ULTRAFILTRATION TREATMENT FOR SPENT TUNGSTEN SLURRY GENERATED BY CHEMICAL POLISHING PROCESS IN WAFER FABRICATION INDUSTRY FOR REUSE

NUR FATIN AMALINA BINTI MUHAMMAD SANUSI

UNIVERSITI SAINS MALAYSIA 2017

ULTRAFILTRATION TREATMENT FOR SPENT TUNGSTEN SLURRY GENERATED BY CHEMICAL POLISHING PROCESS IN WAFER FABRICATION INDUSTRY FOR REUSE

by

NUR FATIN AMALINA BINTI MUHAMMAD SANUSI

Thesis submitted in fulfillment of the requirement for the degree of Master of Science

May 2017

ACKNOWLEDGEMENTS

In the name of Allah the Most Gracious, the Most Merciful.

First and foremost, I would like to thank Allah for His blessings because of Him I was able to finish this thesis without facing any obstacles, hereby completing my Master's studies. I deeply thank Him for the gifts, health and all life opportunities.

I would like to express the deepest appreciation to my supervisor, Professor Dr. Ahmad Zuhairi Abdullah, who continually conveyed a spirit of adventure in regards and without his guidance and persistent help, this thesis and research work would not have been possible. I am also grateful to Associate Professor Dr. Ooi Boon Seng for his support and guiding me throughout the research. I also give my respectful gratitude to my project leader, Mr. Mohmad Sabirin, teammates for their help, supervision throughout this project. I have learned a lot throughout this research period with many challenging yet valuable experiences in order to complete my master's studies. I am also grateful for the opportunities in obtaining the funding through (CREST). This funding does have help me so much covering my research project and my living expenses. I would like to express my eternal appreciation towards my parents (Muhammad Sanusi & Wan Nor Azzah), husband (Muhd Hazim), lovely daughter (Hana Safiya) and family, who have always been there me no matter where I am, for all unconditional supports and patience. Thank you so much for being ever so understanding and supportive. For husband, thank you for being around and for never ending motivation I've been getting all this while.

TABLE OF CONTENT

Page

ACKNOWLEDGEMENTS	ii
TABLE OF CONTENTS	iii
LIST FOR TABLES	viii
LIST FOR FIGURES	ix
LIST FOR SYMBOLS	xii
LIST FOR ABREVIATIONS	xiii
ABSTRAK	xvi
ABSTRACT	xvii
CHAPTER ONE: INTRODUCTION	
1.1 Research background	1

1.1	Research background	1
1.2	Problem statement	4
1.3	Research Objectives	6
1.4	Thesis scopes	7
1.5	Organization of the thesis	8

CHAPTER TWO: LITERATURE REVIEW

2.1	2.1 Chemical Mechanical Polishing Technology		10
	2.1.1	Chemical Mechanical Polishing Operation	12
	2.1.2	CMP Slurry	16
	2.1.3	Characterization of CMP Slurry	20
	2.1.4	Electrochemistry of CMP Colloidal Slurry	22
2.2	Membi	rane Science and Technology	26
	2.2.1	Ultrafiltration Technology	27

	2.2.2 Crossflow Ultrafiltration	29
2.3	Theory of Fouling dynamics	34
	2.3.1 Characterization of membrane fouling	34
	2.3.2 Factors of Membrane Fouling	39
	2.3.3 Prevention of fouling	44
2.4	Critical Flux	45
2.5	Concluding remarks	47
СНА	PTER THREE: METHODOLOGY	
3.1	Introduction	48
3.2	Chemicals and Materials	48
3.3	Experimental Process Flow	49
3.4	Characterization of Chemicals and Physicals Properties of Spent Tungsten Slurry	52
	3.4.1 Particle Size Distribution and Zeta Potential	52
	3.4.2 Turbidity	53
	3.4.3 pH and Conductivity	53
	3.4.4 Scanning Electron Microscope (SEM) and Energy X-ray spectroscopy (EDX)	53
	3.4.5 Transmission electron microscopy (TEM)	54
	3.4.6 Atomic force microscopy (AFM)	54
3.5	Pure water flux for UF membranes	55
3.6	Experimental Rig Set Up 55	
3.7	Transmembrane Pressure (TMP) and Flux 58	
3.8	Membrane Materials and Molecular weight cut off (MWCO) of membrane	59
3.9	Difference Flow Rate in Filtration System	60
3.10	pH Adjustment for Ionic Separation in UF	60

3.11	Proses Evaluation	61
	3.11.1 Transmembrane Pressure (TMP)	61
	3.11.2 Permeate Flux	61
	3.11.3 Rejection or Retention rate	62
	3.11.4 Yield	62
	3.11.5 Purity	62
	3.11.6 Selectivity	63
CHA	PTER FOUR: RESULTS AND DISCUSSION	
4.1	Analysis of pretreated Spent Tungsten Slurry	64
	4.1.1 Characteristic of pretreated Spent Tungsten Surry	64
	4.1.1 (a) pH and Zeta Potential Analysis of Spent Tungsten Slurry	66
	4.1.2 Characteristics of slurry at different sampling points in the plant	67
	4.1.2 (a) pH analysis for Point 1 and Point 4 of Spent tungsten slurry	67
	4.1.2 (b) Metal Elements Analysis for Point 1 and Pont 4 of Spent Tungsten Slurry	68
	4.1.3 Transmission Electron Microscopy (TEM) Analysis	69
4.2	Flux Analysis Study	71
	4.2.1 Pure Water Flux for UF Membrane	71
	4.2.2 Membrane Fouling of Spent Tungsten Slurry	73
	4.2.3 Flux over time for spent tungsten slurry versus pure water	75
	4.2.4 Critical pressure for different membrane materials	77
	and MWCO	
	4.2.5 Study on the Membrane Material of Spent Tungsten Slurry	78
	4.2.5 (a) Analysis of Metal Elements using Different	78

Membrane Material

		4.2.5 (b)	Characteristic Analysis of Spent Tungsten Slurry, Retentate and Permeate	81
		4.2.5 (c)	Mean Size and Zeta Potential Analyses for the Spent Tungsten Slurry and the Retentate for Different Membrane Materials and Different of MWCO	83
		4.2.5 (d)	Analysis of Separation Performances for the Membrane Material	84
		4.2.5 (e)	The Selectivity of the Membrane Materials for Filtration of Spent Tungsten Slurry	86
		4.2.5 (f)	Membrane Morphology	88
		4.2.5 (g)	SEM & EDX Inspection on Membrane Surface	90
4.3	Study of	on the Me	mbrane Fouling of Spent Tungsten Slurry	92
	4.3.1	Effect of	Transmembrane Pressure (TMP) of	92
		Membra	ne Fouling	
	4.3.2	Effect of	Different flow rate on Membrane Fouling	94
4.4	Cyclic	Filtration	Process in Treating Spent Tungsten Slurry	96
	4.4.1	Analysis	of metal elements of spent tungsten slurry with	96
		cyclic fil	tration process	
	4.4.2	The sele	ctivity of Tungsten and Iron metals for cyclic	98
		filtration	L	
	4.4.3	The yield	d and purity of silica particles for cyclic filtration	99
4.5	pH adj	ustment of	f spent tungsten slurry	101
	4.5.1	The mea spent tu	n size distribution and zeta potential for the ingsten slurry at pH 2 and pH 9	101
	4.5.2	Analysis at pH 2 a	of Silica and Tungsten in Spent Tungsten Slurry and pH 9	103
	4.5.3	Membra pH 2 and	ne Fouling by Spent Tungsten Slurry at 1 pH 9	104

CHAPTER FIVE: CONCLUSION AND RECOMMENDATION

5.1	Conclusions	108
5.2	Recommendations	111
REF	ERENCES	112
APPI	ENDIXES	
Appe	ndixe A Flux Calculation	

- Appendixe B Retention Calculation
- Appendixe C Selectivity Calculation

Appendixe D Experimental Flux Value

LIST OF TABLES

		Page
Table 2.1	Performance of various types of slurry and oxidant in CMP process	18
Table 2.2	Table of various type membrane treatment of spent tungsten slurry	31
Table 2.3	Table of common operating condition for CMP filtration treatment using membrane process	32
Table 3.1	List of Chemicals used in the experiments	49
Table 3.2	List of membrane materials used in the experiment	49
Table 4.1	Characteristic of Raw Slurry, Diluted Raw Slurry and Spent Tungsten Slurry	64
Table 4.2	Characteristic of Spent Tungsten Slurry, Retentate and Permeate of membrane PS 50 kDa	82

LIST OF FIGURES

Figure 1.1	Schematic of CMP process	1
Figure 2.1	Schematic diagram of an IC produced by using CMP process	11
Figure 2.2	A principle of tungsten removal in CMP process	13
Figure 2.3	Basic scheme for a typical CMP system	14
Figure 2.4	Diagram showing the stern model of the electrochemical double layer	25
Figure 2.5	Picture of the fouling mechanism on the membrane surface	36
Figure 2.6	A schematic presentation of the three stages in flux decline (I) initial rapid drop from the flux of filtration; (II) long- term gradual flux decline; and (III) time dependent steady state flux	37
Figure 3.1	Process flow diagram of overall experimental works involved	50
Figure 3.2	A schematic diagram of the sampling point for collecting spent tungsten slurry sample	52
Figure 3.3	A schematic diagram of UF membrane testing rig for spent tungsten slurry sample treatment	56
Figure 3.4	Actual experimental rig for the ultrafiltration process	57
Figure 3.5	Schematic diagram of the membrane holder module	58
Figure 4.1	pH and Zeta Potential for Spent Tungsten Slurry	66
Figure 4.2	pH between Point 1 and Point 4 of Spent Tungsten Slurry	67
Figure 4.3	Concentration of metal elements in samples from point 1 and point 4	68
Figure 4.4	TEM image of spent tungsten slurry particle morphology at Point 1	70
Figure 4.5	Initial water flux, J_0 , over time at different MWCO for PS, PES and PVDF membrane	72
Figure 4.6	Flux versus time when the system was operated at 1 bar and	74

flow rate of 0.4L/min by using different membrane materials and different MWCO

- Figure 4.7 Flux over time of spent tungsten slurry and pure water 76 operated at 1 bar and flux flow rate of 0.4L/min by using PS membrane MWCO 50KDa
- Figure 4.8 Critical pressure for different membrane materials and 77 different MWCO operated flux flow rate 0.4L/min
- Figure 4.9 Analysis of metal elements using different membrane 80 materials
- Figure 4.10 Analysis of Mean size distribution and zeta potential for 83 retentate of all membrane material and MWCO
- Figure 4.11 Separation performances of all membrane material in 85 removing tungsten and silica
- Figure 4.12 The selectivity of the membrane material in the filtration of 87 spent tungsten slurry
- Figure 4.13 AFM images for the top view of (a) Fresh 50 PS 89 membrane, (b) Used 50 PS membrane, (c) Fresh 50 PVDF membrane,(d) Used 50 PVDF membrane
- Figure 4.14 Membrane surface morphology at a magnification of 5K 90
- Figure 4.15 Elemental analysis of the membrane surface 91
- Figure 4.16 Flux over time for PS membrane at TMP of 0.5, 1 and 1.5 93 bar at flow rate 0.4 L/min
- Figure 4.17 Flux over time for PS membrane at different flow rates 95 (0.2, 0.4, 0.6 L/min) operated at 1 bar
- Figure 4.18 Concentrations of silica and tungsten in the cyclic filtration 97
- Figure 4.19 The selectivity of tungsten and iron in cyclic filtration 98
- Figure 4.20 Yield and purity of silica colloidal particles in spent 100 tungsten slurry for three cycles
- Figure 4.21 The mean size distribution and zeta potential for the spent 102 tungsten slurry of pH 2 and pH 9
- Figure 4.22 Metal analysis of silica and tungsten in spent tungsten 104 slurry at pH 2 and pH 9
- Figure 4.23 Flux over time for spent tungsten slurry at pH 2 and pH 9 105

LIST OF SYMBOLS

Μ	Molar
Wt	Weight
n.a	Not available
Kda	Kilo Dalton
t	Time
ζ	Zeta Potential
P _F	Feed pressure
P _R	Retentate pressure
P _P	Permeate pressure
R	Retention rate
C _P	Solute concentration permeate
C _R	Solute concentration retentate
C _f	Solute concentration feed
J_w	Membrane pure water flux
t _{ss}	time of steady state

LIST OF ABREVIATIONS

СМР	Chemical mechanical polishing
W	Tungsten
Fe	Iron
Cu	Copper
Al	Aluminium
Na	Natrium
Ti	Titanium
TiN	Titanium Nitride
Та	Tantalum
TaN	Tantalum Nitride
ILD	Inter-level dielectric
STI	Shallow trench isolation
IC	Integrated circuit
W-CMP	Tungsten chemical mechanical polishing
IEP	Isoelectric pH
μm	Micro meter
WO ₃	Tungsten trioxide
WO ₂	Tungsten dioxide
SiO ₂	Silica
$H_{2}O_{2}$	Hydrogen peroxide
Fe(NO ₃)	Ferric nitride
KI0 ₃	Potassium iodate
AI ₂ 0 ₃	Aluminium oxide

K_3 Fe(CN) ₆	Potassium ferricyanide
K_2SiO_3	Potassium silicate
H_2SO_4	Sulfuric acid
<i>K</i> ⁺	Potassium ion
H ⁺	Proton
<i>H</i> ₂ <i>O</i>	Oxygen
SO_{4}^{-2}	Sulfate
SiOH	Silica silanol
EDL	Electrical double layer
DLS	Dynamic light scattering
OH-	Hydroxide
UF	Ultrafiltration
PS	Polysulfone
MF	Microfiltration
НСІ	Hydrochloric
Nm	Nanometer
MPa	Mega Pascal
ТМР	Trans-membrane pressure
PES	Polyethersulfone
MWCO	Molecular weight cut off
HNO ₃	Nitric acid
NaOH	Sodium Hydroxide
C ₂ H ₅ OH	Ethanol
PVDF	Polyvinilydinefluoride
NTU	Nephelometric Turbidity Units

SEM	Scanning Electron Microscope
EDX	Energy Dispersive X-ray Spectroscopy
TEM	Transmission Electron Microscopy
Ra	Mean roughness
V	Volume of permeate water
А	Area of the flat sheet membrane
Р	Pressure
Psi	Pounds per square inch
DI	Distilled water
H ₂ 0	Water
ICP- OES	Inductively coupled plasma optical emission spectrometry
AFM	Atomic Force Microscopy

ULTRAPENAPISAN BAGI BUBURAN TUNGSTEN TERPAKAI DARIPADA PROSES PENGILATAN KIMIA TUNGSTEN DALAM INDUSTRI PEMBUATAN WAFER UNTUK DIGUNA SEMULA

ABSTRAK

Proses pengilatan kimia diaplikasi secara meluas terutamanya dalam industri mikro elektronik bertujuan untuk meratakan permukaan wafer. Proses ini melibatkan mengilatkan permukaan bersalut logam oleh proses kimia dan diikuti pembuangan lapisan logam yang telah terjejas oleh proses mekanikal untuk mencapai pengilatan yang sempurna menggunakan buburan tungsten. Ia terdiri daripada campuran komponen yang kompleks di mana sifat fizikal dan kimianya boleh berubah bergantung kepada jenis dan keadaan rawatan yang diterima. Oleh itu, pencirian kimia dan fizikal buburan tungsten terpakai merupakan maklumat penting dalam usaha mempertimbangkan kaedah pemulihan yang sesuai. Penggunaan semula butiran silika bersaiz nano diperkenalkan sebagai salah satu langkah untuk mengurangkan kos pembuatan dan jumlah pengeluaran air sisa di mana ia memberi kebaikan dari segi kos dan alam sekitar. Sistem ultrapenapisan aliran silang telah dipilih sebagai kaedah untuk merawat dan kitar semula buburan tungsten terpakai hasil daripada proses pengilatan kimia. Ujikaji membran ultrapenapisan telah dijalankan menggunakan terhadap tiga jenis membran yang berbeza (polysulfone, polyethersulfone and polyvinylidene fluoride) dengan MWCO yg berbeza iaitu 10, 30, 50 dan 100KDa. Kesan kesan tekanan dalaman (TMP) membran dan kadar aliran fluks telah juga diuji. Analisis terhadap pemilihan, pengekalan butiran silica dan pengubahsuaian nilai pH pada buburan tungsten terpakai telah dijalankan untuk menentukkan prestasi kecekapan membran penapisan dan fenomena pengotoran. Tekanan membran 1 bar memberikan kadar terbaik dan tahap kotor yang paling rendah berbanding tekanan dalaman yang lain. Keputusan terbaik ditunjukkan oleh membran PS 50KDa yang mempamerkan pencapaian 92% dalam pengekalan butiran silika dan cuma 42% dalam pengekalan tungsten. Ia juga mencapai purata saiz paling rendah di mana perbezaan purata saiz hanya pada tahap 0.5% berbanding purata saiz asal butiran silika isitu 125 nm dalam retentate berbanding membran PVDF dengan MWCO yang sama. Pemulihan buburan tungsten terpakai boleh ditingkatkan dengan pengubahsuaian nilai pH 9 telah mencapai prestasi yang baik kerana mempunyai nilai saiz purata 126 nm berbanding nilai saiz purata asal 125 nm dan memberi pengekalan butiran silika terbesar sebanyak 42% dalam retentate. Ia juga mencapai nilai negatif terbesar -40 mV dalam nilai zeta potensi yang menunjukkan kestabilannya. Secara kesimpulan, membran PS 50KDa menunjukkan potensi terbesar dalam penapisan dan pemekatan butiran silika hasil proses pengilatan kimia dengan keadaan nilai pH 9.

ULTRAFILTRATION TREATMENT FOR SPENT TUNGSTEN SLURRY GENERATED BY CHEMICAL POLISHING PROCESS IN WAFER FABRICATION INDUSTRY FOR REUSE

ABSTRACT

Chemical Mechanical Polishing (CMP) is a widely used process to planarize wafers for microelectronic applications. It involves the polishing of metallic surface by chemical action followed by the removal of the modified layer by mechanical action using tungsten slurry. It is a rather complex mixture that both physical and chemical properties of the components in the slurry are expected to change depending on the condition and type of treatments that they receive. As such, characterization of the spent tungsten slurry and knowledge on its physical and chemical properties are critical in order to consider suitable recovery methods. Recycling of the abrasive slurry is one of the options to reduce the manufacturing cost and to achieve environmental benefits that arise from the reduction of wastewater volume. Crossflow ultrafiltration system was used as a method to recycle the silica based slurry. It investigated using three membrane materials i.e. polysulfone (PS), polyethersulfone (PES) and polyvinylidene fluoride (PVDF) with different molecular weight cut off (MWCO) of 10, 30, 50 and 100KDa. Effects of transmembrane pressure (TMP) on flux flow rate were characterized. Analyses of selectivity, retention and pH adjustment of feed water were done to demonstrate the membrane performance and fouling phenomena. A TMP of 1 bar gave the lowest fouling result as compared to the other TMPs. The PS 50KDa membrane demonstrated the best results with 92% retention of silica particles and only 42% retention of tungsten. It also achieved the lowest mean size particle of 125nm in the

retentate or only 0.5% value difference compared to that of the original spent tungsten slurry than those of other membranes especially PVDF membrane with the same MWCO. The performance in spent tungsten slurry recovery could be improved by adjusting the pH to 9 that gave the best performance in terms of having the lowest mean size of 126 nm which was close to 125 nm of the original size of particles in the spent tungsten slurry. It gave the highest retention of silica particles of 42% in the retentate. It also had the largest negative value of -40mV in zeta potential to suggest its stability. In conclusion, PS ultrafiltration with 50KDa of MWCO membrane showed the highest potential in filtrating and concentrating the spent tungsten slurry of CMP process with the best result achieved at a pH value of 9.

CHAPTER ONE

INTRODUCTION

1.1 Research Background

Chemical mechanical polishing (CMP) is a process of polishing the device side of a semiconductor wafer by applying a chemical action followed by applied mechanical action to eliminate the modified layer. The primary fundamentals of CMP is to ensure the metallic surface of the wafer is polished with a pad together by adding abrasive slurry to remove the excess of metal deposited on the wafer surface, with the purpose of obtaining the required planarization of wafer surface (Xiao, 2001). The CMP process is depicted using the scheme as presented in Figure 1.1.



Figure 1.1 Scheme of CMP process (Stojadinovic et al., 2016)

Throughout the manufacturing process, a thick conductive layer is deposited on the surface wafer with the purpose to fill the vias and trenches. Tungsten is one of the common deposit layers with low resistivity for multilevel interconnection structures of wafers (Kang *et al.*, 2010). The oxidation of wafer surface on the metallic upper layer along with chemical reactants such as mixing of oxidizer, surfactant and catalyst and the abrasive particle of slurry as function of mechanical action will remove the excess tungsten on the wafer surface (Stein, 2004).

Slurry consists of abrasive particles such as silica, alumina, ceria and also other aqueous medium that assists the abrasive slurry suspension, which acts to combine the chemical and mechanical polishing process (Luo and Dornfield, 2001). The slurry is acidic, which has to combine with appropriate oxidizing agent for surface layer polishing by metal passivation and then by the metallic film dissolve process (Wang *et al.*, 2012). Tungsten slurry is designed to polish the conductive layers on the wafer surface, which is important to cover the vias on the surface of inter-level dielectric ILD. One of the important roles in CMP process is the chemical composition of the abrasive particle slurry, especially when mixing with chemical reactants such as the oxidizer (Stein *et al.* 2004). Commonly, the tungsten CMP slurries consist of silica or alumina which is suspended, in the aqueous solution of oxidizing agents (Kang *et al.*, 2010). Tungsten slurry usually contains metal contaminants in the waste slurry to cause some difficulty for their removal in order to recycle the slurry. Hence, developing a method for the removal of metal contaminants in the used slurry is deemed necessary.

Testa *et al.*, (2014) attempted the method to recycle the acidic-based silica for the polishing tungsten CMP. Various studies have been focused on recycling and regeneration of CMP slurry using filtration (Ndieye *et al.*, 2004; Singh and Song, 2007; Coetsier *et al.*, 2011; Testa *et al.*, 2014). After the polishing CMP process, the properties of the abrasive slurry are often disrupted and usually a small portion of the

slurry is degraded. Besides, during rinsing step of the polishing CMP process, all the chemical components and abrasive slurry particles are highly diluted (Ndieye *et al.*, 2004). The reduction of wastewater through filtration and recycling of the used slurry after polishing are among the common goals of research in semiconductor industry.

The most suitable method for recycling the used CMP slurry is the ultrafiltration processes, in view of the abrasive particle size of slurry (Singh and Song, 2007; Coetsier et al., 2011; Testa et al., 2014). Besides the economic interest of recycling the used slurry, a potential reduction the amount of wastewater generated from the CMP process using ultrafiltration, can promise clear benefits to the environment. The initial step in recycling of the used slurry is to concentrate the slurry with the purpose of recovering the abrasive silica particles (Kurisawa, 2001). Then, readjustment of the concentrated used slurry should be done by injection of chemical components (Testa et al., 2014). Through ultrafiltration processes the slurry chemical constituents are not reinstated, so, the regeneration of abrasive slurry particles should be done before recycling back into the CMP process (Testa et al., 2014). The chemical composition of the abrasive particles in the slurry acts as a main role in controlling CMP parameter, uniformity and defective value of wafer surface in the CMP process. Because of this major requirement, there is a need for chemical adjustment for the concentrated spent tungsten slurry (Coetsier et al., 2011). This is important so that the recovered slurry can have specific chemical properties and necessary stability.

On the other hand, ultrafiltration is the critical step in the separation of liquid and solid phases of the slurry. Ultrafiltration permeate can be recycled for rinsing water in the CMP process and the solid is recovered out for CMP usages. Ndiaye *et al.*

(2004) investigated the possibility using an ultrafiltration pilot equipped with a module of polysulfone hollow fibers to concentrate the used slurry and reducing the waste volume sent to wastewater treatment plant. Moreover, backwashing the ultrafiltration is one of the important elements to maintain its operating life and also to reduce the maintenance cost to avoid frequent membrane change. Ultrafiltration showed good recovery of large fraction of initial effluent (almost 95%) so that the relatively transparent and colorless water could be recycled back into the CMP operation. This was a good indication for the use to concentrate CMP, used slurry (Ndieye *et al.*, 2004).

As reviewed by several authors, it was essential to use pretreatment or prefilter. It usually be made up of a single filter to get rid any waste and agglomerated particles in order to obtain particular particle sizes and also as a prevention from quick deterioration of the membrane module (Testa *et al.*, 2014; Coetsier *et al.*, 2011; Ndieye *et al.*, 2004).

Recovery process of spent tungsten slurry involves the removal of dissolved and undissolved metals in the slurry suspension. Abdullah *et al.* (2007) concluded that the recovery method is not an easy task as the composition of the spent tungsten slurry is quite complex. Both physical and chemical properties of the slurry components are expected to change depending on the conditions and type of treatment that they receive.

1.2 Problem Statement

Tungsten slurry for use in the CMP operation is a rather complex mixture with the presence of multi components, each of which serves an individual purpose. Both physical and chemical properties of the components in the slurry are expected to change depending on the condition and type of treatments that they receive. The composition of spent slurry could significantly change as multi foreign components could be introduced into the slurry. As such, thorough characterization of the spent tungsten slurry is general critical for knowledge on its physical and chemical properties is critically needed in order to consider suitable recovery methods. Various methods for recovering abrasive silica from the spent slurry should be considered based on the composition and also their conditions. Therefore, membrane technology is one of the potential methods to treat and recycle the spent tungsten slurry into the CMP process. The treatment of spent tungsten slurry using ultrafiltration membrane process resulted high retention of silica particles of about 93% (Sheikholeslami et al., 2000). Good recovery of water from ultrafiltration of spent tungsten slurry of approximately 95% which good indication for its potential in concentrating spent slurry and reducing pollutants (Ndiaye et. al., 2004). However, the chemical components of spent slurry might not be fully reinstated and it may contain metal contamination that may disturb overall quality of slurry.

The critical problem generally encounter in using ultrafiltration membrane separation process as the recovery system for spent tungsten slurry is membrane fouling. Membrane fouling of the ultrafiltration membrane process is affected by the deposition and build up of silica particles from spent tungsten slurry on the membrane surface and its internal pores. Besides, the extent of membrane fouling by silica particles is greatly influenced by the solution chemistry (ionic strength and pH solution), operating conditions of membrane system (TMP and velocity), membrane type, module configuration and process itself. Too high of TMP and velocity will lead to severe fouling condition causing to high usage of energy and operational cost.