EFFECT OF DIFFERENT FILLER PREPARATION METHOD ON PROPERTIES OF NATURAL RUBBER / PALYGORSKITE NANOCOMPOSITES

SITI NADZIRAH BINTI ABDUL MUTTALIB

UNIVERSITI SAINS MALAYSIA 2017

EFFECT OF DIFFERENT FILLER PREPARATION METHOD ON PROPERTIES OF NATURAL RUBBER / PALYGORSKITE NANOCOMPOSITES

by

SITI NADZIRAH BINTI ABDUL MUTTALIB

Thesis submitted in fulfillment of the requirements

for the degree of

Master of Science

ACKNOWLEDGEMENT

In the name of Allah, the Most Gracious and the Most Merciful.

Alhamdulillah, all praises to Allah for the strengths and His blessing in completing this thesis. First and foremost, I offer my sincerest gratitude to my supervisor, Assoc. Prof. Dr. Nadras binti Othman, for her supervision, guidance, patience, understanding and constant support during the research and writing this thesis. One simply could not wish for a better or friendlier supervisor. Special appreciations also go to my co-supervisor, Prof. Hanafi bin Ismail and PPKBSM lecturers who involved directly or indirectly in giving ideas and encouragements regarding the research. Apart from that, I would also like to express my gratitude to all technicians who have assisted me during laboratory work especially technicians at rubber laboratory. Sincere thanks also dedicated to all my friends for their kindness and moral support during my study. Thank you very much for the friendship and memories.

Finally, I would like to thank my respectful parents, Abdul Muttalib bin Nor and Siti Khatijah binti Yusoff and my siblings for their endless love, prayer and support. Thank you for always being there for me.

Siti Nadzirah binti Abdul Muttalib

TABLE OF CONTENTS

		Page
ACK	KNOWLEDGEMENT	ii
TAB	BLE OF CONTENTS	iii
LIST	T OF TABLES	ix
LIST	T OF FIGURES	X
LIST	T OF ABBREVIATIONS	xiii
LIST	T OF SYMBOLS	xvi
ABS	STRAK	xviii
ABS	STRACT	xix
CHA	APTER ONE: INTRODUCTION	
1.1	Introduction	1
1.2	Problem Statement	5
1.3	Objectives	6
1.4	Outline of the thesis	7
CHA	APTER TWO: LITERATURE REVIEW	
2.1	Polymer/Clay Nanocomposites	8
	2.1.1 Natural Rubber/Clay Nanocomposites	9
2.2	Natural Rubber	11
	2.2.1 Natural Rubber Latex (NRL)	11
	2.2.2 Natural Rubber	12
2.3	Layered Silicates	14

	2.3.1	Palygorskite Clay	14
2.4	Prepa	ration of Natural Rubber/Clay Nanocomposites	17
	2.4.1	Melt Mixing	17
	2.4.2	Latex Compounding	20
		2.4.2 (a) Clay Dispersion by Ball Milling	23
	2.4.3	Solution Blending	27
2.5	Surfac	ce Modification	30
	2.5.1	Silane Treatment	30
	2.5.2	Cation Exchange	33
2.6	Rubbe	er Compounding Ingredients	35
	2.6.1	Curing Agent	35
	2.6.2	Accelerator	35
	2.6.3	Activator	36
	2.6.4	Antidegradants	36
	2.6.5	Fillers	38
СНА	PTER T	THREE: METHODOLOGY	
3.1	Mater	rials	39
3.2	Equip	oments	39
3.3	Prepa	ration of Palygorskite Clay	41
	3.3.1	Dispersion of Palygorskite Clay	41
	3.3.2	Ball Milling	41
	3.3.3	Latex Compounding Method	42
3.4	Form	ulation of Rubber Compounding	42
3.5	Rubbe	er Compounding	43

	3.5.1	Preparation of NR/PAL Nanocomposites Using Conventional	
		Method	45
	3.5.2	Preparation of NR/PAL Nanocomposites Using Masterbatch	
		Method	45
3.6	Surfac	ce Modification	46
	3.6.1	Cation Exchange Treatment Using Octadecylamine (ODA)	46
	3.6.2	Cation Exchange Treatment Using 1-dodecyltrimethylamine	
		chloride (DDA)	46
	3.6.3	Cation Exchange Treatment Using 2-(methacryloyoxy)	
		ethyltrimethylammonium chloride (MQT)	47
3.7	Cure (Characterization Test	48
3.8	Vulca	nization	48
3.9	Mater	ials Characterization	49
	3.9.1	Fourier Transform Infrared Spectroscopy (FTIR)	49
	3.9.2	X-Ray Diffraction (XRD)	49
	3.9.3	X-Ray Fluorescence (XRF)	49
3.10	Swell	ing Test	50
	3.10.1	Crosslink Density	50
	3.10.2	Rubber-Filler Interaction	50
		3.10.2 (a) The Lorenz and Park's Equation	51
		3.10.2 (b) The Kraus Model	51
3.11	Mech	anical Test	52
	3.11.1	Tensile Test	52
	3.11.2	2 Tear Test	52
	3.11.3	Hardness Test	52

	3.11.4	Dynamic Mechanical Analysis (DMA)	53
3.12	Morpl	nology Test	53
	3.12.1	Field Emission Scanning Electron Microscopy (FESEM)	53
	3.12.2	Transmission Electron Microscopy (TEM)	53
СНА	PTER I	FOUR: RESULTS AND DISCUSSION	
4.1	Effect	of Different Filler Preparation Method on Properties	
	of Nat	tural Rubber/Palygorskite Nancomposites	
	4.1.1	X-ray Diffraction Analysis	55
	4.1.2	Curing Characteristics	57
	4.1.3	Swelling Properties	59
		4.1.3 (a) Crosslink Density	59
		4.1.3 (b) Rubber-Filler Interaction	59
	4.1.4	Mechanical Properties	61
		4.1.4 (a) Tensile Properties	61
		4.1.4 (b) Tear Strength	62
		4.1.4 (c) Hardness Test	62
		4.1.4 (d) Dynamic Mechanical Analysis	64
	4.1.5	Morphology Properties	68
		4.1.5 (a) Scanning Electron Microscopy	68
		4.1.5 (b) Transmission Electron Microscopy	70
4.2	Effect	of Different Ball Milling Parameters of Filler Preparation	
	on Pro	operties of Natural Rubber/Palygorskite Nanocomposites	
	4.2.1	Material Characterization	72
		4.2.3 (a) X-ray Diffraction	72

	4.2.2	Curing Characteristics	73
	4.2.3	Swelling Properties	75
		4.2.3 (a) Crosslink Density	75
		4.2.3 (b) Rubber-Filler Interaction	76
	4.2.4	Mechanical Properties	77
		4.2.4 (a) Tensile Properties	77
		4.2.4 (b) Tear Strength	79
		4.2.4 (c) Hardness Test	80
		4.2.4 (d) Dynamic Mechanical Analysis	81
	4.2.5	Morphology Properties	85
		4.2.5 (a) Scanning Electron Microscopy	85
4.3	Effect	of Different Cation Exchange Method Surface Treatment on	
	Proper	rties of Natural Rubber/Palygorskite Nanocomposites	
	4.3.1	Material Characterization	87
		4.3.1 (a) Fourier Transform Infrared Spectroscopy	87
	4.3.2	Curing Characteristics	89
	4.3.3	Swelling Properties	91
		4.3.3 (a) Crosslink Density	91
		4.3.3 (b) Rubber-Filler Interaction	92
	4.3.4	Mechanical Properties	93
		4.3.4 (a) Tensile Properties	93
		4.3.4 (b) Tear Strength	95
		4.3.4 (c) Hardness Test	96
		4.3.4 (d) Dynamic Mechanical Analysis	97
	4.3.5	Morphology Properties	100

	4.3.5 (a)	Scanning Electron Microscopy	100	
	4.3.5 (b)	Transmission Electron Microscopy	103	
CHAPTER FIVE: CONCLUSIONS AND FUTURE STUDIES				
5.1	Conclusion	ons	105	
5.2	Future St	udies	107	
REFERENCES		108		
LIST OF PU	UBLICATI	ONS		

LIST OF TABLES

		Page
Table 2.1	The physical and mechanical properties of natural rubber	13
Table 2.2	Different grades of SMR and its applications	14
Table 2.3	The properties of palygorskite	16
Table 2.4	Classification of accelerators and their characteristics	37
Table 3.1	List of equipments involved	40
Table 3.2	The formulation of PAL dispersion	41
Table 3.3	The ball milling parameters	41
Table 3.4	The formulation of rubber compounding ingredients	42
Table 3.5	The mixing sequence of rubber compounding	44
Table 4.1	The curing characteristics of NR/PAL composites of different filler loading	58
Table 4.2	The glass transition (T_{g}) and tan δ values of the NR composites	67
Table 4.3	The curing characteristics of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	75
Table 4.4	The tensile properties of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	79
Table 4.5	The curing characteristics of NR/PAL nanocomposites of different surface treatment	90

LIST OF FIGURES

		Page
Figure 2.1	The chemical structure of NR	13
Figure 2.2	Structural scheme of palygorskite	16
Figure 2.3	Structure of bifunctional silane coupling agent	31
Figure 2.4	The reaction between clay and alkyl ammonium ion	33
Figure 3.1	The flowchart of masterbatch preparation	45
Figure 3.2	The flowchart of NR/PAL nanocomposites preparation	54
Figure 4.1	Diffractogram of NR, PAL, and NR/PAL for both conventional and masterbatch methods	56
Figure 4.2	The crosslink density of NR and NR/PAL composites of various PAL loading	59
Figure 4.3	Kraus model for swelling test of nanocomposites for both conventional and masterbatch methods	60
Figure 4.4	Stress-strain curves of NR/PAL composites of different PAL loading	61
Figure 4.5	The tear strength of NR/PAL composites of various filler loading	63
Figure 4.6	The hardness of NR/PAL composites of various filler loading	63
Figure 4.7	The storage modulus of (a) conventional NR/PAL composites (b) masterbatch NR/PAL composites	66
Figure 4.8	The loss modulus of (a) conventional NR/PAL composites (b) masterbatch NR/PAL composites	66

Figure 4.9	The tan δ of (a) conventional NR/PAL composites (b) masterbatch NR/PAL composites	66
Figure 4.10	The micrographs of (a) NR, (b) PAL, (c) IN6 and (d) BM6	69
Figure 4.11	TEM micrographs of NR/PAL nanocomposites for conventional method with loadings of (a) 2phr, and (b) 6phr	71
Figure 4.12	TEM micrographs of NR/PAL nanocomposites for masterbatch method with loadings of (a) 2phr, and (b) 6phr	71
Figure 4.13	The diffractogram of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	73
Figure 4.14	The crosslink density of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	76
Figure 4.15	The rubber-filler interaction of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	77
Figure 4.16	The tear strength of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	80
Figure 4.17	The hardness of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	81
Figure 4.18	The storage modulus of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	84
Figure 4.19	The loss modulus of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	84
Figure 4.20	The tan δ value of NR/PAL composites of (a) different BPR, and (b) different ball mill sizes	84
Figure 4.21	The SEM micrograph of ball to weight ratio parameter at 1000X magnification which are (a) R3; (b) R5; (c) R7 (d) S; (e) M; (f) L	86

Figure 4.22	The FTIR band of NR/PAL nanocomposites of different surface treatment	88
Figure 4.23	The crosslink density of NR/PAL nanocomposites of different surface treatment	92
Figure 4.24	The rubber-filler interaction of NR/PAL nanocomposites of different surface treatment	93
Figure 4.25	The stress-strain curve of NR/PAL nanocomposites of different surface treatment	94
Figure 4.26	The tear strength of NR/PAL nanocomposites of different surface treatment	95
Figure 4.27	The hardness of NR/PAL nanocomposites of different surface treatment	96
Figure 4.28	The storage modulus of NR/PAL nanocomposites of different surface treatment	99
Figure 4.29	The loss modulus of NR/PAL nanocomposites of different surface treatment	99
Figure 4.30	The tan δ of NR/PAL nanocomposites of different surface treatment	100
Figure 4.31	The SEM micrograph of (a) UTD; (b) ODA; (c) DDA; (d) MQT at 1.0K X magnification	102
Figure 4.32	The TEM micrograph of (a) UTD; (b) ODA; (c) measured ODA	104

LIST OF ABBREVIATIONS

ABS Acrylonitrile butadiene styrene

ASTM American Society of Testing and Materials

ATP Attapulgite

BA Butyl acrylate

BKF 2,2'-Methylenebis(6-tert-butyl-4-methylphenol)

BPR Ball to Powder Ratio

BR Polybutadiene rubber

CaCl Calcium chloride

CEC Cation exchange capacity

CTAB Cetyltrimethylammonium chloride

CNT Carbon nanotube

CR Polychloroprene rubber

DDA (1-Dodecyl)trimethylammonium

DMA Dynamic mechanical analysis

DSC Differential scanning calorimetry

EDX Energy dispersive x-ray

ENR Epoxidized natural rubber

EPA Environmental Protection Agency

EVA Ethylene vinyl acetate

EPDM Ethylene-propylene-diene rubber

FESEM Field emission scanning electron microscopy

FTIR Fourier transform infrared spectoscopy

GPTMS 3-glycidoxypropyltrimethoxysilane

HA High ammonia

HCl Hydrochloric acid

HDA Hexadecylammonium

HNT Halloysite nanotube

KBr Potassium bromide

KOH Potassium hydroxide

LGM Lembaga Getah Malaysia

LS Layered silicates

MMT Montmorillonite

MQT 2-methacryloyoxy)ethyl trimethyl ammonium chloride

NBR Acrylonitrile-butadiene rubber

NR Natural rubber

OMMT Octadecylammonium montmorillonite

ODA Octadecylamine

PAL Palygorskite

PE Polyethylene

PEEK polyether ether ketone

PET Polyethylene terephthalate

PLLA Poly L-lactic acid

PMMA Polymethylmethacrylate

PVC Polyvinylchloride

PVP Polyvinylpyrrolidone

SBR Styrene butadiene rubber

SEM Scanning electron microscopy

SMR Standard Malaysian Rubber

TEM Transmission electron microscopy

TGA Thermogravimetric analysis

TMTD Tetramethyl thiuram disulfide

XNBR Carboxylated acrylonitrile-butidiene rubber

XRD X-ray diffraction

ZDEC Zinc diethyldithiocarbamate

ZnO Zinc oxide

LIST OF SYMBOLS

Rho/density ρ Phi φ θ Theta Tan delta $tan \ \delta$ Plus minus \pm Degree celcius $^{\circ}C$ Glass transition temperature T_{g} Na^{+} Sodium ion K^{+} Potassium ion h Hour wt % Weight percent Cure time t_{90} Scorch time ts_2 Minimum torque M_L Maximum torque M_{H} Torque difference ΔM Rubber-filler interaction $Q_f\!/Q_g$

Storage modulus

E'

E" Loss modulus

mm Milimeter

ml Mililiter

g Gram

MPa Megapascal

phr Part per hundred rubber

dNm Desinewtonmeter

N Newton

cm⁻¹ Per centimeter

kV Kilo volt

mA Miliampere

KESAN KAEDAH PENYEDIAAN PENGISI YANG BERBEZA KE ATAS SIFAT - SIFAT NANOKOMPOSIT GETAH ASLI / PALYGORSKITE

ABSTRAK

Tanah liat telah digunakan secara meluas sebagai pengisi untuk meningkatkan sifat-sifat dalam nanokomposit polimer pada pembebanan pengisi yang rendah. Bagaimanapun, penambahbaikan sifat-sifat bergantung sepenuhnya kepada keupayaan tanah liat untuk tersebar dalam matriks polimer. Dalam kajian ini, sifatsifat mekanikal, morfologikal dan fizikal nanokomposit yang mengandungi getah asli (GA) dan palygorskite (PAL) telah disediakan dengan menggabungkan kaedah pembekuan lateks dan pencampuran leburan telah dikaji. Dalam usaha untuk menyediakan pengisi nano dengan keserasian yang baik dengan matriks polimer, kaedah penyediaan pengisi yang berbeza telah digunakan (bebola pengisar, kaedah perrtukaran kation). Kesan parameter bebola pengisar yang berbeza pada nanokomposit GA/PAL dikaji untuk menghasilkan penyebaran PAL yang homogen. PAL juga dirawat dengan menggunakan kaedah pertukaran kation. Komposit telah dijalankan ujian ciri-ciri pematangan, sifat tegangan, kekuatan carikan, kekerasan, pembengkakan, spektroskopi fourier inframerah (FTIR), analisis pembelauan x-ray (XRD), mikroskop imbasan elektron (SEM), spektroskopi penghantaran elektron (TEM) dan analisis mekanikal dinamik (DMA). Didapati, nanokomposit GA/PAL dengan parameter bebola pengisar bernisbah 5:1 mengunakan bebola pengisar bersaiz besar (35 mm) dan dirawat dengan octadecylamine (ODA) mempunyai sifatsifat mekanikal, morfologikal dan fizikal yang lebih baik berbanding komposit yang tidak dirawat. Nanokomposit ini mempunyai sifat-sifat tegangan yang lebih tinggi, ketumpatan sambung silang yang tinggi dan interaksi getah-pengisi yang rendah.

EFFECT OF DIFFERENT FILLER PREPARATION METHOD ON PROPERTIES OF NATURAL RUBBER / PALYGORSKITE NANOCOMPOSITES

ABSTRACT

Recently, clay has been widely used as fillers to enhance properties in polymer nanocomposites at lower filler loading. However, the improvement of the properties depends on the ability of the clay to disperse in the polymer matrix. In this study, the mechanical, morphological and physical properties of nanocomposites containing natural rubber (NR) and palygorskite (PAL) prepared by combining latex coagulation and melt mixing method were investigated. In attempts to prepare nanofillers with enhance compatibility with polymer matrix, different types of filler preparation method were applied (ball milling and cation exchange). The effect of different ball milling parameters on NR/PAL nanocomposites was investigated to prepare homogenous PAL dispersion. PAL also was prepared by treated using cation exchange method. The composites were subjected to curing characteristics, tensile properties, tear strength, hardness, swelling, fourier transform infrared spectroscopy (FTIR), x-ray diffraction analysis (XRD), scanning electron microscopy (SEM), transmission electron spectroscopy (TEM) and dynamic mechanical analysis (DMA) test. It was found that NR/PAL nanocomposites with ball milling parameters of 5:1 ratio using large size ball mills (35 mm) and treated with octadecylamine (ODA) have improved mechanical, morphological and physical properties compared to untreated composites. These nanocomposites have higher tensile properties, high crosslink density and lower rubber-filler interaction.

CHAPTER ONE

INTRODUCTION

1.1 Introduction

Polymer-clay nanocomposites have attracted intense industrial and academic interests due to their remarkable and enhanced properties at low filler loading (Munusamy et al. 2009) compared to unfilled polymer or conventional composite materials. Major development in this field including applications to engineering polymers and investigation of the properties has been carried out over the last two decades. There are three classes of polymer which are thermoplastic, thermoset and elastomer.

Natural rubber (NR) is a kind of unique material, it is an elastomer. In uncure state, it is behaved as thermoplastic. Nevertheless, once it is vulcanised, it will become thermoset. Natural rubber is an essential elastomer with a wide range of applications (Zhang et al. 2010; Gu et al. 2010) as a result of reinforcement by fillers. It is found in the sap of a tree named *Hevea brasiliensis*. It is a natural biosynthesis polymer possessing excellent characteristics, such as high tensile strength, due to its ability to crystallise upon stretching (Wang & Chen 2013). The purpose of choosing NR as rubber matrix is to develop renewable rubber composites. NR latex is a natural polymer of isoprene, mostly *cis*-1,4-isoprene and its stereoregularity contributes to the strain-induced crystallization as well as good green and tensile strength. NR is reinforced with various types of fillers with different surface energy and particle size for practical applications in order to enhance its properties especially mechanical properties.

The most important fillers in rubber composites are carbon black and silica because of their features that improve the performance of rubber product (Peng et al. 2007). Recently, researchers have introduced and used nanoscale filler such as clay particles (Rezende et al. 2010) as an initiative to replace the use of carbon black. The improvement of properties depends heavily on the ability of clay to disperse in the polymer matrix, the interfacial interaction between the rubber matrix and nanoclay, the clay loading, and the modification of clay (Lopattananon et al. 2015; Naderi et al. 2014). Examples of clay particles include kaolin (Murray 2000), montmorillonite (Wang & Chen 2013), attapulgite (Galan 1996) and mica (Güven et al. 1992). Significant improvement in the physical properties of thermoplastics and thermosets achieved through the use of clay minerals as fillers in polymer matrix has been frequently reported. This is due to the characteristics of the polymer filled with clay exhibit better properties, such as, improved barrier and chemical resistance, better mechanical properties, improved heat resistance, improved thermal stability (Galimberti et al. 2015; Ramôa et al. 2015; Paul & Robeson 2008) and decreased flammability (Tan et al. 2012; Varghese et al. 2003).

Attapulgite often known as palygorskite, is a term for hydrated magnesium aluminium silicate mineral. The International Nomenclature Committee has specified palygorskite as the preferred name, but in trade circles the name attapulgite is well known and is also used by producers and consumers (Murray 2000). The structure of palygorskite contains ribbons, where the inversion of SiO₄ tetrahedra links each ribbon along a set of Si-O-Si bonds (Galan 1996). Based on the model of Bradley (1940),the ideal structural formula for palygorskite Si₈O₂₀(Mg₂Al₂)(OH)₂(OH₂)₄(H₂O)₄. Palygorskite is widely used in different industrial fields due to its sorptive and rheological properties. It is mostly used as an

animal waste adsorbent, pesticide carrier, decolorizing agent, in the oil refining and pharmaceutical industries, and as a catalyst and catalyst support (Barrios et al. 1995). The term fuller's earth is used to describe clays which have sorptive and bleaching qualities. Since palygorskite has these values, it is also known as fuller's earth.

The use of clay masterbatch has been introduced as a recent method to achieve intercalation or exfoliation of this filler in polymer composites. The masterbatch is usually prepared with a specific surfactant or low molar mass polymer, with structures compatible to that of polymeric material (Kaneko et al. 2010; Hasegawa et al. 2003). The basic idea is to increase the clay interlayer spacing by polymerization of monomers or suspension of clay in water/solvents or low molar mass polymers. Melt intercalation of high polymers is a powerful approach to produce layered silicate polymer nanocomposites. This method is quite general and is broadly applicable to a range of commodity polymers. It is harmless to the environment due to the absence of organic solvents and compatible with current industrial process. However, the dispersion of the clay in the polymer prepared by melt mixing is not good compared to the latex compounding (Kaneko et al. 2010). Rubber-clay nanocomposites prepared from latex by a coagulation method was reported to show an improvement in mechanical properties (Wang & Chen 2013). Tan (2012) has reported that clay layers can be dispersed homogenously and produce exfoliation by using a latex compounding method. Therefore, the advantages of melt mixing and latex compounding have led to the preparation of rubber composites by the combination of both methods through a masterbatch process in order to achieve the intercalation or exfoliation structure (Tan et al. 2012).

Palygorskite is naturally hydrophilic which makes it poorly suited to mixing and interacting to rubber matrix which is hydrophobic. Therefore, the clay must be treated to be able to give a better interaction with rubber matrix as composite made out of untreated clay could not give a good result due to poor interaction between the clay and the matrix. One way to modify or treat the PAL is by using ion exchange method. Cation exchange method is the process of ion exchange between PAL and alkyl-ammonium ions. Alkyl ammonium surfactants are generally used as organic surfactants. The reason for choosing cations exchange method instead of anion is because of the cations feature that is not strongly bound to the clay surface, so, small molecules cations can replace the cations present in the clay (Singla 2012). By exchanging ions present in between layers with organic cations, PAL clay can be compatibilized with NR matrix. In the meantime, this process enables the clay platelets to be separated so that they can be more easily intercalated and exfoliated.

Hence, in this study, the investigation was focused on analysing the properties of NR-PAL nanocomposites prepared by using the combination of latex compounding and melt mixing with the addition of ball milling method to improve the dispersion of PAL. Furthermore, the PAL was undergoing surface treatment using cation exchange method and the properties of treated PAL was compared to untreated PAL using similar method of compounding.