

UNIVERSITI SAINS MALAYSIA

Peperiksaan Semester Pertama  
Sidang Akademik 1997/98

September 1997

**EKC 450 Simulasi dan Pengoptimuman Proses**

Masa: [3 jam]

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**ARAHAN KEPADA CALON:**

Sila pastikan soalan peperiksaan ini mengandungi **LAPAN (8)** mukasurat bercetak dan **TIGA (3)** Lampiran sebelum anda memulakan peperiksaan.

Kertas soalan ini mengandungi **LIMA (5)** soalan.

Jawab mana-mana **EMPAT (4)** soalan.

Satu soalan **MESTI** dijawab dalam Bahasa Malaysia.

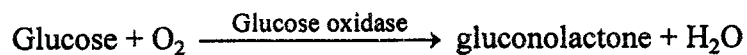
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1. Penukaran glukosa kepada asid glukonik adalah melalui tindakbalas pengoksidaan kumpulan aldehyd gula tersebut kepada kumpulan karboksil. Transformasi itu boleh dicapai dengan proses fermentasi menggunakan mikroorganisma. Kehadiran enzim glucose oxidase di dalam mikroorganisma menukarkan glukosa ke glukonolakton. Seterusnya, glukonolakton dihidrolisiskan kepada asid glukonik. Mekanisma keseluruhan proses fermentasi ini dilakukan dengan transformasi yang dikenali sebagai pertumbuhan sel.

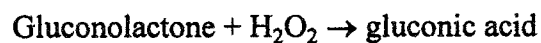
*The conversion of glucose to gluconic acid is a simple oxidation of the aldehyde group of the sugar to a carboxyl group. This transformation can be achieved by a microorganism in a fermentation process. The enzyme glucose oxidase, present in the microorganism, converts glucose to gluconolactone. In turn, the gluconolactone hydrolyzes to form the gluconic acid. The overall mechanism of the fermentation process which performs this transformation can be described as cell growth.*



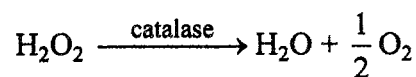
Pengoksidaan glukosa *Glucose oxidation:*



Hidrolisis glukonolakton *Gluconolactone hydrolysis:*



Penguraian peroksida *Peroxide decomposition:*



Model matematik fermentasi menggunakan bakteria *pseudomonas ovadis* di mana ia mengeluarkan asid glukonik telah dibentuk. Model ini menerangkan fasa tumbesaran logaritma secara dinamik yang boleh dirumuskan seperti di bawah:

*A mathematical model of the fermentation using the bacterium pseudomonas ovadis, which produces gluconic acid, has been developed. This model, which describes the dynamics of the logarithmic growth phase, can be summarised as follows:*

Kadar pertumbuhan sel  
*Rate of cell growth*

$$\frac{dy_1}{dt} = b_1 y_1 \left( 1.0 - \frac{y_1}{b_2} \right)$$

Kadar pembentukan glukonolakton:  
*Rate of gluconolactone formation:*

$$\frac{dy_2}{dt} = \frac{b_3 y_1 y_4}{b_4 + y_4} - 0.9082 b_5 y_2$$

Kadar pembentukan asam glukonik:  
*Rate of gluconic acid formation:*

$$\frac{dy_3}{dt} = b_5 y_2$$

Kadar penggunaan glukosa:  
*Rate of glucose consumption:*

$$\frac{dy_4}{dt} = -1.011 \left( \frac{b_3 y_1 y_4}{b_4 + y_4} \right)$$

di mana *where*  $y_1$  = kepekatan sel *concentration of cell*

$y_2$  = kepekatan glukonolakton *concentration of gluconolactone*

$y_3$  = kepekatan asam glukonik *concentration of gluconic acid*

$y_4$  = kepekatan glukosa *concentration of glucose*

$b_1 - b_5$  = sistem parameter berfungsi suhu dan pH.

*parameters of the system which are function of temperature and pH.*

Pada keadaan operasi 30°C dan pH 6.6, nilai-nilai untuk kelima-lima parameter ditentukan dari data eksperimen seperti berikut:

*At the operating conditions of 30°C and pH 6.6, the values of the five parameters were determined from experimental data to be:*

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-4-

$$b_1 = 0.949$$

$$b_2 = 3.439$$

$$b_3 = 18.72$$

$$b_4 = 37.51$$

$$b_5 = 1.167$$

Pada keadaan ini, binakan profil masa untuk kesemua pembolehubah  $y_1, y_2, y_3, y_4$  untuk julat masa  $0 \leq t \leq 3\text{hr}$ .

*At these conditions, develop the time profile of all variable,  $y_1, y_2, y_3, y_4$  for the period  $0 \leq t \leq 3\text{hr}$ .*

Pada permulaan jangkamasa itu, nilai-nilainya adalah

*At the begining of this period the values were*

$$y_1(0) = 0.5 \text{ U.O.D/mL} \quad y_2(0) = 0.0 \text{ mg/mL}$$

$$y_3(0) = 0.0 \text{ mg/mL} \quad y_4(0) = 50.0 \text{ mg/mL}$$

Gunakan 1 jam sebagai saiz langkah anda. Selesaikan masalah tersebut dengan menunjukkan penyelesaian secara terperinci.

*Take a step size of 1 hour. Solve your problem showing all details of solution.*

(25 markah)

2. Satu sistem reaktor tangki teraduk berterusan (CSTR) terdiri daripada M buah tangki bersiri seperti yang ditunjukkan dalam Rajah Q2. Larutan Q ( $\text{m}^3/\text{jam}$ ) disuapkan ke dalam setiap tangki. Suapan ke dalam tangki pertama mengandungi kepekatan komponen A,  $C_0$  mol /  $\text{m}^3$ . Setiap tangki diaduk dengan sempurna. Tindakbalas tak berbalik tertib pertama  $A \rightarrow B$  telah berlaku di dalam setiap tangki. Kadar tindakbalas A telah diberi seperti ungkapan

$-\frac{dW_n}{dt} = (kVC)_n$  mol / jam. Jika pemalar kadar tindakbalas k bagi kesemua tangki adalah sama, tunjukkan bahawa kepekatan B yang tertinggi boleh dihasilkan dengan sistem tangki ini bila  $V_1 = V_2 = \dots = V_M$ , dengan syarat

isipadu keseluruhan  $V = \sum_{n=1}^{n=M} V_n$  adalah malar.

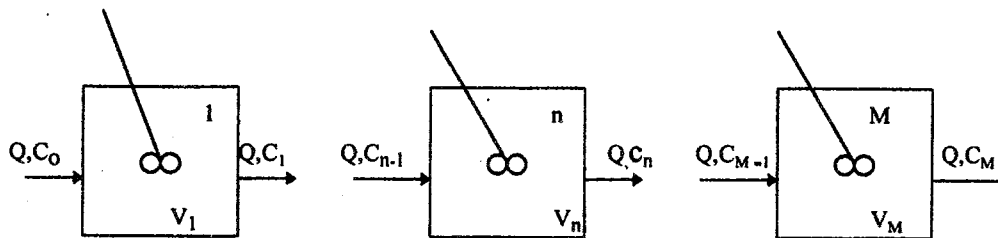
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A continuous stirred-tank reactor system consists of  $M$  tanks in series, as shown in Figure Q2.  $Q$  ( $m^3/h$ ) of solution is fed to each tank. The feed to the first tank contains a concentration of  $C_0$  moles/ $m^3$  of component A. Each tank is well stirred. Inside the tanks, the irreversible first-order reaction  $A \rightarrow B$  takes place. The rate at which A reacts is given by the expression.

$$-\frac{dW_n}{dt} = (kVC)_n \text{ moles / hr}$$

If the reaction-rate constant  $k$  is the same for each tank, show that the highest concentration of B which may be produced by the tank system occurs when

$V_1 = V_2 = \dots = V_M$  provided that the total volume  $V = \sum_{n=1}^{n=M} V_n$  remains constant.



Rajah Q2: Peringkat reaktor tangki teraduk  
Figure Q2: Stirred-tank reactor stages

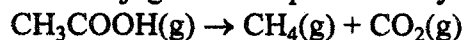
(25 markah)

3. Asid asetik dipecahkan di dalam relau untuk menghasilkan ketene melalui tindakbalas ini.

Acetic acid is cracked in a furnace to produce the intermediate ketene via the reaction



Tindakbalas ini juga berlaku pada kadar yang berpatutan.



The reaction  $\text{CH}_3\text{COOH}(\text{g}) \rightarrow \text{CH}_4(\text{g}) + \text{CO}_2(\text{g})$  also occurs to an appreciable extent.

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Proses ini perlu dipecahkan pada suhu  $T$  K dengan penukaran 80% dan pecahan hasil ketene ialah 0.0722.

*It is desired to carry out cracking at temperature  $T$  K with a conversion of 80% and a fractional yield of ketene of 0.0722.*

Sekiranya asid asetik disuapkan ke dalam reaktor pada  $300^\circ\text{C}$  dan tekanan 1 Bar.

*If acetic acid is fed to the reactor at  $300^\circ\text{C}$  and 1 Bar.*

- [a] Tunjukkan persamaan kepekatan untuk proses ini dan tunjukkan bagaimana kadar pemanasan yang diperlukan oleh relau boleh dikira untuk suapan 100 kg mol/jam asid.

*Show the concentration equations to this process, and also show how the required furnace heating rate can be calculated for a feed of 100 kg mol/h acid.*

- [b] Tunjukkan rajah aliran maklumat sistem ini.

*Show your information flow diagram of the system.*

Data:  $C_p = a + bT + cT^2 + dT^3$

	a	b	c	d
$\text{CH}_2\text{CO}$	4.11	$2.966 \times 10^{-2}$	$-1.79 \times 10^{-3}$	$4.22 \times 10^{-9}$
$\text{CH}_3\text{COOH}$	-3.61	$6.05 \times 10^{-1}$	$-3.94 \times 10^{-4}$	$-5.62 \times 10^{-7}$
$\text{CO}_2$	11.110	1.16	$-7.23 \times 10^{-3}$	$1.55 \times 10^{-5}$
$\text{CH}_4$	-5.71	1.03	$-1.67 \times 10^{-3}$	$1.96 \times 10^{-5}$
$\text{H}_2\text{O}$	18.3	0.47	$-1.34 \times 10^{-3}$	$1.32 \times 10^{-6}$

(25 markah)

4. Satu reaktor menukarkan sebatian organik kepada hasil P dengan bahan tambah (additive) A (pecahan mol =  $x_A$ ). Bahan tambah A boleh disuntik ke dalam reaktor, dan stim boleh disuntik ke dalam lingkaran pemanas di dalam reaktor tersebut dengan tujuan pembekalan haba.

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A reactor converts an organic compound to product P by heating the material in the presence of an additive A (mole fraction =  $x_A$ ). The additive can be injected into the reactor, while steam can be injected into a heating coil inside the reactor to provide heat.

Hasil P dijual pada harga RM50.00 per lb-mol. Untuk 1 lb-mol suapan, kos untuk bahan tambahan A (dalam RM per lb-mol) sebagai fungsi  $x_A$  diberi sebagai formula ini:

The product P can be sold for RM50.00 per lb-mol. For 1 lb-mol of feed, the cost of the additive (in RM per lb-mol) as a function of  $x_A$  is given by the formula:

$$0.1 + 0.5x_A + x_A^2$$

Kos wap (dalam RM) sebagai fungsi S adalah  
The cost of the steam (in RM) as a function of S is

$$0.2 + 0.035S + 0.01S^2.$$

(S = paun wap/paun-mol suapan)

(S = lb steam/lb-mol feed)

Persamaan hasil ialah  $y_p = 0.1 + 0.3x_A + 0.02S + 0.001x_A S$

The yield equation is  $y_p = 0.1 + 0.3x_A + 0.02S + 0.001x_A S$

$y_p$  = paun-mol hasil/paun-mol suapan.

$y_p$  = lb-mol product/lb-mol feed.

- [a] Berikan fungsi keuntungan dengan  $x_A$  dan S.  
Formulate the profit function in terms of  $x_A$  and S.
- [b] Maksimumkan fungsi keuntungan dengan mana-mana kaedah.  
Maximize the profit function using any method.
- [c] Tunjukkan samada fungsi keuntungan adalah cekung atau cembong dan berikan sebabnya.  
Show whether the profit function is concave or convex and why?

(25 markah)

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$$\max G = 50y_p - (0.1 + 0.5x_A + x_A^2) - (0.2 + 0.035S + 0.01S^2)$$

5. Fungsi objektif berikut telah diperolehi  
*The following objective function was obtained*

$$f = 3x_1 + x_2 + x_3$$

yang diperlukan untuk memaksimumkan fungsi objektif berdasarkan kepada:  
*it is required to maximize the objective function subject to:*

$$x_1 - 2x_2 + x_3 \leq 11$$

$$-4x_1 + x_2 + 2x_3 \geq 3$$

$$2x_1 - x_3 = -1$$

$$x_1, x_2, x_3 \geq 0$$

- [a] Dengan menggunakan kaedah Simplex, terangkan langkah-langkah penyelesaian persamaan-persamaan itu.  
*Using the Simplex method explain steps of solution of these equations.*
- [b] Terangkan samada penyelesaian asas yang bersesuaian untuk masalah ini.  
*Explain whether a basic feasible solution exist for this problem.*

Tunjukkan segala maklumat terperinci penyelesaian anda. Anda tidak perlu mengira sebarang penyelesaian yang sesuai.

*Show all details of your solution. You don't have to calculate the feasible solution.*

(25 markah)

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**LAMPIRAN**

## Solution of non-linear Equations

## 1. Newton's Method:

$$f(x) = f(x_0) [x - x_0] + f'(x_0)$$

$$x_{i+1} = x_i - \frac{f(x_i)}{f'(x_i)}$$

## 2. Secant Method:

$$x_{i+1} = x_i - \frac{f(x_i)(x_i - x_{i-1})}{f(x_i) - f(x_{i-1})}$$

## 3. Regula Falsi Method:

$$x_{i+1} = x_i - f(x_i) \frac{(x_i - x_{i-1})}{f(x_i) - f(x_{i-1})}$$

## 4. Illionis Method:

$$x_{i+2} = x_{i+1} - f(x_{i+1}) \frac{x_{i+1} - x_{i-1}}{f(x_{i+1}) - \frac{1}{2}f(x_{i-1})}$$

## 5. Solution of n non-linear Equations

$$\sum_{i=1}^n \left. \frac{\partial f_k(x)}{\partial x_i} \right|_{x^{(j)}} d_i^{(j)} = -f_k[x^{(j)}]$$

$$k = 1, 2, \dots, n$$

$$x_i^{(j+1)} = x_i^{(j)} + d_i^{(j)}$$

$$i = 1, 2, \dots, n$$

$j$  = number of iteration.

...2/-

## Solution of ODE

1. Explicit Euler method:

$$y_{i+1} = y_i + \Delta x f(x_i, y_i)$$

2. Runge-Kutta Second Order Method:

$$y_{i+1} = y_i + \frac{1}{2}(k_1 + k_2)$$

where

$$k_1 = f(x_i, y_i)$$

$$k_2 = f\left(x_i + \frac{1}{2}hx, y_i + k_1\right)$$

3. Implicit Methods:

- 3.1 Euler

$$y_{i+1} = y_i + \Delta x f(x_{i+1}, y_{i+1})$$

- 3.2 Trapezoidal

$$y_{i+1} = y_i + \frac{\Delta x}{2} [f(x_i, y_i) + f(x_{i+1}, y_{i+1})]$$

4. Systems of Coupled ODE'S

Runge-Kutta Second Order:

$$y_{i,j+1} = y_{i,j} + \frac{1}{2}(k_{1,i,j} + k_{2,i,j})$$

$$k_{1,i,j} = f_i(x_j, y_{1,j}, y_{2,j}, \dots, y_{n,j})$$

$$k_{2,i,j} = f_i\left(x_j + \frac{\Delta x}{2}, y_{i,j} + \frac{\Delta x}{2} k_{1,1,j}, \dots, y_{n,j} + \frac{\Delta x}{2} k_{1,n,j}\right)$$

Table Geometric interpretation of a quadratic function

Case	Eigenvalue Relations	Signs		Types of contours	Geometric interpretation	Character of center of contours	Figure
		$x_1$	$x_2$				
1	$x_1 = x_2$	-	-	Circles	Circular hill	Maximum	4.12
2	$x_1 = x_2$	+	-	Circles	Circular valley	Minimum	4.12
3	$x_1 > x_2$	-	-	Ellipses	Elliptical hill	Maximum	4.13
4	$x_1 > x_2$	+	-	Ellipses	Elliptical valley	Minimum	4.13
5	$ x_1  =  x_2 $	+	-	Hyperbolas	Symmetrical saddle	Saddle point	4.14
6	$ x_1  =  x_2 $	-	-	Hyperbolas	Symmetrical saddle	Saddle point	4.14
7	$\alpha_1 > \alpha_2$	+	-	Hyperbolas	Elongated saddle	Saddle point	4.14
8	$\alpha_2 = 0$	-		Straight lines	Stationary* ridge	None	4.15
9	$\alpha_2 = 0$	+		Straight lines	Stationary* valley	None	4.15
10	$\alpha_2 = 0$	-		Parabolas	Rising ridge*†	At $\infty$	4.16
11	$\alpha_2 = 0$	-		Parabolas	Falling valley*†	At $\infty$	4.16

\* These are "degenerate" surfaces.

† The condition of rising or falling must be evaluated from the linear terms in  $f(x)$ .

