



Crystal structure of 5-[4-(dimethylamino)phenyl]-3-(4-methylphenyl)-4,5-dihydro-1H-pyrazole-1-carbaldehyde

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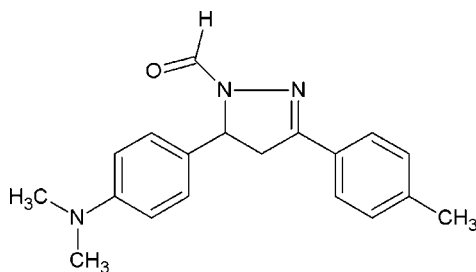
The title compound, C₁₉H₂₁N₃O, comprises a central pyrazole ring which is N-connected to an aldehyde group and C-connected twice to substituted benzene rings. The pyrazole ring is twisted on the C–C single bond, and the least-squares plane through this ring forms dihedral angles of 82.44 (5) and 4.52 (5)° with the (dimethylamino)benzene and *p*-tolyl rings, respectively. In the crystal, weak C–H···O hydrogen bonds link molecules into supramolecular tubes along the *b* axis.

Keywords: crystal structure; substituted pyrazole; pyrazole derivatives; pharmacological properties.

CCDC reference: 1440601

1. Related literature

For pharmacological properties of pyrazole derivatives, see: Sarojini *et al.* (2010); Samshuddin *et al.* (2012). For their industrial applications, see: Wiley *et al.* (1958); Lu *et al.* (1999). For related structures, see Fun *et al.* (2010); Baktr *et al.* (2011).



2. Experimental

2.1. Crystal data

C ₁₉ H ₂₁ N ₃ O	<i>V</i> = 3181.0 (4) Å ³
<i>M_r</i> = 307.39	<i>Z</i> = 8
Monoclinic, <i>C</i> 2/ <i>c</i>	Mo <i>K</i> α radiation
<i>a</i> = 21.9524 (15) Å	<i>μ</i> = 0.08 mm ⁻¹
<i>b</i> = 6.2511 (4) Å	<i>T</i> = 100 K
<i>c</i> = 24.1521 (16) Å	0.45 × 0.26 × 0.15 mm
<i>β</i> = 106.3069 (9)°	

2.2. Data collection

Bruker APEX DUO CCD area-detector diffractometer	27492 measured reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	4750 independent reflections
<i>T</i> _{min} = 0.921, <i>T</i> _{max} = 0.962	4090 reflections with <i>I</i> > 2σ(<i>I</i>)
	<i>R</i> _{int} = 0.028

2.3. Refinement

<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.043	211 parameters
<i>wR</i> (<i>F</i> ²) = 0.122	H-atom parameters constrained
<i>S</i> = 1.04	Δ <i>ρ</i> _{max} = 0.37 e Å ⁻³
4750 reflections	Δ <i>ρ</i> _{min} = -0.22 e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C1–H1A···O1 ⁱ	0.95	2.52	3.4175 (12)	158
C19–H19A···O1 ⁱⁱ	0.98	2.45	3.3902 (15)	161

Symmetry codes: (i) *x*, *y* – 1, *z*; (ii) –*x* + ½, *y* – ½, –*z* + ½.

Data collection: *APEX2* (Bruker, 2014); cell refinement: *SAINTE* (Bruker, 2014); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS2013* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2014* (Sheldrick, 2015); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *PLATON* (Spek, 2009).

Acknowledgements

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Supporting information for this paper is available from the IUCr electronic archives (Reference: TK5413).

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supporting information

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Crystal structure of 5-[4-(dimethylamino)phenyl]-3-(4-methylphenyl)-4,5-dihydro-1*H*-pyrazole-1-carbaldehyde

Farook Adam, Seranthimata Samshuddin, Nadiah Ameram, Subramaya and Laxminarayana Samartha

S1. Introduction

Pyrazolyl derivatives are well-known for their versatile pharmacological activities (Sarojini *et al.*, 2010; Samshuddin *et al.*, 2012). In addition, many 1,3,5-triaryl-2-pyrazolyls have a variety of industrial applications such as functioning as scintillation solutes (Wiley *et al.*, 1958) and fluorescent agents (Lu *et al.*, 1999). The crystal structures of some pyrazolyls containing a *N*-alkyl chain *viz.*, 3,5-bis(4-fluorophenyl)-4,5-dihydro-1*H*-pyrazole-1-carbaldehyde (Baktir *et al.*, 2011) and 1-[3,5-bis(4-fluorophenyl)-4,5-dihydro-1*H*-pyrazol-1-yl]ethanone (Fun *et al.*, 2010) have been reported. In view of the importance of pyrazolines, the title compound (I) was prepared and its crystal structure reported.

S2. Supramolecular features

The asymmetric unit of (I) consists of a single crystallographic independent molecule as shown in Fig. 1. The pyrazoline ring (N1/N2/C7/C8/C9) is twisted about the C8—C7 bond [$Q2 = 0.0964(10)$ Å and $\varphi2 = 133.5(6)^\circ$] with maximum deviations of 0.057(1) and -0.053(1) Å from its mean plane for atoms C7 and C8, respectively. The methyl-substituted phenyl ring (C10—C15) and dimethylamino-substituted phenyl ring (C1—C6) make dihedral angles of 4.52(5) and 82.44(5)°, respectively, with the pyrazoline ring. In crystal, molecules are connected by weak C—H···O hydrogen bonds into one-dimensional spiral-like chains (Fig. 2), propagating along the crystallographic *b*-axis.

S3. Synthesis and crystallization

A mixture of (2*E*)-3-[4-(dimethylamino)phenyl]-1-(4-methylphenyl)prop-2-en-1-one (2.65 g, 0.01 mol) and hydrazine hydrate (1 ml) in 30 ml formic acid was refluxed for 6 h. The reaction mixture was cooled and poured into 250 ml ice-cold water. The precipitate formed was collected by filtration and purified by recrystallization from ethanol. Single crystals were grown from ethyl acetate by slow evaporation (m.p 473–476 K; yield: 68%).

S4. Refinement

The carbon-bound H-atoms were placed in calculated positions (C—H = 0.95–1.00 Å) and were included in the refinement in the riding model approximation, with $U_{\text{iso}}(\text{H})$ set to 1.2–1.5 $U_{\text{equiv}}(\text{C})$. A rotating group model was applied to methyl groups.

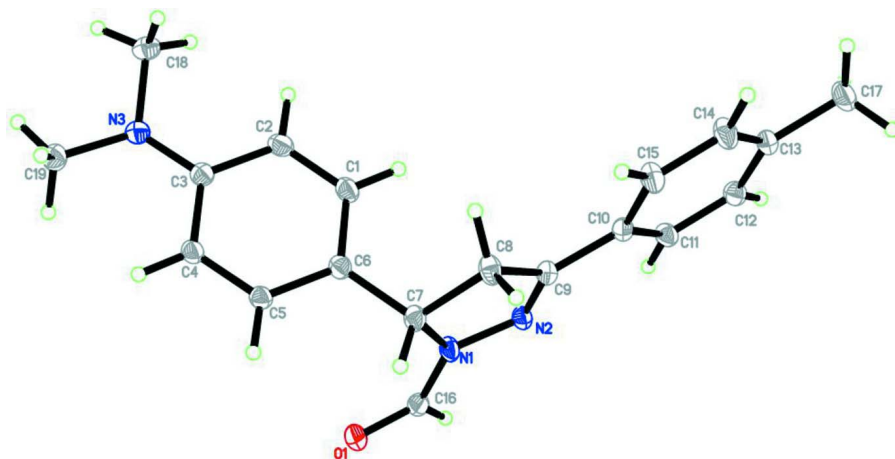


Figure 1

The molecular structure of (I), showing the atom labels and 50% probability displacement ellipsoids.

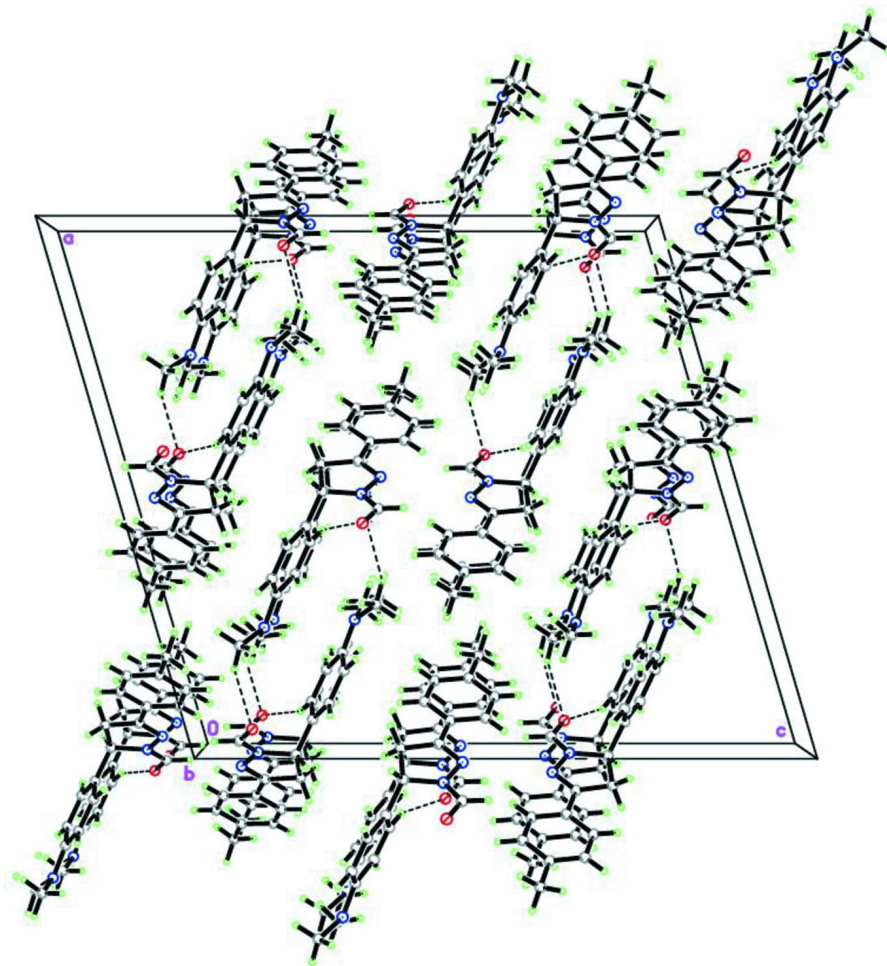


Figure 2

The crystal packing of (I), viewed along the *b* axis.

5-[4-(Dimethylamino)phenyl]-3-(4-methylphenyl)-4,5-dihydro-1H-pyrazole-1-carbaldehyde

Crystal data

C₁₉H₂₁N₃O $M_r = 307.39$ Monoclinic, *C2/c* $a = 21.9524 (15) \text{ \AA}$ $b = 6.2511 (4) \text{ \AA}$ $c = 24.1521 (16) \text{ \AA}$ $\beta = 106.3069 (9)^\circ$ $V = 3181.0 (4) \text{ \AA}^3$ $Z = 8$ $F(000) = 1312$ $D_x = 1.284 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 9179 reflections

 $\theta = 3.0\text{--}30.3^\circ$ $\mu = 0.08 \text{ mm}^{-1}$ $T = 100 \text{ K}$

Block, colourless

 $0.45 \times 0.26 \times 0.15 \text{ mm}$

Data collection

Bruker APEX DUO CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 φ and ω scansAbsorption correction: multi-scan
(*SADABS*; Sheldrick, 1996) $T_{\min} = 0.921$, $T_{\max} = 0.962$

27492 measured reflections

4750 independent reflections

4090 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.028$ $\theta_{\max} = 30.3^\circ$, $\theta_{\min} = 1.8^\circ$ $h = -30 \rightarrow 29$ $k = -8 \rightarrow 8$ $l = -34 \rightarrow 34$

Refinement

Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.043$ $wR(F^2) = 0.122$ $S = 1.04$

4750 reflections

211 parameters

0 restraints

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0683P)^2 + 1.9935P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.37 \text{ e \AA}^{-3}$ $\Delta\rho_{\min} = -0.22 \text{ e \AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.06265 (4)	0.29904 (12)	0.11364 (3)	0.02109 (17)
N1	0.01154 (4)	-0.01958 (13)	0.10464 (3)	0.01627 (17)
N2	-0.01890 (4)	-0.18157 (13)	0.06779 (3)	0.01565 (17)
N3	0.25348 (4)	-0.20563 (14)	0.32043 (4)	0.01865 (18)
C1	0.09931 (5)	-0.28895 (16)	0.21033 (4)	0.01817 (19)
H1A	0.0779	-0.4002	0.1857	0.022*
C2	0.15816 (5)	-0.33020 (16)	0.24891 (4)	0.01840 (19)
H2A	0.1757	-0.4699	0.2508	0.022*
C3	0.19237 (5)	-0.16856 (15)	0.28528 (4)	0.01578 (18)
C4	0.16344 (5)	0.03445 (15)	0.28189 (4)	0.01599 (18)

H4A	0.1847	0.1467	0.3062	0.019*
C5	0.10411 (5)	0.07238 (15)	0.24333 (4)	0.01551 (18)
H5A	0.0858	0.2108	0.2418	0.019*
C6	0.07085 (5)	-0.08737 (15)	0.20693 (4)	0.01555 (18)
C7	0.00689 (5)	-0.04206 (15)	0.16474 (4)	0.01606 (18)
H7A	-0.0115	0.0911	0.1764	0.019*
C8	-0.04155 (5)	-0.22678 (17)	0.15670 (4)	0.01855 (19)
H8A	-0.0255	-0.3427	0.1850	0.022*
H8B	-0.0828	-0.1755	0.1607	0.022*
C9	-0.04769 (4)	-0.30150 (15)	0.09584 (4)	0.01509 (18)
C10	-0.08377 (4)	-0.48998 (15)	0.06946 (4)	0.01530 (18)
C11	-0.08705 (5)	-0.55364 (16)	0.01292 (4)	0.01714 (19)
H11A	-0.0660	-0.4718	-0.0093	0.021*
C12	-0.12078 (5)	-0.73494 (16)	-0.01069 (4)	0.01826 (19)
H12A	-0.1226	-0.7754	-0.0490	0.022*
C13	-0.15214 (5)	-0.85946 (16)	0.02091 (4)	0.0189 (2)
C14	-0.14841 (5)	-0.79645 (18)	0.07700 (5)	0.0228 (2)
H14A	-0.1690	-0.8797	0.0993	0.027*
C15	-0.11508 (5)	-0.61392 (17)	0.10118 (4)	0.0203 (2)
H15A	-0.1136	-0.5733	0.1394	0.024*
C16	0.03751 (4)	0.14662 (16)	0.08404 (4)	0.01661 (18)
H16A	0.0363	0.1457	0.0444	0.020*
C17	-0.18897 (6)	-1.05622 (18)	-0.00455 (5)	0.0265 (2)
H17A	-0.1673	-1.1291	-0.0295	0.040*
H17B	-0.2317	-1.0148	-0.0273	0.040*
H17C	-0.1919	-1.1528	0.0266	0.040*
C18	0.27313 (5)	-0.42547 (17)	0.33548 (5)	0.0208 (2)
H18A	0.3182	-0.4282	0.3570	0.031*
H18B	0.2665	-0.5105	0.3002	0.031*
H18C	0.2479	-0.4857	0.3594	0.031*
C19	0.28414 (5)	-0.04748 (17)	0.36345 (5)	0.0205 (2)
H19A	0.3290	-0.0845	0.3794	0.031*
H19B	0.2634	-0.0456	0.3945	0.031*
H19C	0.2806	0.0941	0.3454	0.031*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0212 (3)	0.0179 (3)	0.0216 (4)	-0.0049 (3)	0.0017 (3)	-0.0014 (3)
N1	0.0194 (4)	0.0166 (4)	0.0116 (3)	-0.0050 (3)	0.0023 (3)	-0.0016 (3)
N2	0.0171 (4)	0.0146 (4)	0.0137 (4)	-0.0027 (3)	0.0017 (3)	-0.0016 (3)
N3	0.0183 (4)	0.0157 (4)	0.0204 (4)	0.0010 (3)	0.0028 (3)	0.0008 (3)
C1	0.0234 (5)	0.0151 (4)	0.0158 (4)	-0.0033 (3)	0.0051 (3)	-0.0029 (3)
C2	0.0233 (5)	0.0137 (4)	0.0184 (4)	0.0003 (3)	0.0063 (4)	-0.0006 (3)
C3	0.0185 (4)	0.0156 (4)	0.0141 (4)	-0.0009 (3)	0.0059 (3)	0.0007 (3)
C4	0.0194 (4)	0.0144 (4)	0.0138 (4)	-0.0018 (3)	0.0041 (3)	-0.0013 (3)
C5	0.0194 (4)	0.0135 (4)	0.0135 (4)	-0.0005 (3)	0.0044 (3)	-0.0003 (3)
C6	0.0187 (4)	0.0157 (4)	0.0121 (4)	-0.0021 (3)	0.0041 (3)	-0.0007 (3)

C7	0.0182 (4)	0.0175 (4)	0.0121 (4)	-0.0028 (3)	0.0037 (3)	-0.0018 (3)
C8	0.0199 (4)	0.0220 (5)	0.0145 (4)	-0.0060 (4)	0.0060 (3)	-0.0034 (3)
C9	0.0143 (4)	0.0165 (4)	0.0138 (4)	-0.0010 (3)	0.0030 (3)	-0.0012 (3)
C10	0.0143 (4)	0.0157 (4)	0.0150 (4)	-0.0014 (3)	0.0026 (3)	-0.0002 (3)
C11	0.0198 (4)	0.0162 (4)	0.0154 (4)	-0.0026 (3)	0.0051 (3)	0.0004 (3)
C12	0.0212 (4)	0.0175 (4)	0.0148 (4)	-0.0018 (3)	0.0029 (3)	-0.0011 (3)
C13	0.0182 (4)	0.0171 (4)	0.0187 (4)	-0.0039 (3)	0.0007 (3)	0.0004 (3)
C14	0.0236 (5)	0.0257 (5)	0.0194 (5)	-0.0110 (4)	0.0063 (4)	0.0002 (4)
C15	0.0214 (5)	0.0238 (5)	0.0160 (4)	-0.0073 (4)	0.0057 (4)	-0.0021 (4)
C16	0.0160 (4)	0.0167 (4)	0.0158 (4)	-0.0017 (3)	0.0024 (3)	0.0018 (3)
C17	0.0296 (5)	0.0218 (5)	0.0240 (5)	-0.0107 (4)	0.0010 (4)	-0.0018 (4)
C18	0.0230 (5)	0.0185 (5)	0.0214 (5)	0.0047 (4)	0.0069 (4)	0.0027 (4)
C19	0.0177 (4)	0.0208 (5)	0.0213 (5)	0.0008 (4)	0.0026 (4)	-0.0018 (4)

Geometric parameters (Å, °)

O1—C16	1.2256 (12)	C8—H8B	0.9900
N1—C16	1.3459 (12)	C9—C10	1.4622 (13)
N1—N2	1.3892 (11)	C10—C15	1.3982 (13)
N1—C7	1.4901 (12)	C10—C11	1.4049 (13)
N2—C9	1.2889 (12)	C11—C12	1.3858 (13)
N3—C3	1.3908 (12)	C11—H11A	0.9500
N3—C19	1.4550 (13)	C12—C13	1.3991 (14)
N3—C18	1.4556 (13)	C12—H12A	0.9500
C1—C2	1.3884 (14)	C13—C14	1.3910 (15)
C1—C6	1.3986 (14)	C13—C17	1.5051 (14)
C1—H1A	0.9500	C14—C15	1.3922 (14)
C2—C3	1.4094 (13)	C14—H14A	0.9500
C2—H2A	0.9500	C15—H15A	0.9500
C3—C4	1.4112 (13)	C16—H16A	0.9500
C4—C5	1.3923 (13)	C17—H17A	0.9800
C4—H4A	0.9500	C17—H17B	0.9800
C5—C6	1.3938 (13)	C17—H17C	0.9800
C5—H5A	0.9500	C18—H18A	0.9800
C6—C7	1.5121 (13)	C18—H18B	0.9800
C7—C8	1.5446 (14)	C18—H18C	0.9800
C7—H7A	1.0000	C19—H19A	0.9800
C8—C9	1.5118 (13)	C19—H19B	0.9800
C8—H8A	0.9900	C19—H19C	0.9800
C16—N1—N2	120.22 (8)	C15—C10—C11	118.47 (9)
C16—N1—C7	125.80 (8)	C15—C10—C9	119.82 (9)
N2—N1—C7	113.78 (7)	C11—C10—C9	121.70 (8)
C9—N2—N1	107.81 (8)	C12—C11—C10	120.48 (9)
C3—N3—C19	119.75 (8)	C12—C11—H11A	119.8
C3—N3—C18	118.50 (8)	C10—C11—H11A	119.8
C19—N3—C18	114.70 (8)	C11—C12—C13	121.23 (9)
C2—C1—C6	121.52 (9)	C11—C12—H12A	119.4

C2—C1—H1A	119.2	C13—C12—H12A	119.4
C6—C1—H1A	119.2	C14—C13—C12	118.05 (9)
C1—C2—C3	121.18 (9)	C14—C13—C17	120.48 (9)
C1—C2—H2A	119.4	C12—C13—C17	121.47 (9)
C3—C2—H2A	119.4	C13—C14—C15	121.39 (9)
N3—C3—C2	120.99 (9)	C13—C14—H14A	119.3
N3—C3—C4	121.77 (9)	C15—C14—H14A	119.3
C2—C3—C4	117.14 (9)	C14—C15—C10	120.38 (9)
C5—C4—C3	120.88 (9)	C14—C15—H15A	119.8
C5—C4—H4A	119.6	C10—C15—H15A	119.8
C3—C4—H4A	119.6	O1—C16—N1	123.54 (9)
C4—C5—C6	121.73 (9)	O1—C16—H16A	118.2
C4—C5—H5A	119.1	N1—C16—H16A	118.2
C6—C5—H5A	119.1	C13—C17—H17A	109.5
C5—C6—C1	117.53 (9)	C13—C17—H17B	109.5
C5—C6—C7	120.91 (8)	H17A—C17—H17B	109.5
C1—C6—C7	121.55 (8)	C13—C17—H17C	109.5
N1—C7—C6	111.69 (8)	H17A—C17—H17C	109.5
N1—C7—C8	100.43 (7)	H17B—C17—H17C	109.5
C6—C7—C8	114.89 (8)	N3—C18—H18A	109.5
N1—C7—H7A	109.8	N3—C18—H18B	109.5
C6—C7—H7A	109.8	H18A—C18—H18B	109.5
C8—C7—H7A	109.8	N3—C18—H18C	109.5
C9—C8—C7	102.92 (8)	H18A—C18—H18C	109.5
C9—C8—H8A	111.2	H18B—C18—H18C	109.5
C7—C8—H8A	111.2	N3—C19—H19A	109.5
C9—C8—H8B	111.2	N3—C19—H19B	109.5
C7—C8—H8B	111.2	H19A—C19—H19B	109.5
H8A—C8—H8B	109.1	N3—C19—H19C	109.5
N2—C9—C10	121.69 (9)	H19A—C19—H19C	109.5
N2—C9—C8	114.10 (8)	H19B—C19—H19C	109.5
C10—C9—C8	124.20 (8)		
C16—N1—N2—C9	-170.18 (9)	C1—C6—C7—C8	38.39 (12)
C7—N1—N2—C9	4.90 (11)	N1—C7—C8—C9	9.05 (9)
C6—C1—C2—C3	-1.62 (15)	C6—C7—C8—C9	-110.93 (9)
C19—N3—C3—C2	-171.16 (9)	N1—N2—C9—C10	-179.08 (8)
C18—N3—C3—C2	-22.08 (13)	N1—N2—C9—C8	2.03 (11)
C19—N3—C3—C4	12.60 (14)	C7—C8—C9—N2	-7.57 (11)
C18—N3—C3—C4	161.68 (9)	C7—C8—C9—C10	173.57 (9)
C1—C2—C3—N3	-174.66 (9)	N2—C9—C10—C15	-179.05 (9)
C1—C2—C3—C4	1.75 (14)	C8—C9—C10—C15	-0.28 (14)
N3—C3—C4—C5	175.25 (9)	N2—C9—C10—C11	2.17 (15)
C2—C3—C4—C5	-1.13 (14)	C8—C9—C10—C11	-179.05 (9)
C3—C4—C5—C6	0.36 (14)	C15—C10—C11—C12	0.19 (15)
C4—C5—C6—C1	-0.14 (14)	C9—C10—C11—C12	178.98 (9)
C4—C5—C6—C7	-179.11 (8)	C10—C11—C12—C13	-0.23 (15)
C2—C1—C6—C5	0.77 (14)	C11—C12—C13—C14	-0.17 (15)

C2—C1—C6—C7	179.72 (9)	C11—C12—C13—C17	179.86 (9)
C16—N1—C7—C6	-72.07 (12)	C12—C13—C14—C15	0.62 (16)
N2—N1—C7—C6	113.18 (9)	C17—C13—C14—C15	-179.41 (10)
C16—N1—C7—C8	165.67 (9)	C13—C14—C15—C10	-0.67 (17)
N2—N1—C7—C8	-9.08 (10)	C11—C10—C15—C14	0.25 (15)
C5—C6—C7—N1	103.78 (10)	C9—C10—C15—C14	-178.56 (9)
C1—C6—C7—N1	-75.15 (11)	N2—N1—C16—O1	176.82 (9)
C5—C6—C7—C8	-142.69 (9)	C7—N1—C16—O1	2.38 (16)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C1—H1 <i>A</i> \cdots O1 ⁱ	0.95	2.52	3.4175 (12)	158
C19—H19 <i>A</i> \cdots O1 ⁱⁱ	0.98	2.45	3.3902 (15)	161

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+1/2, y-1/2, -z+1/2$.