## UNIVERSITI SAINS MALAYSIA

Second Semester Examination 2010/2011 Academic Session

April/May 2011

## EKC 574 – Downstream Processing Of Biochemical and Pharmaceutical Products

Duration : 3 hours

Please ensure that this examination paper contains <u>FIVE</u> printed pages and <u>TWO</u> printed page of Appendix before you begin the examination.

Instruction: Answer ALL questions.

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- 1. [a] Based on the information below, suggest and briefly explain the most likely unit operations that should be used for the isolation and purification of Linomycin:
  - Antibiotic produced by Streptomyces lincolnesis;
  - Free base,
  - *pKa*' 7.6.
  - More stable in salt form.
  - Soluble in methanol, ethyl acetate, acetone, chloroform.
  - Slightly soluble in water.

[6 marks]

- [b] Ultrafiltration process to concentrate protease is to be carried out under continuous operation using modules that consists of tubes of 0.075 cm in diameter and 50 cm long. Knowing that protease has a diffusion coefficient of  $8.8 \times 10^{-7} \text{ cm}^2/\text{s}$ . The solution has a viscosity of 0.012 g /cm.s and a density of  $1.1 \text{ g/cm}^3$ . The bulk stream velocity is 300 cm/s. Based on these information,
  - [i] the thickness of concentration polarization layer
  - [ii] the polarization modulus of the fluid for membrane flux of 50  $l/(m^2.hr)$ .
  - [iii] the intrinsic rejection coefficient of the protease for the same membrane flux if the observed rejection coefficient is 95%.
  - [iv] suggest a method to improve the intrinsic rejection coefficient, justify your suggestion.

[19 marks]

2. [a] Fouling is a major obstacle for the wide spread use of membrane technology. Describe the reversible and irreversible fouling based on the attachment strength of particles to the membrane surface. How membrane fouling can be controlled?

[7 marks]

[18 marks]

[b] In a laboratory centrifuge, it takes a minimum of 17 min to clarify the cell lysate at 12000 rpm. After 17 min, the top of the culture being centrifuged was 32 mm and the top of the packed solids was 79 mm from the center of rotation, respectively.

Determine the maximum flow rate for the clarification of a suspension of lysed *Escherichia coli* cells in a plant scale disk centrifuge. Given that the plant centrifuge has a bowl diameter of 25.4 cm, the maximum dimensionless acceleration, *G* of 14200,  $\theta = 42^{\circ}$ ,  $R_1 = 8$  cm,  $R_0 = 20$  cm and number of disks = 100.

Information:

3. [a] A 0.6 m length and 7.5 x  $10^{-3}$  m inner diameter high performance liquid chromatography (HPLC) column is used for the separation of two proteins (bovine serum albumin (BSA) and myoglobin (Mb) in a 0.1 x  $10^{-6}$  m<sup>3</sup> solution. Pilot runs of HPLC at flow velocity of 2.17 x  $10^{-4}$  m/s indicate that the capacity factors  $K_1$  and  $K_2$  of two protein are 0.8 and 1.1 respectively. The relationship between the heights equivalent to a theoretical plate (HETP) and the flow velocity is obtained as follows,

HETP (m) =  $6.2 \times 10^{-5}$  (m) + 1.13 (s)*u* 

Where u = the flow velocity (ms<sup>-1</sup>)

- [i] Based on the above information, calculate the number of theoretical plates of the HPLC column.
- [ii] What resolution  $R_N$  of two enzymes may be expected? Comments on the result obtained.
- [iii] Calculate the flow rate and sample volume needed to separate the above two enzymes if the column length and inside diameter of a large-scale HPLC are 0.5 m and 0.108 m, respectively

[12 marks]

[b] The equilibrium partitioning of a protein between an organic solvent extract phase and aqueous feed phase has been found to be nonlinear and can be represented by the following equation:

 $y = 1.50 \ln x + 6.0$ 

where y and x are concentrations of the protein in the extract and aqueous feed phases, respectively in g/l. It is desired to extract 96% of the protein from a feed stream having a protein concentration of 4.0 g/l. For a feed stream at a flow rate of 5.5 liters/min and an extract stream at a flow rate of 3.0 liters/min

- [i] What is the concentration of the peptide in the exit extract stream.
- [ii] Graphically estimate how many equilibrium stages will be required for countercurrent flow of the phases
- [iii] As the concentration of the peptide in the raffinate decreases, does the partitioning of the protein become more or less favourable?

[13 marks]

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4. [a] Crystallization is usually the last step in producing highly purified products such as antibiotics. This process is said to be similar with precipitation but there are some differences between these two processes. Briefly discuss their differences.

[7 marks]

[b] It is desired to filter a cell broth at a rate of 2000 l/h on a rotary vacuum filter at a vacuum pressure of 70 kPa. The cycle time for the drum will be 60 s and the cake formation time (filtering time) will be 15 s. The broth to be filtered has a viscosity of 2.0 cP and a cake solid (dry basis) per volume of filtrate of 10 g/l. From laboratory tests, the specific cake resistance has been determined to be 9 x  $10^{10}$  cm/g. Determine the area of the filter that is required. The resistance of the filter can be neglected.

It is then desired to wash the product (antibiotic) out of the cake so that only 5% of the antibiotic in the cake is left after washing. The washing efficiency is 50%. Estimate the washing time per cycle that would be required. Assume that the filtered cake contains 70% water.

[18 marks]

Information:

$\frac{At}{V} = \frac{\mu \alpha \rho_0}{2\Delta P} \left( \frac{V}{A} \right) + \frac{\mu R_M}{\Delta P}$		$\alpha = \alpha' (\Delta P)^s$
$R' = \left(1 - \frac{E}{100}\right)^n$	$\frac{t_w}{t_f} = 2nf$	$\frac{t_w}{t_f} = \frac{2V_w}{V_f}$

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Appendix

$$\frac{Graetz-Leveque \ correlation}{Sh=1.62 \ Re^{0.33}Sc^{0.33} \left(\frac{d}{l}\right)^{0.33}}$$

 $\frac{Dittus-Boelter\ correlation}{Sh = 0.023\ Re^{0.8}Sc^{0.33}}$ 

 $\frac{Column Resolution}{1 - (s - 1)(-k - )}$ 

$$R_{N} = \frac{1}{4} \sqrt{N_{P}} \left( \frac{\delta - 1}{\delta} \right) \left( \frac{k_{2}}{k_{2} + 1} \right)$$

Length	Volume		
1 ft = 12 in = 0.3048 m, 1 yard = 3 ft	$1 \ {\rm ft}^3 = 0.028317 \ {\rm m}^3 = 7.481 \ {\rm gal}$ , $1 \ {\rm bbl} = 42 \ {\rm U.S. \ gal}$		
1 mi = 5280 ft = 1609.344 m	1 U.S. gal = 231 in <sup>3</sup> = $3.7853 \text{ L} = 4qt = 0.833 \text{ lmp.gal}$ .		
1 nautical mile (nmi) = $6076$ ft	$1 L = 0.001 m^3 = 0.035315 ft^3 = 0.2642 U.S. gal$		
Mass	Density		
$1 \text{ slug} = 32.174 \text{ lb}_{m} = 14.594 \text{ kg}$	$1 \text{ slug/ft}^3 = 515.38 \text{ kg/m}^3$ , $1 \text{ g/cm}^3 = 1000 \text{ kg/m}^3$		
$1 \text{ lb}_{\text{m}} = 0.4536 \text{ kg} = 7000 \text{ grains}$	$1 \text{ lb}_{\text{m}}/\text{ft}^3 = 16.0185 \text{ kg/m}^3$ , $1 \text{ lb}_{\text{m}}/\text{in}^3 = 27.68 \text{ g/cm}^3$		
Acceleration & Area	Velocity		
$1 \text{ ft/s}^2 = 0.3048 \text{ m/s}^2$	1 ft/s = 0.3048 m/s , 1 knot = 1 min/h = 1.6878 ft/s		
$1 \text{ ft}^2 = 0.092903 \text{ m}^2$	1  min/h = 1.46666666  ft/s (fps) = 0.44704  m/s		
Mass Flow & Mass Flux	Volume Flow		
$1 \text{ slug/s} = 14.594 \text{ kg/s}$ . $1 \text{ lb}_{\text{m}}/\text{s} = 0.4536 \text{ kg/s}$	1 gal/min = $0.00228 \text{ ft}^3/\text{s} = 0.06309 \text{ L/s}$		
$1 \text{ kg/m}^2 \text{s} = 0.2046 \text{ lb}_{\text{m}}/\text{ft}^2 \text{s}$	1 million gal/day = $1.5472 \text{ ft}^3/\text{s} = 0.04381 \text{ m}^3/\text{s}$		
$= 0.00636 \text{ slug/ft}^2 \text{s}$			
Pressure	Force and Surface Tension		
Pressure $1 \text{ lb}_{f}/\text{ft}^{3} = 47.88 \text{ Pa}, 1 \text{ torr} = 1 \text{ mm Hg}$	Force and Surface Tension $1 \text{ lb}_{f} = 4.448222 \text{ N} = 16 \text{ oz}, 1 \text{ dyne} = 1 \text{ g cm/s}^{2} = 10^{-5} \text{N}$		
Pressure $1 \text{ lb}_{f}/\text{ft}^{3} = 47.88 \text{ Pa}, 1 \text{ torr} = 1 \text{ mm Hg}$ $1 \text{ psi} = 144 \text{ psf}, 1 \text{ bar} = 10^{5} \text{ Pa}$	Force and Surface Tension 1 $lb_f = 4.448222 N = 16 \text{ oz}, 1 \text{ dyne} = 1 \text{ g cm/s}^2 = 10^{-5} \text{N}$ 1 $kg_f = 2.2046 lb_f = 9.80665 N$		
Pressure           1 $lb_f/ft^3 = 47.88$ Pa, 1 torr = 1 mm Hg           1 psi = 144 psf, 1 bar = $10^5$ Pa           1 atm = 2116.2 psf = 14696 psi = 101,325 Pa	$\label{eq:force} \hline {\mbox{Force and Surface Tension}} \\ 1 \ lb_f = 4.448222 \ N = 16 \ oz, \ 1 \ dyne = 1 \ g \ cm/s^2 = 10^{-5} N \\ 1 \ kg_f = 2.2046 \ lb_f = 9.80665 \ N \\ 1 \ U.S. \ (short) \ ton = 2000 \ lb_f \ , \ 1 \ N = 0.2248 \ lb_f \\ \hline \end{tabular}$		
Pressure           1 $lb_f/ft^3 = 47.88$ Pa, 1 torr = 1 mm Hg           1 psi = 144 psf, 1 bar = 10 <sup>5</sup> Pa           1 atm = 2116.2 psf = 14696 psi = 101,325 Pa           = 29.9 in Hg = 33.9 ft H <sub>2</sub> O	$\label{eq:spectral_states} \begin{array}{l} \mbox{Force and Surface Tension} \\ 1 \ lb_f = 4.448222 \ N = 16 \ oz, 1 \ dyne = 1 \ g \ cm/s^2 = 10^{.5} N \\ 1 \ kg_f = 2.2046 \ lb_f = 9.80665 \ N \\ 1 \ U.S. \ (short) \ ton = 2000 \ lb_f \ , 1 \ N = 0.2248 \ lb_f \\ 1 \ N/m = 0.0685 \ lb_f/ft \end{array}$		
Pressure           1 lb <sub>f</sub> /ft <sup>3</sup> = 47.88 Pa, 1 torr = 1 mm Hg           1 psi = 144 psf, 1 bar = $10^5$ Pa           1 atm = 2116.2 psf = 14696 psi = 101,325 Pa           = 29.9 in Hg = 33.9 ft H <sub>2</sub> O           Power	$\label{eq:spectral_state} \begin{array}{l} \mbox{Force and Surface Tension} \\ 1 \ lb_f = 4.448222 \ N = 16 \ oz, 1 \ dyne = 1 \ g \ cm/s^2 = 10^{-5}N \\ 1 \ kg_f = 2.2046 \ lb_f = 9.80665 \ N \\ 1 \ U.S. \ (short) \ ton = 2000 \ lb_f \ , 1 \ N = 0.2248 \ lb_f \\ 1 \ N/m = 0.0685 \ lb_f/ft \\ \hline \mbox{Energy and Specific Energy} \end{array}$		
Pressure           1 lb <sub>f</sub> /ft <sup>3</sup> = 47.88 Pa, 1 torr = 1 mm Hg           1 psi = 144 psf, 1 bar = $10^5$ Pa           1 atm = 2116.2 psf = 14696 psi = 101,325 Pa           = 29.9 in Hg = 33.9 ft H <sub>2</sub> O           Power           1 hp = 550 (ft.lb <sub>f</sub> )/s = 745.7 W	$\label{eq:spectral_states} \begin{array}{l} \mbox{Force and Surface Tension} \\ 1 \ lb_f = 4.448222 \ N = 16 \ oz, 1 \ dyne = 1 \ g \ cm/s^2 = 10^{-5}N \\ 1 \ kg_f = 2.2046 \ lb_f = 9.80665 \ N \\ 1 \ U.S. \ (short) \ ton = 2000 \ lb_f \ , 1 \ N = 0.2248 \ lb_f \\ 1 \ N/m = 0.0685 \ lb_f/ft \\ \hline \mbox{Energy and Specific Energy} \\ 1 \ ft \ lb_f = 1.35582 \ J, 1 \ hp\cdoth = 2544.5 \ Btu \end{array}$		
Pressure           1 lb <sub>f</sub> /ft <sup>3</sup> = 47.88 Pa, 1 torr = 1 mm Hg           1 psi = 144 psf, 1 bar = $10^5$ Pa           1 atm = 2116.2 psf = 14696 psi = 101,325 Pa           = 29.9 in Hg = 33.9 ft H <sub>2</sub> O           Power           1 hp = 550 (ft.lb <sub>f</sub> )/s = 745.7 W           1 (ft.lb <sub>f</sub> )/s = 1.3558 W	$\label{eq:spectral_states} \begin{array}{l} \mbox{Force and Surface Tension} \\ 1 \ lb_f = 4.448222 \ N = 16 \ oz, 1 \ dyne = 1 \ g \ cm/s^2 = 10^{-5} N \\ 1 \ kg_f = 2.2046 \ lb_f = 9.80665 \ N \\ 1 \ U.S. \ (short) \ ton = 2000 \ lb_f \ , 1 \ N = 0.2248 \ lb_f \\ 1 \ N/m = 0.0685 \ lb_f/ft \\ \hline \mbox{Energy and Specific Energy} \\ 1 \ ft \ lb_f = 1.35582 \ J, 1 \ hp \cdot h = 2544.5 \ Btu \\ 1 \ Btu = 252 \ cal = 1055.056 \ J = 778.17 \ ft \ lb_f \end{array}$		
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Pressure         1 $lb_f/ft^3 = 47.88$ Pa, 1 torr = 1 mm Hg         1 psi = 144 psf, 1 bar = 10 <sup>5</sup> Pa         1 atm = 2116.2 psf = 14696 psi = 101,325 Pa         = 29.9 in Hg = 33.9 ft H <sub>2</sub> O         Power         1 hp = 550 (ft.lb <sub>f</sub> )/s = 745.7 W         1 (ft.lb <sub>f</sub> )/s = 1.3558 W         1 Watt = 3.4123 Btu/h = 0.00134 hp         Specific Weight         1 lb <sub>f</sub> /ft <sup>3</sup> = 157.09 N/m <sup>3</sup>	$\begin{tabular}{ c c c c } \hline Force and Surface Tension \\ \hline I \ lb_f = 4.448222 \ N = 16 \ oz, 1 \ dyne = 1 \ g \ cm/s^2 = 10^{-5}N \\ \hline 1 \ kg_f = 2.2046 \ lb_f = 9.80665 \ N \\ \hline 1 \ U.S. \ (short) \ ton = 2000 \ lb_f \ , 1 \ N = 0.2248 \ lb_f \\ \hline 1 \ N/m = 0.0685 \ lb_f/ft \\ \hline \hline Energy \ and \ Specific \ Energy \\ \hline 1 \ ft \ lb_f = 1.35582 \ J, 1 \ hp \cdot h = 2544.5 \ Btu \\ \hline 1 \ Btu = 252 \ cal = 1055.056 \ J = 778.17 \ ft \ lb_f \\ \hline 1 \ cal = 4.1855 \ J, 1 \ ft.lb_f/lb_m = 2.9890 \ J/kg \\ \hline \hline Heat \ Flux \\ \hline 1 \ W/m^2 = 0.3171 \ Btu/(h \ ft^2) \end{tabular}$		
Pressure           1 $lb_f/ft^3 = 47.88$ Pa, 1 torr = 1 mm Hg           1 psi = 144 psf, 1 bar = 10 <sup>5</sup> Pa           1 atm = 2116.2 psf = 14696 psi = 101,325 Pa           = 29.9 in Hg = 33.9 ft H <sub>2</sub> O           Power           1 hp = 550 (ft.lb <sub>f</sub> )/s = 745.7 W           1 (ft.lb <sub>f</sub> )/s = 1.3558 W           1 Watt = 3.4123 Btu/h = 0.00134 hp           Specific Weight           1 lb <sub>f</sub> /ft <sup>3</sup> = 157.09 N/m <sup>3</sup>	Force and Surface Tension         1 $lb_f = 4.448222 N = 16 \text{ oz}, 1 \text{ dyne} = 1 \text{ g cm/s}^2 = 10^{-5}N$ 1 $kg_f = 2.2046 \ lb_f = 9.80665 N$ 1 U.S. (short) ton = 2000 $lb_f$ , 1 N = 0.2248 $lb_f$ 1 N/m = 0.0685 $lb_f/ft$ Energy and Specific Energy         1 ft $lb_f = 1.35582 J$ , 1 $hp \cdot h = 2544.5 Btu$ 1 Btu = 252 cal = 1055.056 J = 778.17 ft $lb_f$ 1 cal = 4.1855 J, 1 ft. $lb_f/lb_m = 2.9890 J/kg$ Heat Flux         1 W/m <sup>2</sup> = 0.3171 Btu/(h ft <sup>2</sup> )         Kinematic Viscosity		
Pressure           1 $lb_f/ft^3 = 47.88$ Pa, 1 torr = 1 mm Hg           1 psi = 144 psf, 1 bar = 10 <sup>5</sup> Pa           1 atm = 2116.2 psf = 14696 psi = 101,325 Pa           = 29.9 in Hg = 33.9 ft H <sub>2</sub> O           Power           1 hp = 550 (ft.lb <sub>f</sub> )/s = 745.7 W           1 (ft.lb <sub>f</sub> )/s = 1.3558 W           1 Watt = 3.4123 Btu/h = 0.00134 hp           Specific Weight           1 lb <sub>f</sub> /ft <sup>3</sup> = 157.09 N/m <sup>3</sup> Viscosity           1 slug/(ft.s) = 47.88 kg/(m.s) = 478.8 poise (p)	Force and Surface Tension1 lb <sub>f</sub> = 4.448222 N = 16 oz, 1 dyne = 1 g cm/s <sup>2</sup> = 10 <sup>-5</sup> N1 kg <sub>f</sub> = 2.2046 lb <sub>f</sub> = 9.80665 N1 U.S. (short) ton = 2000 lb <sub>f</sub> , 1 N = 0.2248 lb <sub>f</sub> 1 N/m = 0.0685 lb <sub>f</sub> /ftEnergy and Specific Energy1 ft lb <sub>f</sub> = 1.35582 J, 1 hp·h = 2544.5 Btu1 Btu = 252 cal = 1055.056 J = 778.17 ft lb <sub>f</sub> 1 cal = 4.1855 J, 1 ft.lb <sub>f</sub> /lb <sub>m</sub> = 2.9890 J/kgHeat Flux1 W/m <sup>2</sup> = 0.3171 Btu/(h ft <sup>2</sup> )Kinematic Viscosity1 ft <sup>2</sup> /h = 2.506 .10 <sup>-5</sup> m <sup>2</sup> /s, 1 ft <sup>2</sup> /s = 0.092903 m <sup>2</sup> /s		
Pressure         1 $lb_f/ft^3 = 47.88$ Pa, 1 torr = 1 mm Hg         1 psi = 144 psf, 1 bar = 10 <sup>5</sup> Pa         1 atm = 2116.2 psf = 14696 psi = 101,325 Pa         = 29.9 in Hg = 33.9 ft H <sub>2</sub> O         Power         1 hp = 550 (ft.lb <sub>f</sub> )/s = 745.7 W         1 (ft.lb <sub>f</sub> )/s = 1.3558 W         1 Watt = 3.4123 Btu/h = 0.00134 hp         Specific Weight         1 lb <sub>f</sub> /ft <sup>3</sup> = 157.09 N/m <sup>3</sup> Viscosity         1 slug/(ft.s) = 47.88 kg/(m.s) = 478.8 poise (p)         1 p=1 g/(cm.s) 0.1 kg/(m.s) = 0.002088 slug/(ft s)	Force and Surface Tension1 lb <sub>f</sub> = 4.448222 N = 16 oz, 1 dyne = 1 g cm/s <sup>2</sup> = 10 <sup>-5</sup> N1 kg <sub>f</sub> = 2.2046 lb <sub>f</sub> = 9.80665 N1 U.S. (short) ton = 2000 lb <sub>f</sub> , 1 N = 0.2248 lb <sub>f</sub> 1 N/m = 0.0685 lb <sub>f</sub> /ftEnergy and Specific Energy1 ft lb <sub>f</sub> = 1.35582 J, 1 hp·h = 2544.5 Btu1 Btu = 252 cal = 1055.056 J = 778.17 ft lb <sub>f</sub> 1 cal = 4.1855 J, 1 ft.lb <sub>f</sub> /lb <sub>m</sub> = 2.9890 J/kgHeat Flux1 W/m <sup>2</sup> = 0.3171 Btu/(h ft <sup>2</sup> )Kinematic Viscosity1 ft <sup>2</sup> /h = 2.506 .10 <sup>-5</sup> m <sup>2</sup> /s, 1 ft <sup>2</sup> /s = 0.092903 m <sup>2</sup> /s1 stoke (st) = 1 cm <sup>2</sup> /s = 0.0001 m <sup>2</sup> /s = 0.001076 ft <sup>2</sup> /s		

## **Common Engineering Conversion Factors**

Temperature Scale Readings			
$^{\circ}F = (9/5)^{\circ}C + 32$	$^{\circ}C = (5/9) (^{\circ}F - 32)$	${}^{o}R = {}^{o}F + 459.69$	${}^{\rm o}{\rm K} = {}^{\rm o}{\rm C} + 273.16$
Thermal (	Conductivity*	Gas Constant*	
1  cal/(s.cm.°C) = 242  Btu	/(h.ft. <sup>o</sup> R)	$R = 82.057 \text{ atm.cm}^{3}/(\text{gmol})$	.K) = 62.361 mm Hg.L/(gmol.K)
1 Btu/(h.ft.°R) = 1.7307 W/(m.K)		$= 1.134 \text{ atm.ft}^{3}/(\text{lbmol.K}) = 0.083144 \text{ bar.L}/(\text{gmol.K})$	
		$= 10.73 \text{ psi. ft}^{3}/(\text{lbmol.}^{\circ})$	R) = 555.0 mm Hg.ft <sup>3</sup> /(lbmol. $^{\circ}$ R)

• Note that the intervals in absolute (Kelvin) and  $^{\circ}C$  are equal. Also, 1  $^{\circ}R = 1 {}^{\circ}F$ .

Latent heat:  $1 \text{ J/kg} = 4.2995 \times 10^{-4} \text{ Btu/lb}_m = 10.76 \text{ lb}_f.\text{ft/slug} = 0.3345 \text{ lb}_f.\text{ft/lb}_m$ ,  $1 \text{ Btu/lb}_m = 2325.9 \text{ J/kg}$ 

Heat transfer coefficient: 1 Btu/(h.ft<sup>2.o</sup>F) =  $5.6782 \text{ W/(m^2.oC)}$ .

Heat generation rate:  $1 \text{ W/m}^3 = 0.09665 \text{ Btu/(h ft}^3)$ 

Heat transfer per unit length: 1 W/m = 1.0403 Btu/(h ft)

Mass transfer coefficient: 1 m/s = 11.811 ft/h, 1  $lb_{mol}/(h.ft^2) = 0.013562 \text{ kgmol}/(s.m^2)$