein interactions were studied in recent years on plant-based proteins. He et al. (2020b) explored the use of mixed soy protein and wheat gluten proteins to increase protein solubility. Wang et al. (2019a) also found that a mixture of 1% rice proteins and whey protein isolates at a ratio of 1:1 increased the solubility to over 50% based on the synergistic interaction; hydrophobic, hydrogen, and ionic interactions, which may be taking place. These molecular forces may have collaboratively favoured the formation of the protein complexes. Yan et al. (2019) showed that the complexation of WPIs with lactoferrins can control the rheological properties of the emulsion gels of protein complexes by varying the pH through non-covalent bonds, such as electrostatic interactions between droplets. Wang et al. (2018b) reported that a mixture of 1% rice proteins and soy protein isolates, increased the solubility to over 80% through synergistic interaction.

LPs are a strong alternative to soy protein because they are not genetically engineered and do not cause allergic problems. Developing a design for generating fermented LPs and protein complexes of LPs with WPIs, CPs or QPs with improved nutritional properties is extremely desirable for food technology and industrial food production.

1.2. Rationale of Study

1.2.1 Aim and Objectives

The aim of the study is to produce LPs-based protein complexes with enhanced properties (e.g. water solubility, digestibility, and bitterness) via the aid of synergistic interaction between LPs and other proteins (WPIs, QPs, and CPs). In order to achieve this aim, the study was carried with the following objectives:

- To investigate protein quality, microstructure, solubility in water, protein structure, secondary protein structure, and interfacial properties of LPs, QPs, WPIs, and CPs.
- To determine the effects of ratio and types of proteins (WPIs, QPs, and CPs) on the microstructure, water-solubility, protein structures, molecular bindings, and interfacial properties of LPs-based protein complexes using PPI.
- 3. To determine the effect of water kefir seeds fermentation on the digestibility and saponins content of the LPs-based protein complexes.

1.2.2 Problem Statements

 Lentil proteins have poor solubility in water. The solubility of lentil proteins were reported to be around 55% at pH 7.0, compared to 93, 90, and 80% for WPIs, CPs, and CPs, respectively.

- 2. Lentil protein has poor digestibility (approximately 77%) compared to dairybased proteins, such as 93 and 82% for WPIs and CPs, respectively.
- 3. Saponins are found in many edible legumes. Lentils have saponins content, between 700–1,260 mg/kg, which is responsible for the bitterness in lentils.

1.2.3 Hypotheses

- 1 Study basic solubility and structure of lentil proteins and their application in food production.
- 2 Study the effect of ratio and types of different proteins on the structural and solubility of lentil protein based on protein complexation.
- 3 Identifying effect fermentation on complexes protein using water kefir grains, to increase digestibility and decrease the bitterness of lentil proteins.





Figure 1.1. Schematic diagram of the framework of the study: Phase I, Phase II, and Phase III

LPs = lentil proteins; WPIs = whey protein isolates; CPs = casein proteins; QPs = quinoa proteins; WKS = water kefir seeds; DW = distilled water

CHAPTER 2

LITERATURE REVIEW

2.1 Plant-Based Proteins

With the increasing popularity of plant-based diets and growing environmental issues, there has been a growing interest in the use of alternative protein sources (Lin et al., 2019; Sun et al., 2018) and a decrease in animal-based protein diets. Thus, plant-based proteins (e.g. quinoa, soya, and lentil) as an alternative to animal-based proteins (e.g. whey protein isolates, caseins, and gelatines) have gained widespread interests in recent years due to their lowest possible environmental footprints as compared to those related to the development and consumption of animal-based proteins (Jędrejek et al., 2016; Koneswaran & Nierenberg, 2008; Zheng et al., 2020). As shown in Table 2.1, animals, plants, and fungi are the three main dietary sources of protein. Interactions between plant proteins with other types of protein undoubtedly occur in different ways. These interactions could have significant impacts on the structural and functional properties of plant proteins due to the molecular forces, e.g. hydrogen bonds and hydrophobic and electrostatic forces (He et al., 2020a).

	y טו וטטע ףוטנכוונא אטר		
Sources of Protein		Types and Ratio of Proteins	References
Animal	Milk	Whey 15–20% [β -lactoglobulin (35–65%), α -lactalbumin (12–25%); immunoglobulins (8%], albumin (5%), and lactoferrin (1%)] Casein [80%) (α S1-casein (40%), α S2-casein (10%), κ -casein (15%), and β -casein (35%)]	(Camargo et al., 2018; Ma et al., 2017b) (Beliciu & Moraru, 2013)
	Egg	Ovoglobulin (G2 and G3, 8%), ovotransferrin (12%), lysozyme (3.5%), ovomucin (3.5%), ovalbumin (54%), and ovomucoid (11%)	(Abeyrathne et al., 2018)
	Blood	Bovine serum albumin, haemoglobin, and ferritin	(Alimam et al., 2018)
	Meat	Elastin, myosin, actin, and collagen	(Hurrell et al., 2006)
	Insect	Fibroin, resilin, chemosensory, Sf caspase-1, major royal jelly protein, melittin, myoblast city, doublesex, and transfer genes	(Ardell & Andersen, 2001; Angeli et al., 1999)
Plant Cereals	Wheat	Albumin; globulin; glutelin (80%); prolamin/gliadin	(Shewry et al., 2009)
	Corn	Albumin; globulin; glutelin (18%); prolamin/zein (60%) (α- zein, β-zein, γ-zein, δ-zein)	(Hojilla-Evangelista, 2012; 2014; Muthukumarappan & Swamy, 2018; Zhuang et al., 2013)
	Barley	Albumin (22%); globulin; glutelin (47–27%); prolamin (35– 55%)	(Kirkman et al., 1982)

Table 2.1 Summary of food proteins sources commonly used in the food industry

Table 2-1. Continued			
Sources of Protein		Types and Ratio of Proteins	References
	Oats	Albumin; globulin (75%); glutelin (4–14%); prolamin (10%)	(Friesen, 2017; Webster, 2016)
	Rice	Albumin (5–10%), globulin (7–17%), glutelin (75–81%) and prolamin (3–6%)	(Amagliani et al., 2017)
Legumes and pulses	Soybeans	2S globulin (17%), 7S globulin (33%), 11S globulin (45%); 15S globulin (5%)	(Singh et al., 2010: Amagliani et al., 2017)
	Peas	2S globulin (25%), 7S globulin (43%), 11S globulin (28%); 15S globulin (4%)	(Amagliani et al., 2017)
	Mung bean	2S globulin (15%), 7S (79%) globulin; 11S globulin (6%)	(Amagliani et al., 2017)
	Lima bean	2S globulin (52%), 7S globulin (21%); 15S globulin (27%)	(El Fiel et al., 2002)
	Chickpea	Albumin (8.39–12.31%), globulin (53.44–60.29%) prolamin (3.12–6.89%); glutelin (19.38–24.40%).	(Portari et al., 2005)
	Lentil	Albumins (16.8%), globulins (11S/legumin (44.8%), 7S/vicilin (4.2%)), glutelins (11.2%); prolamins (3.5%).	(Boye et al., 2010)
Tubers	Potato	Sporamin +A N-termina (80%), glycoprotein, arabinogalactan	(Wang et al., 2016)
Oilseeds	Rapeseed, peanut, sunflower, hemp seed	Albumin + globulin (60%), gluten	(Bur et al., 2004)

Sources of 1	Protein		Types and Ratio of Proteins	References
		Cottonseed	Globulins (2S, 7S/congossypin, 12S)	
	Edible seeds	Quinoa, buckwheat	albumin (35%), globulin (37%), prolamins (0.7–7%)	(Steffolani et al., 2016)
	Pseudocereals	Chia	Albumin (22.8%), globulin (22.4%), glutelin (48.9%) prolamin (5.8%)	(Attalla & El-Hussieny, 2017)
	Algae	Micromonas		(Simon et al., 2017)
Fungi		Mycoprotein		(Stoffel et al., 2019)

2.1.1 Lentil Proteins (LPs)

Legume seeds that are mature are usually rich in proteins although they produce these proteins during their growth (Bennetau-Pelissero, 2019). They are primarily processing proteins categorised into glutelins, albumins and globulins based on their solubility behaviour, with salt-soluble globulins constituting the majority of the fraction contained in the pulses (Kiosseoglou & Paraskevopoulou, 2011). Globulin is classified into two forms, vicilin or legumin, called fraction 7S or fraction 11S, respectively (Jarpa-Parra et al., 2017). Albumin has a solubility in water about 10–20% of the legume proteins. At the same time, glutelin has a solubility in acid and base solution of approximately 10–20% of legumin proteins (Shewry & Tatham, 1997). Lentil seeds (*Lens culinaris*) have a high amount of protein between 20.6 and 31.4 g/100 g (Jarpa-Parra et al., 2014), are low in fat, and are high in fibre (Urbano et al., 2007).

2.1.1(a) Chemistry and Structure of LPs

LPs are composed of four protein fractions; 70% (7S and 11S) globulins, 16% albumins, 11% glutelins, and 3% prolamins (Boye et al., 2010). Glutelins, albumins, and prolamins, which have polypeptides around 4, 13, and 10, respectively. These proteins have low molecular weights (MWs), approximately 17–46, 20, and 16–64 kDa, respectively. Globulins are divided into two fractions: legumin and vicilin. Legumin has a molecular weight of 320–380 kDa and is made up of 6 polypeptides. They consist of a 20 kDa basic dimer and 40kDa acidic dimer bound by a disulphide bond (Shewry & Tatham, 1997), whereas vicilin has MWs ranging between 50 and 60 kDa, made up of subunits of glycosylated, and is usually referred to as 7S fraction

(Aydemir & Yemenicioğlu, 2013; Bhatty, 1988; Argos et al., 1985). The glycosylated subunits contain carbohydrates, which are made up of around 2.8% (Barbana & Boye, 2011; Bamdad et al., 2009; Martinez & Urbano, 2007).

2.1.1(b) Quality and Health Benefits of LPs

Interest in LPs has recently increased because of their excellent nutritional benefits, which are suitable leucine/isoleucine and leucine/lysine ratios; 1.24–1.98 and 1.08–2.03 mg/g, respectively (Urbano et al., 2007). According to the Food and Agriculture Organisation (FAO) and the World Health Organisation (WHO), the recommended ratios of essential amino acids of leucine/isoleucine and leucine/lysine are 1.3-1.6 and 1.3-1.6 mg/g, respectively (FAO/WHO, 1991). The dietary qualities of lentils are related to reduced cholesterol and lipid levels in humans and a low risk of colon cancer and type 2 diabetes (Roy et al., 2010). LPs have high contents of tannins, flavonoids, and phenolics, with values of 5.97 mg of catechin equivalents/g, 5.97 mg of catechin equivalents/g, and 6.56 mg of gallic acid equivalents/g, respectively (Jarpa-Parra et al., 2017).

2.1.1(c) Functional Properties of LPs

Techniques used in protein extraction and its conditions have a direct effect on the functional properties, physicochemical properties, and composition of proteins, as well as the protein structure (Farooq & Boye, 2011). In general, protein functionalities, such as solubility, emulsifying, foaming, and water- and oil-absorption capabilities have been extensively studied (Table 2.1).

Protein solubility in water is an essential functional property because it affects most other functional characteristics (Joshi et al., 2011). Therefore, the profile of protein solubility in various pH levels can be a major predictor of proteins in food applications and the degree of protein denaturation affected by extraction. Studies by Jarpa-Parra et al. (2014), Boye et al. (2010), and Bora (2002) reported that LPs have lower solubility in water between pH 4.0 and 6.0 but have higher solubility at lower and higher pH ranges. The solubility levels of LPs are 17–49% at pH 3.0 and 55–58% at pH 7.0 (Jarpa-Parra et al., 2014; Joshi et al., 2011). Therefore, LPs have poor solubility of approximately 55% in water (Jarpa-Parra et al., 2014). Solubility is an important functional property that affects its other functional properties. Thus, considerable efforts have focused on the solubility of proteins to prepare soluble biopolymer composites (Apichartsrangkoon et al., 1998; He et al., 2020b; Kong et al., 2007; Wang et al., 2019b). The solubility of 7S and 11S globulins depends on the degree of dissociation, aggregation, denaturation, composition, and structure (Tezuka et al., 2004; Renkema et al., 2000; Puppo & Afñón, 1999).

Protein Type	Functional Properties	Description	Outcomes	References
LP_S	Solubility	LPs extract at pH 8.5	15 to 35% at pH 2.0	(Bora, 2002)
		Protein solution (LPs 15 mg/mL)	55% at pH 7.0; 5% at pH 3.5; 98% at pH 8.0; 95% at pH2.5	
		Different genotypes of LPs	30% at pH3.0 and 57% at pH7.0 31% at pH3.0 and 56% at pH 7.0	(Ma et al., 2011)
		LPs extract at pH 9.0 Protein solution (0.5%)	50% at pH7.0	(Joshi et al., 2011)
		LPs extract at pH 10	97% at pH 10.0; 96% at pH 1.0; 5% at pH 5.0	(Jarpa-Parra et al., 2014)
		LPs extract at pH 9.0	5% at pH 5.0; 82% at pH 1.0; 97% at pH 10.0	
		LPs extract at pH 8.0	5% at pH 5.0; 96% at pH 10; 78% at pH 2.0	
	WAC	LPs extract at pH 8.5 Protein solution (LPs 15 mg/mL)	1.1 mL/g of protein	(Bora, 2002)
		LPs extract at pH 9.0	3.8 mL/g of protein	(Boye et al., 2010)
			0.4 g/g of protein at pH 7.0	(Joshi et al., 2011)

Table 2.2 Summary outcomes of functional properties of lentil proteins (LPs), quinoa proteins (QPs), whey protein isolates (WPIs), and casein

Protein Type	Functional Properties	Description	Outcomes	References
	OAC	LPs extract at pH 8.5	2.6 mL/g of protein	(Bora, 2002)
		Protein solution (LPs 15 mg/mL)		
		LPs extract at pH 9.0	120%	(Boye et al., 2010)
	EAI	LPs extract at pH 8.5	54%	(Bora, 2002)
		Protein solution (LPs 15 mg/mL)		
		LPs extract at pH 8.0	41% to 47 at pH 7.0	(Lee et al., 2007)
		LPs extract at pH 9.0	$5.0-5.9 \text{ m}^2/\text{g}$ of protein at pH 7.0	(Boye et al., 2010)
		LPs extract at pH 10.0	44 m ² /g of protein at pH 7.0	(Karaca et al., 2011)
		Prepared by isoelectric precipitation	90 m ² g ⁻¹ protein at pH 7.0	(Joshi et al., 2012)
	ESI	LPs extract at pH 8.5	52%	(Bora, 2002)
		Protein solution (LPs 15 mg/mL)		
		LPs 15 mg/mL	83 to 89% at pH 7.0	(Lee et al., 2007)
		LPs extract at pH 9.0	18 min at pH 7.0	(Boye et al., 2010)
		LPs 10 mg/mL	101 h at pH 7.0; 92 h at pH 6.0; 27h at pH 5.0, 245 h at pH 3.0	(Joshi et al., 2012)

Table 2-2. Continued

Protein Type	Functional Properties	Description	Outcomes	References
QPs	Solubility	QPs extracted at pH 9.0	30% at 3.0 pH solution, 60% at pH 8.0 and 75% at pH 10	(Elsohaimy et al., 2015)
		QPs extracted at	100% at pH 10.	(Paredes-López et al., 1993)
			76% at pH 7.0	(Shen et al., 2021)
	WAC	QPs extracted at pH 9.0	3.94 mL/g protein	(Elsohaimy et al., 2015)
	DAC	ODe extracted at nH 0 0	1 88 ml /a nrotein	(Elechaimy et al 2015)
		Protein solution (3%)	1	
	EAI	QPs extracted at pH 9. Protein solution (3%)0	$3.38 \pm 0.31 \text{ m}^2/\text{g of protein}$	(Elsohaimy et al., 2015)
		QPs extracted at pH 9.5 Protein solution (3%)	40.60% at pH 5; 44.72% at pH 6; 56.63%at pH 7; 61.01% at pH 8	(Shen et al., 2021)
		Protein solution (1%)	85.45%.	(Sánchez-Reséndiz et al., 2019
		QPs extracted at pH 9.0	30.4 and 46.34	(Elsohaimy et al., 2015)
			30.18 at pH 5; 42.04 at pH 6; 51.90 at pH 7; 57.98 at pH 8	(Shen et al., 2021)

Table 2-2. Continued

Protein Type	Functional Properties	Description	Outcomes	References
WPIs	Solubility	ND	95% at pH 7.0	(Floris et al., 2008)
			90% at pH 7.0	(Zembyla et al., 2021)
			95% at pH 7.0	(Gani et al., 2015)
	EAI	ND	30 min at pH 7.0	(Xu et al., 2019a)
			45 30 min at pH 7.0	(Ping-Ping et al., 2020)
			$7.07 \text{ m}^2/\text{g}$	(Chen et al., 2019)
			66.05%.	(Chen et al., 2019)
			35%	(Ping-Ping et al., 2020)
CPs	Solubility	ND	95% at pH %	(Gani et al., 2015)
		pH 7.0	55%	(Augusta et al., 2007)
	ESI	ND	45.13 min.mL oil/g protein	(Gani et al., 2015)
		pH 7.0	$1885 \pm 34.20 \text{ nm at pH } 7.0$	(Li & Zhao, 2019)
		pH 7.0	43 min.mL oil/g protein	(Augusta et al., 2007)
		concentration 1%	0.08g/mg	(Augusta et al., 2007)
	EAI	pH 7.0	103 m ² /g	(Augusta et al., 2007)
	WAC	ND	18.48 mL/100g.	(Gani et al., 2015)
WAC= wate	r absorption capac	ity; OAC= oil absorpti	on capacity; EAI= emulsion acti	vity index; ESI= emulsion stability index; ND= not
		_		

determined; NM= pH value not mentioned

20

The concepts oil absorption capacity (OAC) and water absorption capacity (WAC) are usually used to refer to the amount of oil and liquid which could be absorbed per gram of sample, respectively. WAC values are good indicators of a protein's ability to resist liquid release from food during processing, transportation or storage (Farooq & Boye, 2011; Kiosseoglou & Paraskevopoulou, 2011). The OAC and WAC rates of LPs are commonly used to describe the outcomes. For instance, both Boye et al. (2010) and Bora (2002) reported that WAC level at the range between 1.1 and 4.2 mL/g of protein. Meanwhile, Lee et al. (2007) found that WAC of LPs at the range between 0.43 to 0.49 mL/g. For OAC, Bora (2002) found that the OAC for LPs to be between 2.0 to 2.6 mL/g. Meanwhile, according to Boye et al. (2010), the OAC of LPs ranges from 120 to 225%, depending on the technique used in the protein extraction. There has not been a great deal of research recently on the mechanisms and compounds involved in both OAC and WAC.

The emulsification capacity of a protein is its capacity to aid in the preparation of an emulsion with oil droplets of a relatively small size via creating a film around them. These droplets need to be distributed in the liquid suspension to avoid droplet accumulation that may lead to phase separation (Karaca et al., 2011). Protein emulsification characteristics are typically defined by two parameters: emulsifying activity index (EAI) and emulsifying stability index (ESI). EAI is the highest area produced from each unit of protein and this is typically defined with high-dilute emulsion samples through the application of turbidimetry. ESI evaluates the emulsion's ability to maintain its shape over even a given time period, and it is measured by monitoring turbidity decreases with diluted emulsion during storage (Kiosseoglou & Paraskevopoulou, 2011). Studies on the emulsifying properties of LPs show a lot of variation depending on the processing and protein extraction conditions (Table 2.1). Therefore, the comparison of the findings becomes extremely difficult. Generally, the emulsifying properties of LPs are close to those of other pulses, varying from 5 to 90 m²/g for EAI (Joshi et al., 2012; Karaca et al., 2011; Boye et al., 2010). While the ESI of LPs are varying from 19 min to several days depending on the conditions of the method used. Except for Joshi et al. (2012), who measured both EAI and ESI in the pH range of 3.0 to 7.0, the majority of the studies on the EAI and ESI of LPs were conducted at pH 7.0.

2.1.2 Quinoa Proteins (QPs)

Quinoa seeds (*Chenopodium quinoa* Wild under the family Chenopodiaceae) are a great nutrient source as recognised by the FAO (FAO, 2011). Quinoa seeds are healthy ingredients and are characterised as food that reduces the risk of several diseases and has health-promoting characteristics (Repo-Carrasco et al., 2003). The FAO (2017) reported that quinoa seeds contribute to agricultural productivity in the 21st century because of their tolerance to salinity and stress conditions and potential to grow in marginal areas. Scientists have attempted to use quinoa seeds as a replacement for animal-based proteins (Ruiz et al., 2016; Miranda et al., 2010; Paśko et al., 2009; Drzewiecki et al., 2003). The seed contains 12–20% proteins (Toapanta et al., 2016; Elsohaimy et al., 2015; Abugoch et al., 2008).

2.1.2(a) Chemistry and Structure of QPs

The main fractions of quinoa protein are prolamins (0.5-7.0%), albumins (35%), and globulins (37%) (Kaspchak et al., 2017; Abugoch et al., 2008). The

solubility of QPs depends on the pH of the solution (Elsohaimy et al., 2015). The globular chenopedin has a hexamer structure and consists of six pairs of basic and acidic polypeptides with MWs of 22–23 and 32–39 kDa, respectively, linked by a single disulphide bond (Dakhili et al., 2019; Ruiz et al., 2016).

2.1.2(b) Quality and Health Benefits of QPs

QPs have higher amounts of essential amino acids than proteins in other cereals, such as rice, maize wheat, and barley (Zhu, 2020; Montaño et al., 2006). QPs have 0.4–1.0% methionine and 5.1–6.4% lysine (Bhargava et al., 2007). Quinoa seeds also have higher concentrations of methionine, lysine, and cysteine than other legumes and cereals and are thus a good replacement for these crops. Quinoa seeds have high nutritional value because of their high amounts of minerals, such as potassium, copper, magnesium, iron, and calcium, as well as vitamins, such as vitamin B, E, and C (Konishi et al., 2004). In addition, quinoa seeds have high levels of fibre and antioxidants, such as phenolic compounds (Repo-Carrasco-Valencia & Serna, 2011; Hirose et al., 2010). Hence, quinoa seed may be a good potential to be included in the development of designer foods since it may help to reduce the risk of common diseases and promote human health (Shen et al., 2021).

2.1.2(c) Functional Properties of QPs

The solubility levels of QPs are about 30% at 3.0 pH solution, 60% at pH 8.0 and 75% at pH 10 (Elsohaimy et al., 2015; Dench, 1982; Kumar & Venkataraman, 1980). The balance of hydrophilic and hydrophobic components affects the solubility of QPs, which then can interact with solvent. At alkaline pH, the ionisation of carboxyl

groups and the deprotonation of amine groups produces negatively charged species, which improve the protein-solvent interaction and increase the protein solubility (Adebowale & Lawal, 2004; Pearce & Kinsella, 1978). General the QPs have low solubility at pH range 4-6, and its solubility increased in an alkaline environment (Ruiz et al., 2016; Steffolani et al., 2016; Elsohaimy et al., 2015; Bejarano-Luján et al., 2010; Tömösközi et al., 2008). Generally, QPs demonstrates low solubility at acidic pH (around 5% at pH 4–6), which is close to the isoelectric point (pI) of the 11S quinoa globulin (Paredes-López et al., 1993). The maximum solubility of QPs approximately 100% at pH 10.0. Both Shen et al. (2021) and López-Castejón et al. (2020) investigated the solubility of QPs at different pH values and reported that the solubility of QPs around 76% at pH 7.0. OAC and WAC are very important between functional properties of food because it enhances the retention of flavour and mouthfeel (Jarpa-Parra et al., 2017). Both Elsohaimy et al. (2015) and Ashraf et al. (2012) studied the WAC and OAC of QPs and found them to be around 3.94 ± 0.06 and 1.88 ± 0.02 mL/g protein, respectively. Recently, Sánchez-Reséndiz et al. (2019) studied the OAC of QPs and discovered that QPs have an OAC of about 2.66 mL/g.

Emulsion properties are essential functional properties of proteins that influence food product behaviour. EAI and ESI of QPs were listed in Table 2.2. Elsohaimy et al. (2015) investigated the EAI and ESI of QPs and discovered that the EAI of QPs at a 3% (w/v) ratio is about $3.38 \pm 0.31 \text{ m}^2/\text{g}$ and the ESI is between 30.4and 46.34-min. Sánchez-Reséndiz et al. (2019) studied the EAI of QPs and reported it to be around 85.45%. A recent study by Shen et al. (2021) found that the EAIs of QPs were 40.60, 44.72, 56.63, and 61.0 1% at different pH values; pH 5, 6, 7, and 8, respectively. Meanwhile, the ESI of QPs were reported to be 30.18, 42.04, 51.90 and 57.98% at pH 5, pH 6, pH 7 and pH 8, respectively.