

**THE EFFECT OF AMINO FUNCTIONAL  
STARCH ON THE MECHANICAL AND  
DEGRADATION PROPERTIES OF  
CARBOXYLATED NITRILE BUTADIENE  
RUBBER LATEX FILMS**

**MUHAMMAD AFIQ MISMAN**

**UNIVERSITI SAINS MALAYSIA  
2017**

**THE EFFECT OF AMINO FUNCTIONAL STARCH ON THE  
MECHANICAL AND DEGRADATION PROPERTIES OF  
CARBOXYLATED NITRILE BUTADIENE RUBBER LATEX FILMS**

**by**

**MUHAMMAD AFIQ MISMAN**

**Thesis submitted in fulfillment of the  
requirements for degree of  
Doctor of Philosophy**

**May 2017**

## ACKNOWLEDGEMENTS

**In the name of Allah, the most gracious and the most merciful,**

*“Corruption has appeared on land and sea because of what the **Hands** of men have earned, that He (Allah) may make them **Taste** a part of that which they have done, in order that they may **Return**” – Ar-Rum (30 : 41).*

Based on the revelation from the holy Quran, we know that, all the corruption on earth is manmade. This is true when it comes into pollution and solid waste disposal. According to the World Bank data, if immediate actions are not taken, by the year 2025, 2.2 billion metric tons of solid waste are expected to be generated by the world’s major city. 10 % from this accumulation is contributed by plastics and rubber products; of which being dominated by the latex based products. As a result, global environmental instability occurred as a “**Taste**” for what we has done. To cater the problem, Allah S.W.T had specifically given an instruction in the quoted holy verse. The word “**Return**” hold the key to solve the problem and significantly led the journey to finish this research.

By referring to the Oxford dictionary, the word “Return” means; “go back to” or in the extended definition is; “go back to (a specific place or state or origin)”. As this revelation was applied in the context of solving the solid waste (specifically latex based products) disposal problem, a concept of reversion of the solid waste into its elemental compound is transpiring. In academics, this concept is known as the process of degradation/retro-gradation/regression. Thus, by degrading the complex

macromolecules of latex based products into its elemental cycle, it could help to restore the environmental problems created. As the latex based products usually end up in the landfill, a process of microbial facilitated degradation or biodegradation process could be manipulated where the by-products is expected to be a simple element or compound.

Thus, by holding the “Key” as instructed by the Almighty God, I finished this research as part of my duty as a believed servant to contribute a little, but significant knowledge as a legacy to our children and for a future betterment, as quoted from Mahatma Gandhi’s saying; “We did not inherit this world from our forefathers, but we borrow it from our children”. With that, I would like to thank my wise supervisor, Associate Professor Dr. Azura A. Rashid, and my co – supervisor, Dr. Zuratul Ain Abd Hamid, for giving a significant advice and assistance as well as trust for me to complete this research.

Special thanks to the government of Malaysia thru Ministry of Higher Education which had been supporting me financially by providing scholarship (MyPhD), to my father and mother, which continuously uplifting my passion, my wife and daughters which has been a loyal and supportive companion, family relatives, friends as well as the School’s Dean, Prof. Dr. Zuhailawati Hussain and skillful laboratory technicians of School of Materials and Mineral Resources Engineering (SMMRE) and Science and Engineering Research Center (SERC), Universiti Sains Malaysia.

*Alhamdulillah, JazakhAllah Hu Khairan Jazakh.*

*Muhammad Afiq Misman, 1 May 2017.*

## TABLE OF CONTENTS

	Page
<b>ACKNOWLEDGEMENTS</b>	ii
<b>TABLE OF CONTENTS</b>	iv
<b>LIST OF TABLES</b>	ix
<b>LIST OF FIGURES</b>	xi
<b>LIST OF SYMBOLS</b>	xvi
<b>LIST OF ABBREVIATIONS</b>	xviii
<b>ABSTRAK</b>	xix
<b>ABSTRACT</b>	xxi
<b>CHAPTER ONE : INTRODUCTION</b>	
1.1 Overview	1
1.2 Problem statement	6
1.4 Research objectives	9
1.5 Thesis outline	10
<b>CHAPTER TWO : LITERATURE REVIEW</b>	
2.1 Carboxylated nitrile butadiene rubber (XNBR) latex	12
2.1.1 XNBR crosslinking system	13
2.1.2 Zinc oxide as XNBR crosslinker	17
2.1.3 Latex film formation	18
2.1.4 Minimum film formation temperature (MFFT)	25
2.1.5 XNBR latex thin film properties	26
2.2 Sago starch	32

2.2.1	Physico-chemistry	33
2.3	Starch grafted acrylonitrile (ANS)	36
2.3.1	Synthesis of ANS	37
2.3.2	Properties of ANS	44
2.4	Rubber – Starch interaction	46
2.5	XNBR latex film degradation	48
2.6	ANS degradation	53

### **CHAPTER THREE : METHODOLOGY**

3.1	Materials	58
3.1.1	Carboxylated nitrile butadiene rubber latex	58
3.1.2	Sago starch	58
3.1.3	Other chemical	59
3.2	Purification of acrylonitrile monomer	60
3.3	Preparation of initiator reagent	61
3.4	Preparation of amino functional starch (ANS)	61
3.5	Characterization of ANS	62
3.5.1	Determination of degree of substitution (DS) by CHNS elemental analyzer	62
3.5.2	Fourier transform infrared (FTIR) analysis	64
3.5.3	Proton nuclear magnetic resonance ( <sup>1</sup> H-NMR)	65
3.5.4	Particle size and zeta potential analysis	66
3.5.5	The viscosity average molecular weight ( $M_v$ ) measurement	66
3.6	Compounding and films preparation process	68
3.7	Accelerated aging test	69

3.8	Biodegradation procedure and testing	70
3.9	Characterization of latex films	70
3.9.1	Tensile properties	70
3.9.2	Tear strength	71
3.9.3	Swelling test	72
3.9.4	Optical microscope analysis	73
3.9.5	Scanning electron microscopy (SEM)	73
3.9.6	Thermo-gravimetric analysis (TGA)	74
3.9.7	Water vapor transmission rates	74
3.9.8	Mass loss analysis	75
3.9.9	Calculation of Pearson's coefficient of correlation, r-value	76
3.10	Experimental chart	78

#### **CHAPTER FOUR : THE SYNTHESIS AND CHARACTERIZATION PROCESS OF AMINO FUNCTIONAL STARCH**

4.1	An overview	79
4.2	Determination of ANS grafting	80
4.2.1	Proton nuclear magnetic resonance ( $^1\text{H-NMR}$ )	80
4.2.2	Fourier transform infrared (FTIR) analysis	84
4.3	Determination of ANS degree of substitution	87
4.3.1	CHNS elemental analysis	87
4.3.2	$^1\text{H-NMR}$ spectrum analysis	91
4.3.3	Zeta potential analysis	92
4.4	Intrinsic viscosity and average viscosity molecular weight ( $M_v$ )	94
4.5	Zeta potential analysis of optimized ANS in XNBR latex medium	96

4.6	Summary	100
-----	---------	-----

## **CHAPTER FIVE : THE EFFECT OF ANS LOADINGS ON XNBR LATEX THIN FILMS MECHANICAL, OPTICAL, AND THERMAL PROPERTIES**

5.1	An overview	101
5.2	Crosslink density	102
5.3	Tensile strength	104
5.4	Proposed interaction theory of ANS-XNBR latex	108
5.5	Stress-strain curve	111
5.6	Elongation at break	113
5.7	Modulus of elongation	115
5.8	Tear strength	117
5.9	Morphological analysis	119
5.10	Thermogravimetric analysis–derivative thermal analysis (TGA–DTG)	122
5.11	Summary	124

## **CHAPTER SIX : THE DEGRADATION (AGING AND BIODEGRADATION) OF ANS/XNBR LATEX FILMS**

6.1	An overview	126
6.2	Thermal aging process of ANS/XNBR latex	127
6.2.1	Tensile properties	127
6.2.2	Elongation at break	129
6.2.3	Modulus of elongation	131
6.2.4	Tear strength	133
6.2.5	Crosslink density	134

6.2.6	Morphological analysis	136
6.3	Biodegradation process of ANS/XNBR latex films.	140
6.3.1	Fourier transform infrared (FTIR) analysis	140
6.3.2	Mass loss analysis	142
6.3.3	Water vapor transmission (WVT) analysis	144
6.3.4	Tensile strength	146
6.3.5	Elongation at break	149
6.3.6	Tear strength	151
6.3.7	Crosslink density	152
6.3.8	Morphological analysis	154
6.3.9	Thermo-gravimetric analysis – Differential thermal analysis (TGA – DTG)	162
6.4	Summary	166

## **CHAPTER SEVEN: CONCLUSION AND FUTURE WORKS**

7.1	Conclusions	168
7.2	Future works	170

<b>REFERENCES</b>	171
-------------------	-----

## **LIST OF PUBLICATIONS**

## **APPENDIX**

## LIST OF TABLES

		<b>Page</b>
Table 1.1	The consumption of rubber by type of products in Malaysia (in Tonnes).	2
Table 1.2	Malaysia output of selected rubber products.	3
Table 2.1	Physico chemical properties of different type of starches	32
Table 2.2	Typical composition of sago starch.	34
Table 3.1	Properties of XNBR latex as received from Synthomer (M) Sdn. Bhd.	58
Table 3.2	Properties of sago starch as received from Sago Link (M) Sdn. Bhd.	59
Table 3.3	The formulation for purification process of acrylonitrile monomer solution.	60
Table 3.4	The formulation for initiator reagent.	61
Table 3.5	The formulations for preparation of amino functional starch.	62
Table 3.6	The wavenumber for certain functional group.	65
Table 3.7	Compounding formulation for production of amino functional starch/carboxylated nitrile butadiene rubber (ANS/XNBR) latex film.	69
Table 3.8	The relationship characteristics for r - value obtained.	77
Table 4.1	The proton assignments in control starch anhydroglucose units.	80
Table 4.2	ANS intrinsic viscosity and viscosity average molecular weight, $M_v$ .	94
Table 4.3	The particle size of dispersing species.	98
Table 5.1	The thermal properties of CS, ANS, XNBR, and ANS/XNBR.	122
Table 6.1	Percentage of tensile strength retention for control and ANS/XNBR latex films over period of biodegradation.	147
Table 6.2	Percentage of elongation at break retention for control and ANS/XNBR latex films over period of biodegradation.	151

Table 6.3	Percentage of tearing strength retention for control and ANS/XNBR latex films over period of biodegradation.	152
Table 6.4	Percentage of crosslink density retention for control and ANS/XNBR latex films over period of biodegradation.	153

## LIST OF FIGURES

		<b>Page</b>
Figure 1.1	The production of natural and synthetic rubber over the past 15 years.	3
Figure 1.2	The consumption of natural and synthetic rubber in Malaysia over the past 15 years.	4
Figure 2.1	Schematic diagram of conventional XNBR latex crosslinking system where; (a) is the divalent metal oxide ionic bonding and (b) the (mono/poly-) sulfidic crosslinks.	14
Figure 2.2	The schematic representation of ionic (a) crosslinks, (b) multiplets, and (c) clusters in (d) ionomers.	15
Figure 2.3	The formation of biphasic structure of (a) Untreated XNBR-ZnO (b) ammonia treated XNBR-ZnO crosslinking system.	16
Figure 2.4	Schematic diagram of: (a) solvation of carboxylic acid with ammonia; (b) co-ordination of ammonia with zinc ion of carboxylic acid salt.	17
Figure 2.5	The schematic diagram for (a) the characteristics of two adjacent particles, and (b) the potential energy with respect to the distant of two particles.	20
Figure 2.6	The schematic diagram of the formation of electrokinetic charges on the particles slipping plane.	21
Figure 2.7	Particle interaction potential energy (a) formation of net energy (b) Suppression of electrostatic repulsion leading to diminishing of net energy.	22
Figure 2.8	Formation of covalent bond among latex colloidal particles.	23
Figure 2.9	Particle coalescence in latex colloidal particles.	24
Figure 2.10	Cementing mechanism of non rubber substances within rubber particles.	24
Figure 2.11	The tensile strength of XNBR latex films with different ZnO loading.	26
Figure 2.12	The elongation at break of XNBR latex films with different ZnO loading.	27
Figure 2.13	The relaxed modulus of XNBR latex films at different ZnO loading	28

Figure 2.14	The tear strength of XNBR latex films with different ZnO loading.	28
Figure 2.15	The crosslink density of XNBR latex films at different ZnO loading.	30
Figure 2.16	SEM micrograph of typical sago starch granules.	33
Figure 2.17	The flow diagram for extraction process of ANS by DMSO and $\gamma$ -butyrolactone.	41
Figure 2.18	General mechanism of enzymatic reaction on the polymer surfaces under aerobic condition.	56
Figure 2.19	Typical microorganism growth curve.	57
Figure 3.1	Schematic diagram of ubbelohde viscometer size 2B.	67
Figure 3.2	Schematic representation of latex films in soil.	70
Figure 3.3	Schematic representation of dumbbell shaped specimen according to ASTM D 412-06. All dimensions in millimeter (mm).	71
Figure 3.4	Schematic representation of crescent shaped specimen according to ASTM D 624-00. All dimensions in millimeter (mm), or otherwise stated.	72
Figure 3.5	Typical WVT diagram to obtain the inclination of the best straight line.	75
Figure 3.6	Overall research flow chart.	78
Figure 4.1	$^1\text{H}$ -NMR spectrum for control sago starch with their respective molecular structure and proton assignments.	81
Figure 4.2	Comparison of $^1\text{H}$ -NMR spectrum for (a) control, (b) ANS 0.5, (c) ANS 1.5, (d) ANS 3.0 and (e) ANS 5.0 with their respective molecular structure. Insert diagram is the molecular structure of (f) control starch (g) ANS with their respective proton assignments.	83
Figure 4.3	FTIR spectra for all samples with respect to the concentration of acrylonitrile used for ANS. A region marked as (a), (d), and (f) represents the starch fingerprint peaks whilst region marked with (b), (c), and (e) represents the acrylonitrile peaks.	85
Figure 4.4	Effect of acrylonitrile concentration on the degree of substitution of amino functional starch (ANS).	88

Figure 4.5	Comparison between particle size and effective degree of substitution with respect to concentration of acrylonitrile.	90
Figure 4.6	The degree of substitution obtained by elemental analysis (EA) and proton-nuclear magnetic resonance ( <sup>1</sup> H-NMR).	92
Figure 4.7	Comparison between specific degrees of substitution with particle zeta potential with respect to concentration of acrylonitrile.	93
Figure 4.8	The comparison of starch particle size reduction with their respective intrinsic viscosity with coefficient of correlation, $R^2 = 0.9183$ .	95
Figure 4.9	Distribution of apparent zeta potential for (a) ANS, (b) XNBR, and (c) ANS/XNBR.	99
Figure 5.1	Effect of ANS loadings on the ANS/XNBR latex crosslink density.	102
Figure 5.2	Effect of ANS loading on the tensile strength of ANS/XNBR latex film.	104
Figure 5.3	Proposed molecular interaction between (i) carboxylated nitrile butadiene rubber (XNBR) latex and (ii) aminofunctional starch; (a) an ionic bonding between zinc and oxide ions, (b) the covalent bonding of mono-/polysulfidic linkages, and (c) the proposed bipolar interaction between ANS and XNBR.	107
Figure 5.4	The role transition effect of dual phase ANS filler with respect to the filler loading. (a) Dispersion of the respective particles in the latex compound, (b) the arrangement of the dispersing particles during the film formation process, (c) particle inter-diffused and forming a honeycomb structure, and (d) continuous phase of cured latex films.	109
Figure 5.5	Stress-strain curve of control and ANS/XNBR latex films.	112
Figure 5.6	Effect of ANS filler loading on the XNBR latex film elongation at break.	113
Figure 5.7	Effect of ANS filler loadings on the ANS/XNBR latex film modulus of elongation.	116
Figure 5.8	Effect of ANS loading on the tear strength of ANS/XNBR latex film.	117

Figure 5.9	The fractured sample from tear test experiment. The number on the specimens is representing the ANS loading in the films.	118
Figure 5.10	Surface morphology of the control and ANS/XNBR latex films with respect to their filler loading. (a) Control films, (b) 5, (c) 10, (d) 15, (e) 20 phr ANS filler loading.	121
Figure 5.11	TGA-DTG thermograph for ANS, CS, ANS/XNBR, and XNBR.	124
Figure 6.1	Tensile strength of control, 72, and 96 hours of aging of ANS/XNBR latex film tensile strength with their respective percentage of retention.	128
Figure 6.2	Elongation at break of control, 72, and 96 hours of aging of ANS/XNBR latex films with their respective percentage of retention.	130
Figure 6.3	Modulus of elongation for (a) 100 %, (b) 300 %, and (c) 500 % elongation for ANS/XNBR latex films with 72 and 96 hours of aging process.	132
Figure 6.4	Tear strength of control, 72, and 96 hours of aging of ANS/XNBR latex films with their respective percentage of retention.	133
Figure 6.5	Crosslink density of control, 72, and 96 hours of aging of ANS/XNBR latex films with their respective percentage of retention.	136
Figure 6.6	Optical analysis of the aged (a) control XNBR and (b) ANS/XNBR 10 latex films at 72 and 96 hours of aging.	137
Figure 6.7	Morphological analysis of control XNBR latex films after 96 hours of aging test at magnification of (a) 200 x, (b) 5000 x and (c) 20 000 x.	138
Figure 6.8	Morphological analysis of ANS/XNBR 10 latex films at 96 hours of aging test at magnification of (a) 500 x, (b) 1000 x, and (c) micrograph of the void rich region.	139
Figure 6.9	FTIR spectra for ANS/XNBR 10 with respect to biodegradation duration.	142
Figure 6.10	Mass loss retention for control and ANS/XNBR latex films with different biodegradation duration.	143

Figure 6.11	Water vapor transmission (WVT) rates for control and ANS/XNBR latex films with different biodegradation duration.	145
Figure 6.12:	The tensile strength of control and ANS/XNBR latex films with different biodegradation duration.	146
Figure 6.13	The elongation at break for the control and ANS/XNBR latex films with different biodegradation duration.	150
Figure 6.14	The tear strength for control and ANS/XNBR latex films with different biodegradation duration.	151
Figure 6.15	The crosslink densities of control and ANS/XNBR latex films with different biodegradation duration.	153
Figure 6.16	The optical analysis of the appearance of control and ANS/XNBR latex films with respect to their filler loading and biodegradation weeks.	156
Figure 6.17	The morphology of biodegraded ANS/XNBR 10 latex films. (a) selected area of biodegraded ANS/XNBR latex films, (b – c) the remains of microorganism colony and, (d) a part of microorganism's hyphae and mycelia structure embedded in the ANS/XNBR latex films matrix (at magnification of 5000 x for (a), and 20 000 x for (b-d)).	157
Figure 6.18	Optical microscope images of (a) XNBR latex film (control), (b) ANS/XNBR latex films before the soil burial test, (c) selected starch crystal region on ANS/XNBR latex films for microorganisms growth observation, (d) microorganism colonies emerge at the surface of the second (2) weeks of biodegraded sample, (e) growth of microbial colonies on fourth (4) weeks, and (f) the microbial colonies at week 8 of biodegradation duration.	159
Figure 6.19	SEM images for infected region on ANS/XNBR latex film's surfaces at 8 weeks of biodegradation period (at magnification of (a) 1000 x, (b) 10 000 x, and (c) 30 000 x.	160
Figure 6.20	TGA-DTG thermograph for control XNBR latex films with different biodegradation duration.	164
Figure 6.21	TGA-DTG thermograph for control ANS/XNBR 10 latex films with a different biodegradation duration.	165

## LIST OF SYMBOLS

$n$	Number of designated carbon atom
mol	Mol
mol/AGU	Mol of substances with respect to anhydroglucose unit
$R_p$	Rates of polymerization
$K_n, n = 1, 11, 12, 2, 22$	Polymerization constant
$k_n, n = 1, 11, 12, 2, 22, 4, 5$	Polymerization constant
$R_i$	Rates of initiation
$k_p$	Polymerization constant
pH	Acidity/alkalinity constant
$\alpha$	Confidence level
$\mu\text{m}$	Micrometer
$\%.\mu\text{m}^{-2}$	Percentage over area
mV	milivolts
$R^2$	Coefficient of correlation
$\text{cm}^{-1}$	Per centimeter
%T	Percentage of transmission
$M_v$	Viscosity average molecular weight
dL/g	Deciliter over grams
K	Kelvin
$V_s$	Particle's settling velocity
G	Gravitational constant
$\rho_s$	Density of the particles
$\rho_l$	Density of the fluid

$d_p$	Particle radius in equivalent sphere
MPa	Mega Pascal
N/mm	Newton per millimeter
$T_{max}$	Maximum degradation temperature
°C	Degree Celsius
$\Delta T$	Temperature difference
$T_o$	Onset temperature
$pN\ nm^{-1}$	Pico Newton per nanometer
nm	Nanometer
M100	Modulus at 100 % elongation
M300	Modulus at 300 % elongation
M500	Modulus at 500 % elongation
$molcm^{-3}$	Mol per centimeter cubic

## LIST OF ABBREVIATIONS

AGU	Anhydroglucose unit
ANS	Amino functional starch
ASTM	American standard for testing materials
CS	Control starch
DMSO	Dimethyl sulfoxide
DMSO-d <sub>6</sub>	Dimethyl sulfoxide deuterated
DS	Degree of substitution
DS/SA	Specific degree of substitution
DTG	Derivative thermal analysis
EA	Elemental analysis
EB	Elongation at break
FTIR	Fourier transform infra-red
<sup>1</sup> H-NMR	Proton nuclear magnetic resonance
MFFT	Minimum film formation temperature
phr	Part per hundred rubber
ppm	Part per million
TGA	Thermo-gravimetric analysis
XNBR	Carboxylated nitrile butadiene rubber

**KESAN PENAMBAHAN KANJI BERFUNGSI AMINO KE ATAS SIFAT-  
SIFAT MEKANIKAL DAN DEGRADASI FILEM LATEKS GETAH NITRIL  
BUTADIENA TERKARBOKSIL**

**ABSTRAK**

Lateks getah nitril butadiena terkarboksil (XNBR) telah dicampurkan dengan kanji berfungsi amino (ANS) dan filem getah nipis telah dihasilkan melalui teknik pencelupan berkoagulan. Sebelum proses tersebut, ANS telah disintesis dengan kepekatan akrilonitril yang berbeza untuk mengkaji kesan kepekatan akrilonitril ke atas tahap penggantian (DS) dan kestabilan partikel ANS di dalam lateks XNBR. ANS yang dihasilkan dicirikan dengan menganalisa keputusan nuklear proton resonans magnetik ( $^1\text{H-NMR}$ ), analisis unsur (EA), dan infra merah terubah fourier (FTIR). Kelikatan produk dan berat molekul berdasarkan purata kelikatan ditentukan dengan viskometer 'Ubbelohde' dan kestabilan partikel ditentukan melalui analisa potensi zeta. Kajian mengenai kesan jumlah pembebanan ANS (0, 5, 10, 15, 20 phr) ke atas sifat - sifat mekanikal, morfologikal, termal, proses pecutan penuaan, dan biodegradasi filem lateks ANS/XNBR telah dijalankan. Keputusan kajian menunjukkan DS ANS bergantung kepada kepekatan akrilonitril akan tetapi kestabilan ANS bergantung kepada saiz partikel dan 'zeta potential'. Sifat akhir filem bergantung kepada kepekatan ANS. Penambahan ANS tidak meningkatkan sifat mekanikal filem lateks ANS/XNBR tetapi memberikan kesan ketara keatas sifat biodegradasi filem. Kajian morfologi menunjukkan penggunaan ANS di dalam sebatian lateks XNBR menjadikan permukaan filem lateks kasar dengan filem yang mempunyai jumlah ANS tertinggi menunjukkan pembentukan beberapa rongga dan gumpalan pengisi. Analisa TGA-DTG menunjukkan bahawa penambahan ANS

meningkatkan rintangan filem lateks terhadap degradasi terma. Walaubagaimanapun, ujian pecutan penuaan (isoterma) pada 100 °C menunjukkan bahawa penambahan ANS meningkatkan kadar penuaan lateks filem. Pembentukan rongga ini mempercepatkan serangan mikroorganisma ketika proses biodegradasi sekaligus meningkatkan kadar biodegradasi filem. Akan tetapi, jumlah pembebanan optimum untuk filem lateks ANS/XNBR dicapai pada 10 phr dengan merujuk kepada ketinggian sifat - sifat mekanikal filem yang dipamerkan berbanding filem ANS/XNBR pada pembebanan yang lain.

**THE EFFECT OF AMINO FUNCTIONAL STARCH ON THE  
MECHANICAL AND DEGRADATION PROPERTIES OF  
CARBOXYLATED NITRILE BUTADIENE RUBBER LATEX FILMS**

**ABSTRACT**

Carboxylated nitrile butadiene rubber (XNBR) latex was compounded with amino functional starch (ANS) and latex thin films were prepared by coagulant dipping technique. Prior to the process, ANS was synthesized with different acrylonitrile concentration to study the effect of acrylonitrile concentration on the degree of substitution (DS) and the stability of the ANS particle in XNBR latex. The ANS produced was characterized by proton nuclear magnetic resonance (<sup>1</sup>H-NMR), elemental analysis (EA), Fourier transform infrared (FTIR) analyses. The product viscosity and viscosity average molecular weight were determined by ubbelohde viscometer and the stability of the particle was obtained via zeta potential analysis. A study was conducted to determine the effect of ANS loadings (0, 5,10,15,20 phr) on the films mechanical, morphological, thermal, accelerated aging process, and biodegradation properties of ANS/XNBR latex films. Results show that, the DS of ANS depends on the acrylonitrile content, but the stability of ANS depends on the particle size and its zeta potential. The final properties of the films depend on the ANS concentration. The addition of ANS did not improve the mechanical properties of ANS/XNBR latex films, but give a significant contribution towards films biodegradability. Morphological study shown the incorporation of ANS in XNBR latex compound results in surface roughness with higher loading films shown a number of void and agglomeration of filler. TGA-DTG analysis shows that the addition of ANS increased the latex film resistance towards thermal degradation.

However, accelerated ageing test (isothermal) at 100 °C indicates that introduction of ANS increased the ability of the film aged. The formation of void in higher loading films accelerates the microbial attack during biodegradation test and contribute significantly towards films biodegradability. However, optimum filler loading for ANS/XNBR latex obtained was 10 phr due to higher mechanical properties showed by the films compared to other loading of ANS/XNBR latex films.

## CHAPTER ONE

### INTRODUCTION

#### 1.1. Overview

Elastomeric materials are one of the highest utilized products in the world. The materials offer several advantages such as; the ability to be stretched with low hysteresis and high tensile strength with low modulus. Out of total elastomer based products produced, 82 % of the products are latex based products. **Table 1.1** shown the total consumption of rubber products in Malaysia for 16 consecutive years (MRB 2016). Based on **Table 1.2**, in year 2014, latex thin film product (gloves and condoms) made up to almost 43 billion pieces or 73 % of the total rubber products consumed.

Latex thin film products usually are made by a coagulant dipping process. One of the most manufactured products via this method are rubber glove. Latex gloves vary from the medical to industrial type and the products was made distinctively in their physical properties, appearance, patterns, as well as the materials to meet the needs of the end-user.

However, latex gloves usually being distinguished either as natural or synthetic based gloves. Synthetic rubber glove industries started to flourish at the end of the 19<sup>th</sup> century, when rubber industries started to produce synthetic latexes due to the shortage of natural rubber supply and increased in demand for non-allergy latex gloves. Since then, the industry continue to expand in the 20<sup>th</sup> century with an