



Three closely related (2*E*,2'*E*)-3,3'-(1,4-phenylene)-bis[1-(methoxyphenyl)prop-2-en-1-ones]: supramolecular assemblies in one dimension mediated by hydrogen bonding and C—H... π interactions

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In the title compounds, (2*E*,2'*E*)-3,3'-(1,4-phenylene)bis[1-(2-methoxyphenyl)prop-2-en-1-one], C₂₆H₂₂O₄ (I), (2*E*,2'*E*)-3,3'-(1,4-phenylene)bis[1-(3-methoxyphenyl)prop-2-en-1-one], C₂₆H₂₂O₄ (II) and (2*E*,2'*E*)-3,3'-(1,4-phenylene)bis[1-(3,4-dimethoxyphenyl)prop-2-en-1-one], C₂₈H₂₆O₆ (III), the asymmetric unit consists of a half-molecule, completed by crystallographic inversion symmetry. The dihedral angles between the central and terminal benzene rings are 56.98 (8), 7.74 (7) and 7.73 (7)° for (I), (II) and (III), respectively. In the crystal of (I), molecules are linked by pairs of C—H... π interactions into chains running parallel to [101]. The packing for (II) and (III), features inversion dimers linked by pairs of C—H...O hydrogen bonds, forming *R*₂²(16) and *R*₂²(14) ring motifs, respectively, as parts of [201] and [101] chains, respectively.

1. Chemical context

Chalcones and their derivatives are natural or synthetic 1,3-diaryl-2-propenones that may exist in *cis* and *trans* isomeric forms, the *trans* form being thermodynamically stable. The α,β -unsaturated enone C=C—C(=O)—C moiety in the chalcone structure plays an important role in the biological activities of these species (Husain *et al.*, 2013; Abdel Ghani *et al.*, 2008). As a result of the -enone system, these molecules present relatively low redox potentials and have a greater probability of undergoing electron-transfer reactions. Apart from the biological activities, the photophysical properties of chalcone derivatives also attracted considerable attention from both chemists and physicists. Many chalcone derivatives have been reported in relation to non-linear optics (NLO), photorefractive polymers, holographic recording materials and fluorescent probes for sensing metal ions (Ruzié *et al.*, 2009; Wei *et al.*, 2011; Chandra Shekhara Shetty *et al.*, 2016). As part of our studies in this area, we report herein the syntheses and structures of three new bischalcone derivatives, (2*E*,2'*E*)-3,3'-(1,4-phenylene)bis(1-(2-methoxyphenyl)prop-2-

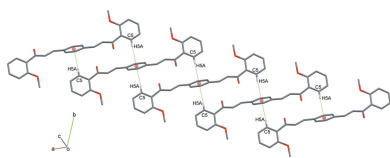


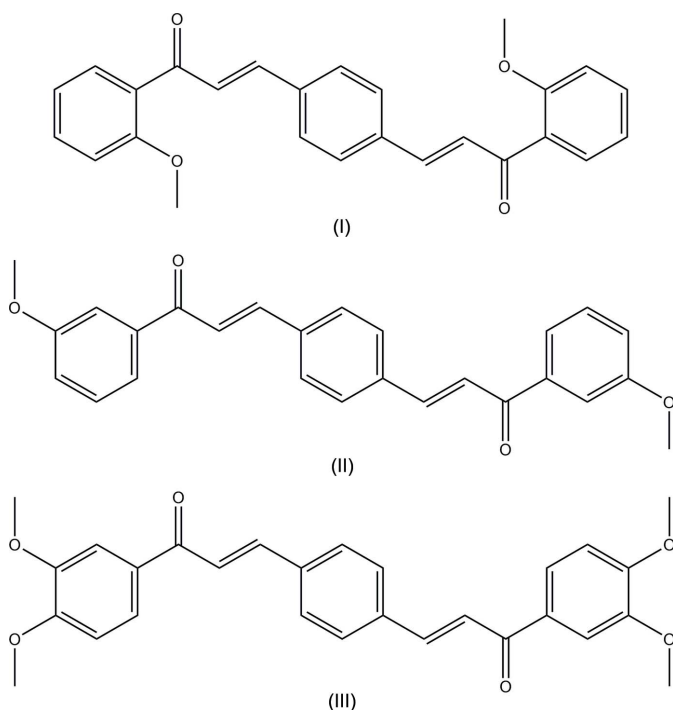
Table 1

 Selected torsion and dihedral angles ($^{\circ}$) for compounds (I)–(III).

Dihedral 1 is the dihedral angle between the mean planes of the terminal methoxyphenyl and central benzene rings.

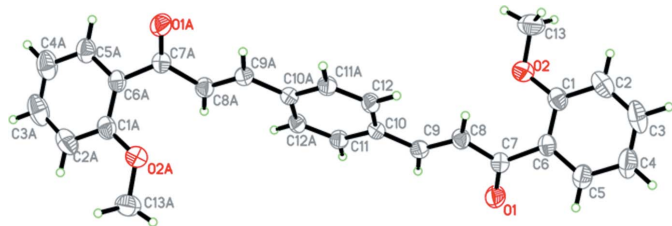
Compound	C1–C6–C7–O1	O1–C7–C8–C9	C8–C9–C10–C11	Dihedral 1
(I)	–13.5 (2)	143.60 (15)	167.44 (15)	56.98 (8)
(II)	0.2 (2)	6.5 (2)	–1.6 (2)	7.74 (7)
(III)	7.3 (2)	7.3 (2)	–5.6 (2)	7.73 (7)

en-1-one), $C_{26}H_{22}O_4$ (I), (2*E*,2'*E*)-3,3'-(1,4-phenylene)bis(1-(3-methoxyphenyl)prop-2-en-1-one), $C_{26}H_{22}O_4$ (II) and (2*E*,2'*E*)-3,3'-(1,4-phenylene)bis(1-(3,4-dimethoxyphenyl)prop-2-en-1-one), $C_{28}H_{26}O_6$ (III).



2. Structural commentary

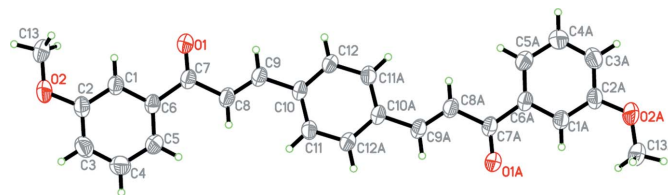
The molecular structures of (I)–(III) are shown in Figs. 1–3, respectively. The asymmetric unit of each compound consists of a half-molecule, with the complete molecule generated by a


Figure 1

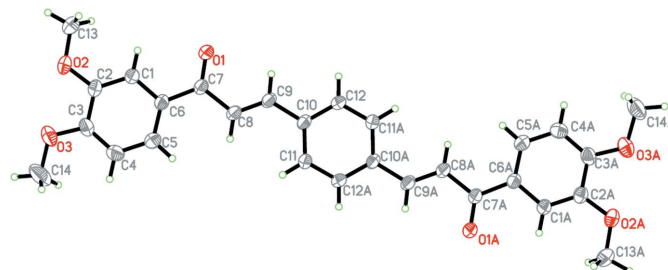
The molecular structure of (I), showing 40% probability displacement ellipsoids. [Symmetry code: (A) $-x, 1 - y, -z$.]

crystallographic inversion centre at the centroid of the central benzene ring.

Each compound is constructed from two aromatic rings (centre benzene and terminal methoxyphenyl rings), which are linked by a C=C(C=O)–C enone bridge. Despite having an extra methoxy substituent, the conformation of compounds (II) and (III) are very similar, as indicated by the dihedral angles between the rings of 7.74 (7) and 7.73 (7) $^{\circ}$, respectively. The enone linkage moiety of compounds (II) and (III) has similar torsion angles [O1–C7–C8–C9 = 0.2 (2) and 7.3 (2) $^{\circ}$, respectively], but compound (II) has a higher overall planarity than compound (III), as its enone bridge forms a smaller torsion angle with the methoxyphenyl ring [C1–C6–C7–O1 = –6.5 (2) $^{\circ}$] and benzene ring [C8–C9–C10–C11 = –1.7 (2) $^{\circ}$; C1–C6–C7–O1 = 7.3 (2) $^{\circ}$ and C8–C9–C10–C11 = –5.6 (2) $^{\circ}$ in (III)]. Compared to the nearly coplanar arrangement of rings in compounds (II) and (III), compound (I) is substantially twisted [O1–C7–C8–C9 = –13.5 (2) $^{\circ}$ and C1–C6–C7–O1 = 143.60 (15) $^{\circ}$] about the enone bridge, which may arise from steric repulsion with the *ortho*-O2 atom. Hence, the dihedral angle between the 2-methoxyphenyl and benzene rings in (I) increases to 56.98 (8) $^{\circ}$. Key torsion angles are tabulated in Table 1. The C


Figure 2

The molecular structure of (II), showing 40% probability displacement ellipsoids. [Symmetry code: (A) $1 - x, 1 - y, 1 - z$.]


Figure 3

The molecular structure of (III), showing 40% probability displacement ellipsoids. [Symmetry code: (A) $2 - x, -y, 1 - z$.]

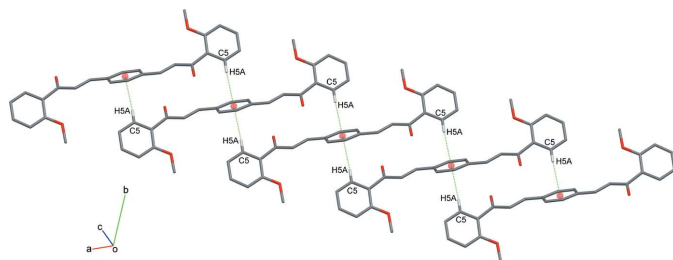


Figure 4
Fragment of a [101] chain of molecules of (I) linked by pairs of weak C—H··· π interactions (dashed lines).

atoms of the methoxy groups are close to the planes of their attached rings in all cases: for (I), deviation of C13 = 0.163 (2) Å, for (II), deviation of C13 = 0.329 (2) Å, and for (III), deviations of C13 and C14 = 0.091 (2) and -0.266 (2) Å, respectively.

3. Supramolecular features

The packing of (I) is consolidated by a weak C—H··· π contact (Table 2) involving a hydrogen atom from the phenyl ring and the centroid of the central benzene ring (C10—C12/C10A—C12A). This C—H··· π interaction connects the molecules of (I) into chains parallel to the [101] direction with a C—H··· π distance of 2.74 Å (Fig. 4). In the supramolecular assemblies of compounds (II) and (III), the molecules are connected by pairs of C—H···O hydrogen bonds (Table 2) into inversion dimers, which form $R_2^2(16)$ and $R_2^2(14)$ ring motifs, respectively. The dimers are parts of [201] chains (Fig. 5) in (II), while molecules in compound (III) are parts of chains propagating in the [101] direction (Fig. 6).

4. Database survey

A search of the Cambridge Structural Database (CSD, Version 5.38, last update November 2016; Groom *et al.*, 2016) using (2*E*,2'*E*)-3,3'-(1,4-phenylene)bis(1-phenylprop-2-en-1-one) as the main skeleton, revealed the presence of four structures containing the bischalcone moiety with different substituents, similar to the title compounds in this study. These include 3,3'-(1,4-phenylene)bis[1-(*X*)prop-2-en-1-one], where *X* = 2-hydroxyphenyl (DIDNUB; Gaur & Mishra, 2013), 4-chlorophenyl (KIKFUG; Harrison *et al.*, 2007), 4-methoxy-

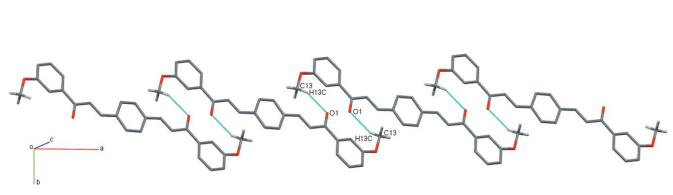


Figure 5
Fragment of a [201] chain of molecules of (II) linked by pairs of weak C—H···O interactions (dashed lines).

Table 2
Hydrogen-bonding geometry (Å, °) for compounds (I)–(III).

Cg1 is the centroid of the C10—C12/C10A—C12A ring.

Compound	D—H···A	D—H	H···A	D···A	D—H···A
(I)	C5—H5A···Cg1 ⁱ	0.93	2.74	3.491 (3)	139
(II)	C13—H13C···O1 ⁱⁱ	0.96	2.60	3.503 (3)	157
(III)	C12—H12A···O1 ⁱⁱⁱ	0.96	2.47	3.337 (3)	156

Symmetry codes: (i) 1 + *x*, *y*, 1 + *z*; (ii) 3 - *x*, 1 - *y*, 2 - *z*; (iii) 1 - *x*, -*y*, -*z*.

phenyl (UDUPUF; Harrison *et al.*, 2007*a*) and 3,4-methoxyphenyl (UDUQAM; Harrison *et al.*, 2007*b*). It is notable that UDUPUF and UDUQAH are positional isomers of compounds (I) and (II), and (III), respectively, differing from them only in the location of the methoxy substituent (see scheme below). The dihedral angles between the central and terminal phenyl ring in these compounds vary from 10.9 to 46.3°. In terms of the title compounds, (II) and (III) are more planar [7.74 (7) and 7.73 (7)°] while compound (I) is more twisted [56.98 (8)°]. The supramolecular assembly in UDUPUF also depends upon a single C—H···O hydrogen bond between inversion-related pairs of molecules, forming a centrosymmetric dimer.

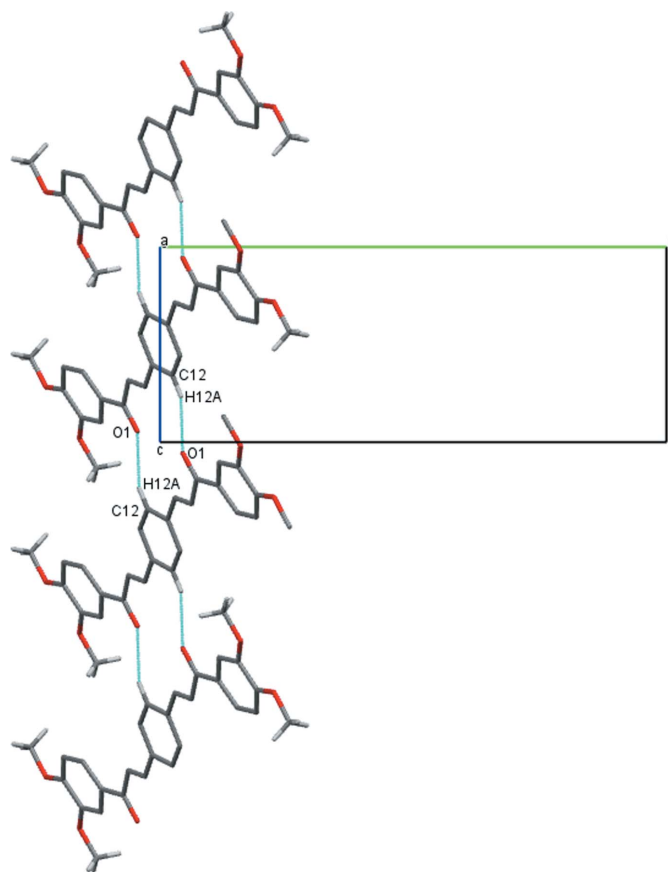
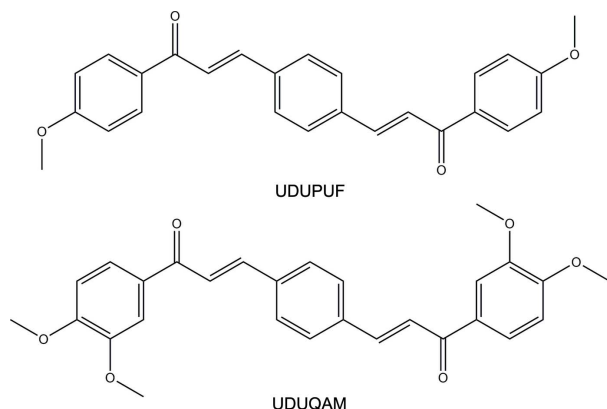


Figure 6
Fragment of a [101] chain of molecules of (III) linked by pairs of weak C—H···O interactions (dashed lines).



5. Synthesis and crystallization

A mixture of the corresponding methoxyacetophenone (0.02 mol) and terephthalaldehyde (0.01 mol) was dissolved in methanol (20 ml). A catalytic amount of NaOH was added to the solution dropwise with vigorous stirring. The reaction mixtures were stirred for about 5–6 h at room temperature. The resultant crude products were filtered, washed successively with distilled water and recrystallized from ethanol to obtain the title compounds. Yellow blocks [(I) and (III)] and yellow needles (II) were recrystallized using the solvents noted below.

Table 3
Experimental details.

	(I)	(II)	(III)
Crystal data			
Chemical formula	C ₂₆ H ₂₂ O ₄	C ₂₆ H ₂₂ O ₄	C ₂₈ H ₂₆ O ₆
<i>M_r</i>	398.43	398.43	458.49
Crystal system, space group	Monoclinic, <i>P</i> ₂ ₁ / <i>c</i>	Monoclinic, <i>P</i> ₂ ₁ / <i>c</i>	Monoclinic, <i>P</i> ₂ ₁ / <i>n</i>
Temperature (K)	294	294	294
<i>a</i> , <i>b</i> , <i>c</i> (Å)	7.1078 (11), 24.544 (4), 6.0449 (9)	5.2626 (5), 15.6157 (14), 12.4824 (11)	6.9595 (6), 21.0272 (17), 8.3297 (7)
β (°)	101.898 (2)	98.760 (2)	103.602 (2)
<i>V</i> (Å ³)	1031.9 (3)	1013.83 (16)	1184.77 (17)
<i>Z</i>	2	2	2
Radiation type	Mo <i>K</i> α	Mo <i>K</i> α	Mo <i>K</i> α
μ (mm ⁻¹)	0.09	0.09	0.09
Crystal size (mm)	0.39 × 0.35 × 0.18	0.90 × 0.48 × 0.09	0.35 × 0.27 × 0.16
Data collection			
Diffractometer	Bruker APEXII DUO CCD area-detector	Bruker APEXII DUO CCD area-detector	Bruker APEXII DUO CCD area-detector
Absorption correction	Multi-scan (<i>SADABS</i> ; Bruker, 2012)	Multi-scan (<i>SADABS</i> ; Bruker, 2012)	Multi-scan (<i>SADABS</i> ; Bruker, 2012)
<i>T</i> _{min} , <i>T</i> _{max}	0.883, 0.985	0.874, 0.992	0.890, 0.985
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	8322, 2183, 1705	17791, 2458, 1574	12532, 3165, 2328
<i>R</i> _{int}	0.024	0.043	0.030
(<i>sin</i> θ / λ) _{max} (Å ⁻¹)	0.634	0.662	0.683
Refinement			
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.040, 0.119, 1.03	0.044, 0.132, 1.03	0.049, 0.136, 1.04
No. of reflections	2183	2458	3165
No. of parameters	137	137	156
H-atom treatment	H-atom parameters constrained	H-atom parameters constrained	H-atom parameters constrained
$\Delta\rho$ _{max} , $\Delta\rho$ _{min} (e Å ⁻³)	0.19, -0.14	0.14, -0.14	0.20, -0.16

Computer programs: *APEX2* and *SAINT* (Bruker, 2012), *SHELXS97* (Sheldrick, 2008), *SHELXL2013* (Sheldrick, 2015), *Mercury* (Macrae *et al.*, 2006) and *PLATON* (Spek, 2009).

(2*E*,2'*E*)-3,3'-(1,4-phenylene)bis(1-(2-methoxyphenyl)prop-2-en-1-one), C₂₆H₂₂O₄ (I)

Solvent for growing crystals: acetone; yield 85%, m.p. 429–431 K. FT-IR [ATR (solid) cm⁻¹]: 3010 (Ar, C–H, ν), 2945 (methyl, C–H, ν s), 2842 (methyl, C–H, ν), 1658 (C=O, ν), 1598, 1417(Ar, C=C, ν), 1245, 1055 (C–O, ν). ¹H NMR (500 MHz, CDCl₃): δ ppm 7.677–7.632 (*m*, 8H, ⁵CH, ⁸CH, ¹¹CH, ¹²CH), 7.536–7.504 (*t*, 2H, *J* = 8.0 Hz ³CH), 7.496–7.437 (*d*, 2H, *J* = 15.9 Hz, ⁹CH), 7.095–7.065 (*t*, 2H, *J* = 8.0 Hz, ⁴CH), 7.048–7.031 (*d*, 2H, *J* = 8.0 Hz, ²CH), 3.944 (*s*, 6H, ¹³CH₃). ¹³C NMR (125 MHz, CDCl₃): δ ppm 192.61 (C7), 158.23 (C1), 141.48 (C9), 136.95 (C10), 133.09 (C3), 130.47 (C5), 129.17 (C6), 128.84 (C11, C12), 127.91 (C4), 120.84 (C8), 111.69 (C2), 55.80 (C13).

(2*E*,2'*E*)-3,3'-(1,4-phenylene)bis(1-(3-methoxyphenyl)prop-2-en-1-one), C₂₆H₂₂O₄ (II)

Solvent for growing crystals: chloroform and methanol; yield 85%, m.p. 444–446 K. FT-IR [ATR (solid) cm⁻¹]: 3074 (Ar, C–H, ν), 2952 (methyl, C–H, ν s), 2839 (methyl, C–H, ν), 1658 (C=O, ν), 1582, 1414 (Ar, C=C, ν), 1259, 1022 (C–O, ν). ¹H NMR (500 MHz, CDCl₃): δ ppm 7.855–7.823 (*d*, 2H, *J* = 15.7 Hz, ⁸CH), 7.722 (*s*, 4H, ¹¹CH, ¹²CH), 7.650–7.635 (*d*, 2H, *J* = 8.0 Hz, ⁵CH), 7.606–7.574 (*m*, 2H, ¹CH, ⁹CH), 7.473–441 (*t*, 2H, *J* = 8.0 Hz, ⁴CH), 7.189–7.172 (*d*, 2H, *J* = 8.0 Hz, ³CH), 3.924 (*s*, 6H, ¹³CH₃). ¹³C NMR (125 MHz, CDCl₃): δ ppm 189.99 (C7), 159.99 (C2), 143.57 (C9), 139.46 (C10),

136.92 (C6), 129.65 (C5), 128.98 (C11, C12), 123.14 (C8), 121.08 (C3), 119.45 (C4), 112.96 (C1), 55.53 (C13)

(2*E*,2'*E*)-3,3'-(1,4-phenylene)bis(1-(3,4-dimethoxyphenyl)-prop-2-en-1-one), C₂₈H₂₆O₆ (III)

Solvent for growing crystals: acetone; yield 85%, m.p. 479–481 K. FT-IR [ATR (solid) cm⁻¹]: 3018 (Ar, C–H, *ν*), 2962 (methyl, C–H, *ν*_s), 2836 (methyl, C–H, *ν*), 1651 (C=O, *ν*), 1592, 1418 (Ar, C=C, *ν*), 1240, 1017 (C–O, *ν*). ¹H NMR (500 MHz, CDCl₃): δ ppm 7.857–7.826 (*d*, 2H, *J* = 15.6 Hz, ⁸CH), 7.740–7.720 (*m*, 6H, ⁵CH, ¹¹CH, ¹²CH), 7.666 (*s*, 2H, ¹CH), 7.651–7.619 (*d*, 2H, *J* = 15.6 Hz, ⁹CH), 6.984–6.967 (*d*, 2H, *J* = 8.4 Hz, ⁴CH), 4.012–4.009 (*d*, 12H, ¹³CH₃, ¹⁴CH₃). ¹³C NMR (125 MHz, CDCl₃): δ ppm 188.34 (C7), 153.47 (C3), 149.37 (C2), 142.80 (C9), 136.94 (C10), 131.22 (C6), 128.89 (C11, C12), 123.11 (C5), 122.62 (C8), 110.82 (C4), 110.01 (C1), 56.15 (C14), 66.10 (C13)

6. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 3. In (I), (II) and (III), the C-bound H atoms were positioned geometrically [C–H = 0.93–0.96 Å] and refined using a riding model with *U*_{iso}(H) = 1.5*U*_{eq}(C-methyl) and 1.2*U*_{eq}(C) for other H atoms.

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supporting information

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Three closely related (2*E*,2'*E*)-3,3'-(1,4-phenylene)bis[1-(methoxyphenyl)-prop-2-en-1-ones]: supramolecular assemblies in one dimension mediated by hydrogen bonding and C—H \cdots π interactions

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Computing details

For all compounds, data collection: *APEX2* (Bruker, 2012); cell refinement: *SAINT* (Bruker, 2012); data reduction: *SAINT* (Bruker, 2012); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2013* (Sheldrick, 2015); molecular graphics: *SHELXL2013* (Sheldrick, 2015) and *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXL2013* (Sheldrick, 2015) and *PLATON* (Spek, 2009).

(I) (2*E*,2'*E*)-3,3'-(1,4-Phenylene)bis[1-(2-methoxyphenyl)prop-2-en-1-one]

Crystal data

$C_{26}H_{22}O_4$	$F(000) = 420$
$M_r = 398.43$	$D_x = 1.282 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$a = 7.1078 (11) \text{ \AA}$	Cell parameters from 3004 reflections
$b = 24.544 (4) \text{ \AA}$	$\theta = 2.9\text{--}26.4^\circ$
$c = 6.0449 (9) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 101.898 (2)^\circ$	$T = 294 \text{ K}$
$V = 1031.9 (3) \text{ \AA}^3$	Block, yellow
$Z = 2$	$0.39 \times 0.35 \times 0.18 \text{ mm}$

Data collection

Bruker APEXII DUO CCD area-detector diffractometer	8322 measured reflections
Radiation source: fine-focus sealed tube	2183 independent reflections
Graphite monochromator	1705 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.024$
Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2012)	$\theta_{\text{max}} = 26.8^\circ$, $\theta_{\text{min}} = 1.7^\circ$
$T_{\text{min}} = 0.883$, $T_{\text{max}} = 0.985$	$h = -8 \rightarrow 8$
	$k = -24 \rightarrow 31$
	$l = -7 \rightarrow 7$

Refinement

Refinement on F^2	137 parameters
Least-squares matrix: full	0 restraints
$R[F^2 > 2\sigma(F^2)] = 0.040$	Primary atom site location: structure-invariant direct methods
$wR(F^2) = 0.119$	Hydrogen site location: inferred from neighbouring sites
$S = 1.03$	
2183 reflections	

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0541P)^2 + 0.2315P]$
 where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.19 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.14 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.51518 (16)	0.40818 (5)	0.74340 (18)	0.0615 (4)
O2	0.57299 (19)	0.32593 (5)	0.2133 (2)	0.0652 (4)
C1	0.7420 (2)	0.33535 (6)	0.3621 (3)	0.0492 (4)
C2	0.9148 (3)	0.31024 (8)	0.3517 (4)	0.0689 (5)
H2A	0.9237	0.2877	0.2307	0.083*
C3	1.0731 (3)	0.31908 (9)	0.5223 (4)	0.0802 (6)
H3A	1.1891	0.3026	0.5138	0.096*
C4	1.0640 (3)	0.35160 (8)	0.7040 (4)	0.0731 (6)
H4A	1.1715	0.3565	0.8194	0.088*
C5	0.8930 (2)	0.37687 (7)	0.7125 (3)	0.0545 (4)
H5A	0.8854	0.3988	0.8357	0.065*
C6	0.7315 (2)	0.37030 (6)	0.5413 (2)	0.0420 (3)
C7	0.5545 (2)	0.40138 (6)	0.5584 (2)	0.0426 (3)
C8	0.4371 (2)	0.42573 (6)	0.3525 (2)	0.0430 (3)
H8A	0.4840	0.4266	0.2198	0.052*
C9	0.2652 (2)	0.44645 (6)	0.3551 (2)	0.0416 (3)
H9A	0.2239	0.4435	0.4909	0.050*
C10	0.13276 (19)	0.47340 (5)	0.1704 (2)	0.0379 (3)
C11	-0.0569 (2)	0.48284 (6)	0.1921 (2)	0.0417 (3)
H11A	-0.0958	0.4713	0.3222	0.050*
C12	0.18710 (19)	0.49116 (6)	-0.0255 (2)	0.0413 (3)
H12A	0.3122	0.4854	-0.0442	0.050*
C13	0.5741 (4)	0.29130 (8)	0.0256 (3)	0.0783 (6)
H13A	0.4472	0.2896	-0.0672	0.117*
H13B	0.6140	0.2554	0.0785	0.117*
H13C	0.6619	0.3054	-0.0614	0.117*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0571 (7)	0.0849 (9)	0.0427 (6)	0.0192 (6)	0.0106 (5)	0.0041 (6)
O2	0.0741 (8)	0.0619 (8)	0.0550 (7)	0.0175 (6)	0.0022 (6)	-0.0104 (6)
C1	0.0528 (9)	0.0429 (8)	0.0539 (9)	0.0073 (7)	0.0153 (7)	0.0099 (7)
C2	0.0705 (12)	0.0593 (11)	0.0838 (13)	0.0179 (9)	0.0322 (11)	0.0019 (9)
C3	0.0479 (11)	0.0703 (13)	0.1266 (19)	0.0173 (9)	0.0276 (12)	0.0082 (13)
C4	0.0414 (9)	0.0636 (12)	0.1088 (16)	0.0044 (8)	0.0027 (10)	0.0043 (11)

C5	0.0437 (8)	0.0469 (9)	0.0693 (11)	0.0002 (7)	0.0032 (7)	0.0037 (7)
C6	0.0406 (7)	0.0372 (7)	0.0488 (8)	0.0027 (6)	0.0107 (6)	0.0094 (6)
C7	0.0398 (7)	0.0430 (8)	0.0448 (8)	0.0017 (6)	0.0084 (6)	0.0031 (6)
C8	0.0436 (8)	0.0441 (8)	0.0421 (8)	0.0055 (6)	0.0108 (6)	0.0051 (6)
C9	0.0436 (8)	0.0444 (8)	0.0369 (7)	0.0047 (6)	0.0082 (6)	0.0003 (6)
C10	0.0388 (7)	0.0374 (7)	0.0371 (7)	0.0033 (5)	0.0071 (5)	-0.0034 (5)
C11	0.0422 (7)	0.0478 (8)	0.0370 (7)	0.0042 (6)	0.0127 (6)	0.0026 (6)
C12	0.0339 (7)	0.0485 (8)	0.0430 (7)	0.0043 (6)	0.0116 (6)	-0.0003 (6)
C13	0.1169 (18)	0.0615 (12)	0.0537 (11)	0.0150 (11)	0.0114 (11)	-0.0064 (9)

Geometric parameters (Å, °)

O1—C7	1.2187 (17)	C7—C8	1.474 (2)
O2—C1	1.364 (2)	C8—C9	1.3267 (19)
O2—C13	1.419 (2)	C8—H8A	0.9300
C1—C2	1.387 (2)	C9—C10	1.4616 (19)
C1—C6	1.396 (2)	C9—H9A	0.9300
C2—C3	1.378 (3)	C10—C12	1.3897 (19)
C2—H2A	0.9300	C10—C11	1.4002 (19)
C3—C4	1.370 (3)	C11—C12 ⁱ	1.3760 (19)
C3—H3A	0.9300	C11—H11A	0.9300
C4—C5	1.375 (2)	C12—C11 ⁱ	1.3761 (19)
C4—H4A	0.9300	C12—H12A	0.9300
C5—C6	1.387 (2)	C13—H13A	0.9600
C5—H5A	0.9300	C13—H13B	0.9600
C6—C7	1.4932 (19)	C13—H13C	0.9600
C1—O2—C13	118.66 (14)	C9—C8—C7	120.50 (13)
O2—C1—C2	124.22 (16)	C9—C8—H8A	119.8
O2—C1—C6	115.79 (13)	C7—C8—H8A	119.8
C2—C1—C6	119.87 (16)	C8—C9—C10	127.89 (13)
C3—C2—C1	119.34 (18)	C8—C9—H9A	116.1
C3—C2—H2A	120.3	C10—C9—H9A	116.1
C1—C2—H2A	120.3	C12—C10—C11	118.03 (13)
C4—C3—C2	121.72 (17)	C12—C10—C9	122.99 (12)
C4—C3—H3A	119.1	C11—C10—C9	118.97 (12)
C2—C3—H3A	119.1	C12 ⁱ —C11—C10	121.57 (12)
C3—C4—C5	118.73 (19)	C12 ⁱ —C11—H11A	119.2
C3—C4—H4A	120.6	C10—C11—H11A	119.2
C5—C4—H4A	120.6	C11 ⁱ —C12—C10	120.40 (12)
C4—C5—C6	121.43 (18)	C11 ⁱ —C12—H12A	119.8
C4—C5—H5A	119.3	C10—C12—H12A	119.8
C6—C5—H5A	119.3	O2—C13—H13A	109.5
C5—C6—C1	118.81 (14)	O2—C13—H13B	109.5
C5—C6—C7	117.92 (14)	H13A—C13—H13B	109.5
C1—C6—C7	123.27 (14)	O2—C13—H13C	109.5
O1—C7—C8	121.49 (13)	H13A—C13—H13C	109.5
O1—C7—C6	119.30 (13)	H13B—C13—H13C	109.5

C8—C7—C6	119.12 (12)		
C13—O2—C1—C2	-4.7 (2)	C5—C6—C7—O1	-36.6 (2)
C13—O2—C1—C6	179.21 (14)	C1—C6—C7—O1	143.60 (15)
O2—C1—C2—C3	-174.24 (17)	C5—C6—C7—C8	140.06 (15)
C6—C1—C2—C3	1.7 (3)	C1—C6—C7—C8	-39.8 (2)
C1—C2—C3—C4	0.8 (3)	O1—C7—C8—C9	-13.5 (2)
C2—C3—C4—C5	-1.5 (3)	C6—C7—C8—C9	169.90 (14)
C3—C4—C5—C6	-0.4 (3)	C7—C8—C9—C10	178.03 (14)
C4—C5—C6—C1	2.8 (2)	C8—C9—C10—C12	-13.8 (2)
C4—C5—C6—C7	-177.02 (15)	C8—C9—C10—C11	167.44 (15)
O2—C1—C6—C5	172.83 (13)	C12—C10—C11—C12 ⁱ	0.1 (2)
C2—C1—C6—C5	-3.4 (2)	C9—C10—C11—C12 ⁱ	178.90 (13)
O2—C1—C6—C7	-7.4 (2)	C11—C10—C12—C11 ⁱ	-0.1 (2)
C2—C1—C6—C7	176.38 (14)	C9—C10—C12—C11 ⁱ	-178.84 (13)

Symmetry code: (i) $-x, -y+1, -z$.

(II) (2*E*,2'*E*)-3,3'-(1,4-Phenylene)bis[1-(3-methoxyphenyl)prop-2-en-1-one]

Crystal data

$C_{26}H_{22}O_4$

$M_r = 398.43$

Monoclinic, $P2_1/c$

$a = 5.2626$ (5) Å

$b = 15.6157$ (14) Å

$c = 12.4824$ (11) Å

$\beta = 98.760$ (2)°

$V = 1013.83$ (16) Å³

$Z = 2$

$F(000) = 420$

$D_x = 1.305$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2940 reflections

$\theta = 2.6$ – 27.7°

$\mu = 0.09$ mm⁻¹

$T = 294$ K

Needle, yellow

$0.90 \times 0.48 \times 0.09$ mm

Data collection

Bruker APEXII DUO CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2012)

$T_{\min} = 0.874$, $T_{\max} = 0.992$

17791 measured reflections

2458 independent reflections

1574 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.043$

$\theta_{\max} = 28.1^\circ$, $\theta_{\min} = 2.1^\circ$

$h = -6 \rightarrow 6$

$k = -20 \rightarrow 20$

$l = -16 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.044$

$wR(F^2) = 0.132$

$S = 1.03$

2458 reflections

137 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0567P)^2 + 0.138P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.14$ e Å⁻³

$\Delta\rho_{\min} = -0.14$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	1.3466 (2)	0.50160 (7)	0.83740 (9)	0.0753 (4)
O2	2.0704 (2)	0.69588 (8)	1.00094 (10)	0.0771 (4)
C1	1.6959 (3)	0.63162 (9)	0.89251 (12)	0.0502 (4)
H1A	1.6962	0.5856	0.9399	0.060*
C2	1.8795 (3)	0.69465 (10)	0.91308 (12)	0.0552 (4)
C3	1.8786 (3)	0.76279 (11)	0.84322 (15)	0.0680 (5)
H3A	2.0034	0.8052	0.8572	0.082*
C4	1.6943 (3)	0.76859 (11)	0.75284 (16)	0.0723 (5)
H4A	1.6945	0.8150	0.7062	0.087*
C5	1.5086 (3)	0.70582 (10)	0.73089 (13)	0.0583 (4)
H5A	1.3841	0.7100	0.6697	0.070*
C6	1.5086 (3)	0.63679 (9)	0.80022 (11)	0.0461 (3)
C7	1.3200 (3)	0.56519 (9)	0.78002 (11)	0.0499 (4)
C8	1.1041 (3)	0.57169 (10)	0.69022 (12)	0.0525 (4)
H8A	1.0879	0.6207	0.6473	0.063*
C9	0.9345 (3)	0.51020 (9)	0.66922 (11)	0.0505 (4)
H9A	0.9579	0.4629	0.7149	0.061*
C10	0.7126 (2)	0.50652 (9)	0.58281 (10)	0.0450 (3)
C11	0.6491 (3)	0.57310 (9)	0.50998 (11)	0.0514 (4)
H11A	0.7478	0.6228	0.5161	0.062*
C12	0.5587 (3)	0.43363 (10)	0.57116 (11)	0.0522 (4)
H12A	0.5971	0.3884	0.6194	0.063*
C13	2.1228 (4)	0.61900 (12)	1.05960 (15)	0.0766 (6)
H13A	2.2657	0.6275	1.1163	0.115*
H13B	2.1640	0.5748	1.0117	0.115*
H13C	1.9743	0.6024	1.0908	0.115*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0691 (8)	0.0679 (7)	0.0765 (8)	−0.0179 (6)	−0.0289 (6)	0.0158 (6)
O2	0.0662 (8)	0.0728 (8)	0.0805 (8)	−0.0176 (6)	−0.0268 (6)	−0.0033 (6)
C1	0.0443 (8)	0.0523 (8)	0.0509 (8)	−0.0004 (6)	−0.0027 (6)	−0.0046 (6)
C2	0.0451 (8)	0.0557 (9)	0.0609 (9)	−0.0035 (7)	−0.0040 (7)	−0.0106 (7)
C3	0.0536 (10)	0.0573 (10)	0.0894 (12)	−0.0104 (8)	−0.0009 (9)	−0.0014 (9)
C4	0.0619 (11)	0.0631 (10)	0.0882 (13)	−0.0042 (8)	−0.0010 (9)	0.0156 (9)
C5	0.0494 (9)	0.0604 (9)	0.0612 (9)	0.0040 (7)	−0.0042 (7)	0.0039 (7)
C6	0.0369 (7)	0.0511 (8)	0.0480 (7)	0.0035 (6)	−0.0007 (6)	−0.0063 (6)
C7	0.0418 (8)	0.0558 (8)	0.0486 (8)	0.0010 (6)	−0.0044 (6)	−0.0040 (7)

C8	0.0435 (8)	0.0583 (8)	0.0513 (8)	0.0010 (7)	-0.0072 (6)	-0.0015 (6)
C9	0.0450 (8)	0.0561 (8)	0.0464 (7)	0.0026 (6)	-0.0063 (6)	-0.0029 (6)
C10	0.0366 (7)	0.0524 (8)	0.0432 (7)	0.0042 (6)	-0.0031 (6)	-0.0071 (6)
C11	0.0442 (8)	0.0511 (8)	0.0550 (8)	-0.0035 (6)	-0.0048 (7)	-0.0032 (6)
C12	0.0453 (8)	0.0549 (8)	0.0516 (8)	0.0015 (6)	-0.0076 (6)	0.0035 (6)
C13	0.0648 (11)	0.0864 (13)	0.0689 (11)	-0.0082 (9)	-0.0214 (9)	0.0006 (9)

Geometric parameters (Å, °)

O1—C7	1.2199 (17)	C7—C8	1.4729 (18)
O2—C2	1.3698 (17)	C8—C9	1.310 (2)
O2—C13	1.411 (2)	C8—H8A	0.9300
C1—C2	1.376 (2)	C9—C10	1.4652 (17)
C1—C6	1.3998 (18)	C9—H9A	0.9300
C1—H1A	0.9300	C10—C11	1.3882 (19)
C2—C3	1.375 (2)	C10—C12	1.392 (2)
C3—C4	1.374 (2)	C11—C12 ⁱ	1.3765 (18)
C3—H3A	0.9300	C11—H11A	0.9300
C4—C5	1.382 (2)	C12—C11 ⁱ	1.3766 (18)
C4—H4A	0.9300	C12—H12A	0.9300
C5—C6	1.382 (2)	C13—H13A	0.9600
C5—H5A	0.9300	C13—H13B	0.9600
C6—C7	1.491 (2)	C13—H13C	0.9600
C2—O2—C13	117.70 (12)	C9—C8—C7	121.65 (14)
C2—C1—C6	119.92 (14)	C9—C8—H8A	119.2
C2—C1—H1A	120.0	C7—C8—H8A	119.2
C6—C1—H1A	120.0	C8—C9—C10	128.20 (14)
O2—C2—C3	115.34 (13)	C8—C9—H9A	115.9
O2—C2—C1	124.61 (14)	C10—C9—H9A	115.9
C3—C2—C1	120.05 (14)	C11—C10—C12	117.76 (12)
C4—C3—C2	120.36 (15)	C11—C10—C9	122.57 (13)
C4—C3—H3A	119.8	C12—C10—C9	119.67 (13)
C2—C3—H3A	119.8	C12 ⁱ —C11—C10	120.65 (13)
C3—C4—C5	120.36 (16)	C12 ⁱ —C11—H11A	119.7
C3—C4—H4A	119.8	C10—C11—H11A	119.7
C5—C4—H4A	119.8	C11 ⁱ —C12—C10	121.59 (13)
C4—C5—C6	119.78 (14)	C11 ⁱ —C12—H12A	119.2
C4—C5—H5A	120.1	C10—C12—H12A	119.2
C6—C5—H5A	120.1	O2—C13—H13A	109.5
C5—C6—C1	119.53 (13)	O2—C13—H13B	109.5
C5—C6—C7	122.86 (12)	H13A—C13—H13B	109.5
C1—C6—C7	117.59 (13)	O2—C13—H13C	109.5
O1—C7—C8	120.62 (13)	H13A—C13—H13C	109.5
O1—C7—C6	119.81 (12)	H13B—C13—H13C	109.5
C8—C7—C6	119.57 (13)		
C13—O2—C2—C3	-164.12 (17)	C1—C6—C7—O1	-6.5 (2)

C13—O2—C2—C1	16.2 (2)	C5—C6—C7—C8	-7.6 (2)
C6—C1—C2—O2	179.81 (14)	C1—C6—C7—C8	173.97 (13)
C6—C1—C2—C3	0.2 (2)	O1—C7—C8—C9	0.2 (2)
O2—C2—C3—C4	-179.41 (16)	C6—C7—C8—C9	179.71 (14)
C1—C2—C3—C4	0.3 (3)	C7—C8—C9—C10	-179.20 (14)
C2—C3—C4—C5	-0.3 (3)	C8—C9—C10—C11	-1.7 (2)
C3—C4—C5—C6	0.0 (3)	C8—C9—C10—C12	177.64 (16)
C4—C5—C6—C1	0.5 (2)	C12—C10—C11—C12 ⁱ	-0.4 (2)
C4—C5—C6—C7	-177.93 (16)	C9—C10—C11—C12 ⁱ	178.90 (14)
C2—C1—C6—C5	-0.5 (2)	C11—C10—C12—C11 ⁱ	0.4 (2)
C2—C1—C6—C7	177.94 (14)	C9—C10—C12—C11 ⁱ	-178.91 (14)
C5—C6—C7—O1	171.91 (15)		

Symmetry code: (i) $-x+1, -y+1, -z+1$.

(III) (2*E*,2'*E*)-3,3'-(1,4-Phenylene)bis[1-(3,4-dimethoxyphenyl)prop-2-en-1-one]

Crystal data

C₂₈H₂₆O₆

$M_r = 458.49$

Monoclinic, $P2_1/n$

$a = 6.9595$ (6) Å

$b = 21.0272$ (17) Å

$c = 8.3297$ (7) Å

$\beta = 103.602$ (2)°

$V = 1184.77$ (17) Å³

$Z = 2$

$F(000) = 484$

$D_x = 1.285$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 3549 reflections

$\theta = 2.7$ – 28.5 °

$\mu = 0.09$ mm⁻¹

$T = 294$ K

Block, yellow

$0.35 \times 0.27 \times 0.16$ mm

Data collection

Bruker APEXII DUO CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2012)

$T_{\min} = 0.890$, $T_{\max} = 0.985$

12532 measured reflections

3165 independent reflections

2328 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.030$

$\theta_{\max} = 29.0$ °, $\theta_{\min} = 1.9$ °

$h = -9 \rightarrow 8$

$k = -28 \rightarrow 26$

$l = -11 \rightarrow 11$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.049$

$wR(F^2) = 0.136$

$S = 1.04$

3165 reflections

156 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0557P)^2 + 0.2808P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.20$ e Å⁻³

$\Delta\rho_{\min} = -0.16$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.26810 (16)	0.04480 (7)	0.05640 (13)	0.0667 (4)
O2	-0.37696 (15)	0.15896 (5)	0.01870 (14)	0.0560 (3)
O3	-0.33472 (18)	0.22511 (6)	0.28240 (17)	0.0694 (4)
C1	-0.04399 (19)	0.11677 (6)	0.10365 (16)	0.0377 (3)
H1A	-0.0610	0.0919	0.0090	0.045*
C2	-0.1959 (2)	0.15447 (6)	0.12601 (17)	0.0399 (3)
C3	-0.1726 (2)	0.19126 (7)	0.2696 (2)	0.0459 (3)
C4	0.0049 (2)	0.19044 (7)	0.3851 (2)	0.0525 (4)
H4A	0.0217	0.2154	0.4796	0.063*
C5	0.1585 (2)	0.15265 (7)	0.36105 (18)	0.0469 (4)
H5A	0.2779	0.1525	0.4398	0.056*
C6	0.13686 (19)	0.11510 (6)	0.22134 (16)	0.0369 (3)
C7	0.2924 (2)	0.07224 (7)	0.18910 (16)	0.0413 (3)
C8	0.4765 (2)	0.06233 (7)	0.31697 (16)	0.0419 (3)
H8A	0.4879	0.0794	0.4218	0.050*
C9	0.62488 (19)	0.02971 (7)	0.28517 (16)	0.0385 (3)
H9A	0.6062	0.0140	0.1782	0.046*
C10	0.81526 (18)	0.01514 (6)	0.39711 (15)	0.0348 (3)
C11	0.8724 (2)	0.03954 (7)	0.55600 (16)	0.0422 (3)
H11A	0.7872	0.0664	0.5947	0.051*
C12	0.9466 (2)	-0.02452 (7)	0.34276 (16)	0.0429 (3)
H12A	0.9117	-0.0412	0.2364	0.051*
C13	-0.4046 (3)	0.12540 (11)	-0.1316 (2)	0.0727 (6)
H13A	-0.5375	0.1316	-0.1950	0.109*
H13B	-0.3816	0.0809	-0.1092	0.109*
H13C	-0.3135	0.1409	-0.1929	0.109*
C14	-0.3309 (3)	0.25570 (12)	0.4350 (3)	0.0920 (8)
H14A	-0.4595	0.2727	0.4331	0.138*
H14B	-0.2360	0.2896	0.4516	0.138*
H14C	-0.2948	0.2255	0.5234	0.138*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0427 (6)	0.1049 (10)	0.0443 (6)	0.0272 (6)	-0.0066 (5)	-0.0257 (6)
O2	0.0389 (6)	0.0653 (7)	0.0565 (6)	0.0185 (5)	-0.0031 (5)	-0.0026 (5)
O3	0.0576 (7)	0.0680 (8)	0.0820 (9)	0.0234 (6)	0.0149 (6)	-0.0190 (7)
C1	0.0349 (7)	0.0413 (7)	0.0355 (6)	0.0040 (5)	0.0056 (5)	0.0013 (5)
C2	0.0344 (7)	0.0387 (7)	0.0443 (7)	0.0040 (5)	0.0048 (5)	0.0051 (6)

C3	0.0434 (8)	0.0380 (7)	0.0578 (9)	0.0059 (6)	0.0149 (7)	-0.0026 (6)
C4	0.0523 (9)	0.0492 (9)	0.0544 (9)	-0.0001 (7)	0.0092 (7)	-0.0169 (7)
C5	0.0378 (7)	0.0522 (8)	0.0469 (8)	-0.0014 (6)	0.0024 (6)	-0.0087 (7)
C6	0.0308 (6)	0.0414 (7)	0.0372 (6)	0.0006 (5)	0.0052 (5)	0.0015 (5)
C7	0.0306 (7)	0.0553 (8)	0.0357 (6)	0.0048 (6)	0.0030 (5)	-0.0022 (6)
C8	0.0327 (7)	0.0572 (8)	0.0325 (6)	0.0042 (6)	0.0010 (5)	-0.0027 (6)
C9	0.0303 (7)	0.0504 (8)	0.0316 (6)	0.0003 (5)	0.0010 (5)	0.0006 (5)
C10	0.0274 (6)	0.0443 (7)	0.0306 (6)	-0.0005 (5)	0.0026 (5)	0.0037 (5)
C11	0.0323 (7)	0.0559 (8)	0.0365 (7)	0.0091 (6)	0.0040 (5)	-0.0049 (6)
C12	0.0358 (7)	0.0574 (9)	0.0316 (6)	0.0054 (6)	0.0004 (5)	-0.0066 (6)
C13	0.0523 (11)	0.0953 (14)	0.0582 (10)	0.0179 (10)	-0.0117 (8)	-0.0150 (10)
C14	0.0722 (14)	0.0934 (16)	0.1191 (19)	0.0054 (12)	0.0398 (13)	-0.0499 (15)

Geometric parameters (Å, °)

O1—C7	1.2229 (17)	C8—C9	1.318 (2)
O2—C2	1.3660 (16)	C8—H8A	0.9300
O2—C13	1.410 (2)	C9—C10	1.4621 (16)
O3—C3	1.3593 (18)	C9—H9A	0.9300
O3—C14	1.419 (2)	C10—C11	1.3877 (18)
C1—C2	1.3688 (19)	C10—C12	1.3893 (19)
C1—C6	1.4022 (18)	C11—C12 ⁱ	1.3781 (18)
C1—H1A	0.9300	C11—H11A	0.9300
C2—C3	1.401 (2)	C12—C11 ⁱ	1.3780 (18)
C3—C4	1.376 (2)	C12—H12A	0.9300
C4—C5	1.384 (2)	C13—H13A	0.9600
C4—H4A	0.9300	C13—H13B	0.9600
C5—C6	1.3848 (19)	C13—H13C	0.9600
C5—H5A	0.9300	C14—H14A	0.9600
C6—C7	1.4807 (19)	C14—H14B	0.9600
C7—C8	1.4756 (17)	C14—H14C	0.9600
C2—O2—C13	117.30 (12)	C7—C8—H8A	119.4
C3—O3—C14	117.82 (15)	C8—C9—C10	128.07 (12)
C2—C1—C6	120.95 (13)	C8—C9—H9A	116.0
C2—C1—H1A	119.5	C10—C9—H9A	116.0
C6—C1—H1A	119.5	C11—C10—C12	118.04 (11)
O2—C2—C1	125.04 (13)	C11—C10—C9	122.96 (12)
O2—C2—C3	115.10 (12)	C12—C10—C9	119.00 (12)
C1—C2—C3	119.85 (12)	C12 ⁱ —C11—C10	120.96 (13)
O3—C3—C4	125.24 (14)	C12 ⁱ —C11—H11A	119.5
O3—C3—C2	115.11 (13)	C10—C11—H11A	119.5
C4—C3—C2	119.65 (13)	C11 ⁱ —C12—C10	121.00 (12)
C3—C4—C5	120.21 (14)	C11 ⁱ —C12—H12A	119.5
C3—C4—H4A	119.9	C10—C12—H12A	119.5
C5—C4—H4A	119.9	O2—C13—H13A	109.5
C4—C5—C6	120.91 (13)	O2—C13—H13B	109.5
C4—C5—H5A	119.5	H13A—C13—H13B	109.5

C6—C5—H5A	119.5	O2—C13—H13C	109.5
C5—C6—C1	118.41 (12)	H13A—C13—H13C	109.5
C5—C6—C7	124.03 (12)	H13B—C13—H13C	109.5
C1—C6—C7	117.55 (12)	O3—C14—H14A	109.5
O1—C7—C8	119.85 (12)	O3—C14—H14B	109.5
O1—C7—C6	119.98 (12)	H14A—C14—H14B	109.5
C8—C7—C6	120.17 (12)	O3—C14—H14C	109.5
C9—C8—C7	121.13 (12)	H14A—C14—H14C	109.5
C9—C8—H8A	119.4	H14B—C14—H14C	109.5
C13—O2—C2—C1	-4.5 (2)	C2—C1—C6—C5	-0.1 (2)
C13—O2—C2—C3	176.83 (15)	C2—C1—C6—C7	178.99 (13)
C6—C1—C2—O2	-179.73 (13)	C5—C6—C7—O1	-173.70 (15)
C6—C1—C2—C3	-1.1 (2)	C1—C6—C7—O1	7.3 (2)
C14—O3—C3—C4	-8.5 (3)	C5—C6—C7—C8	6.5 (2)
C14—O3—C3—C2	170.79 (17)	C1—C6—C7—C8	-172.50 (13)
O2—C2—C3—O3	1.14 (19)	O1—C7—C8—C9	7.3 (2)
C1—C2—C3—O3	-177.63 (13)	C6—C7—C8—C9	-172.95 (14)
O2—C2—C3—C4	-179.52 (14)	C7—C8—C9—C10	-179.46 (13)
C1—C2—C3—C4	1.7 (2)	C8—C9—C10—C11	-5.6 (2)
O3—C3—C4—C5	178.13 (15)	C8—C9—C10—C12	174.89 (15)
C2—C3—C4—C5	-1.1 (2)	C12—C10—C11—C12 ⁱ	-0.4 (2)
C3—C4—C5—C6	0.0 (2)	C9—C10—C11—C12 ⁱ	-179.97 (14)
C4—C5—C6—C1	0.7 (2)	C11—C10—C12—C11 ⁱ	0.4 (2)
C4—C5—C6—C7	-178.35 (14)	C9—C10—C12—C11 ⁱ	179.98 (13)

Symmetry code: (i) $-x+2, -y, -z+1$.