UNIVERSITI SAINS MALAYSIA

First Semester Examination Academic Session of 2003/2004

September/October 2003

EBB 512/3 - Phase Diagram and Phase Equilibra

Time: 3 hours

Please ensure that this paper consists of TEN printed pages before you proceed with the examination.

This paper contains SEVEN questions.

Answer any FIVE questions. If a candidate answer more than five questions, only the first five answered will be examined and awarded marks.

Answer to any question must start on a new page.

All questions can be answered in Bahasa Malaysia or English.

1. Use the phase diagram in Figure 1 to answer the following questions.

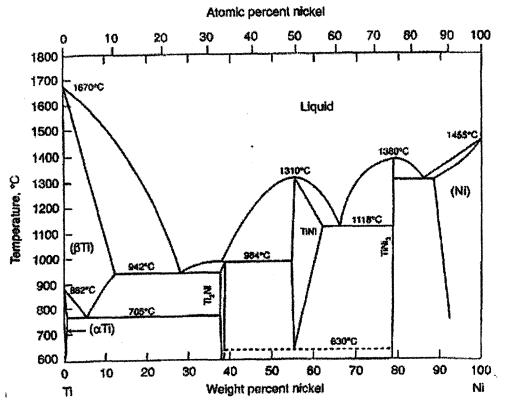


Figure 1

(a) List the composition and temperature of ALL of the invariant points, congruent points, monotectics and miscibility gaps and what kind of reaction it is.

(30 marks)

(b) Given what you know the binary system in the phase diagram, how many free energy curves would be required to FULLY determine the phase diagram? Why?

(30 marks)

(c) Based on the phase diagram, would you expect Ni and Ti to have the same crystal structure? Why or why not?

(40 marks)

...3/-

2. [a] (i) A eutectoid steel is cooled at a steady rate from 727°C to 200°C in exactly 1 day. Superimpose this cooling curve on the TTT diagram of Figure 2 (ii) From the result of your plot for part (a), determine at what temperature a phase transformation would first be observed. (iii) What would be the first phase to be observed?

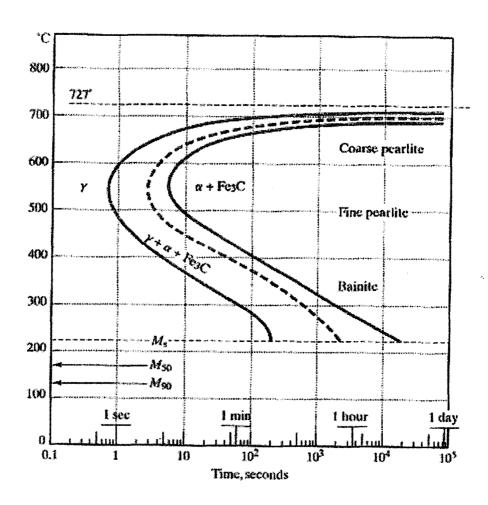


Figure 2

(50 marks)

[b] Derive an expression for r_c as a function of σ , the surface energy per unit area of the nucleus, and ΔG_v , the volume energy reduction per unit volume. Recall that the area of a sphere is $4\pi r^2$, and its volume is $\left(\frac{4}{3}\right)\pi r^3$.

The work of formation, W, for a stable nucleus is the maximum value of the net energy change (at r_c). Derive and expression for W in terms of σ and ΔG_v . *r is the critical radius.

(50 marks)

[a] The ideal stoichiometry of the γ phase in the Al-Mg system is Al₁₂Mg₁₇.
(i) What is the atomic percentage of excess Al in the most aluminum-rich γ composition at 450°C? (ii) What is the atomic percentage of excess Mg in the most magnesium-rich γ composition at 437°C? (See Figure 3)

Atomic weight: AI: 26.98

: Mg: 24.31

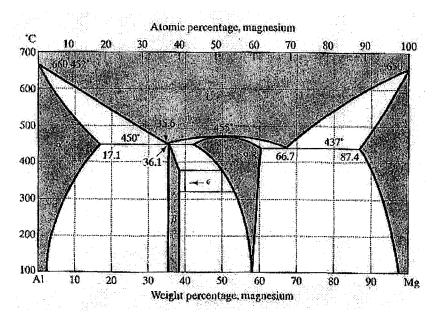


Figure 3

(75 marks)

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[b] Suppose that you have a crucible containing 1 kg of an alloy of composition 90 wt % Sn-10 wt % Pb at a temperature of 184°C. How much Sn would you have to add to the crucible to completely solidify the alloy without changing the system temperature? (Figure 4)

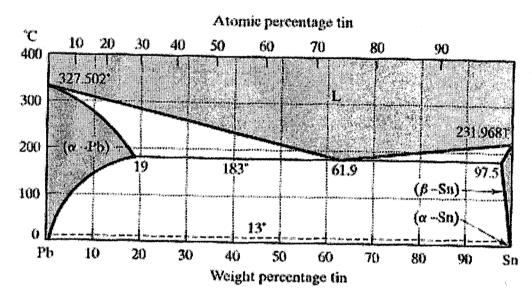


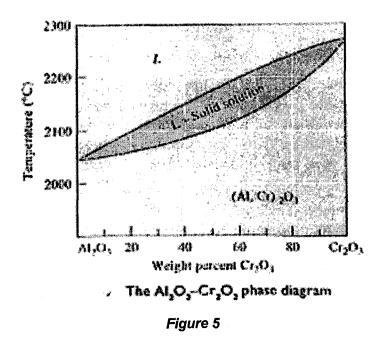
Figure 4

(25 marks)

4. [a] One way to repair dents in a metal is to wipe a partly liquid-partly solid material into the dent, then allow this filler material to solidify. For our application, the wiping material should have the following specifications: (1) a melting temperature below 230°C, (2) a tensile strength in excess of 6000 psi, (3) be 60% to 70% liquid during application, and (4) the lowest possible cost. Design an alloy and repair procedure that will meet these specifications. Consider the Pb-Sn system for your answer. (Refer again Figure 4)

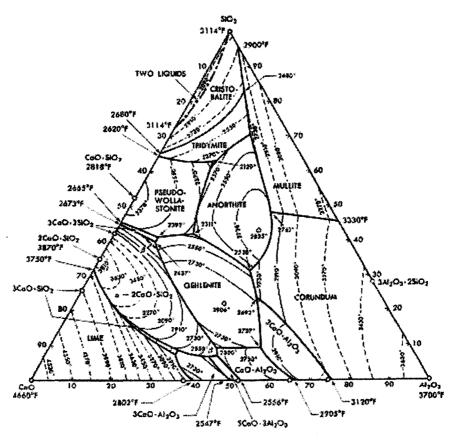
(50 marks)

[b] One method to improve the fracture toughness of a ceramic materials is to reinforce the ceramic matrix with ceramic fibers. A materials designer has suggested that Al₂O₃ could be reinforced with 25% Cr₂O₃ fibers, which would interfere with the propagation of any cracks in the alumina. The resulting composite is expected to operate under load at 2000°C for several months. Criticize the appropriateness of this design.



(50 marks)

5. A 500 kg batch of impure clay has been fired to give a mixture containing 68% SiO₂, 28% Al₂O₃ and 4% CaO. It is mixed with calcined limestone (95% CaO, 5% SiO₂) to make portland cement.



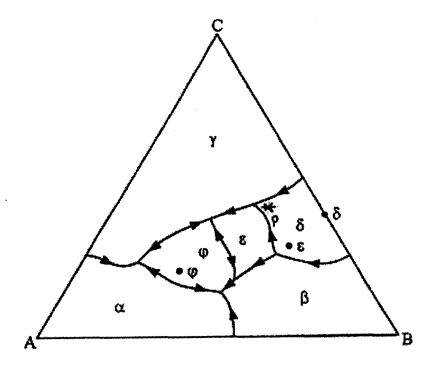
The CaO-Al₂O₃-SiO₂ liquidus diagram. The isotherms may be compared with elevation contours on a topographic map. (U.S. Steel, Making, Shaping and Treating of Steel, 7th ed. Pittsburgh: U.S. Steel Corp., 1957)

- (a) How many kilograms of calcined limestone are required if the CaO/SiO_2 ratio of the cement is to be $\frac{7}{3}$?
- (b) What would be the composition and the liquidus temperature of the mixture?

(100 marks)

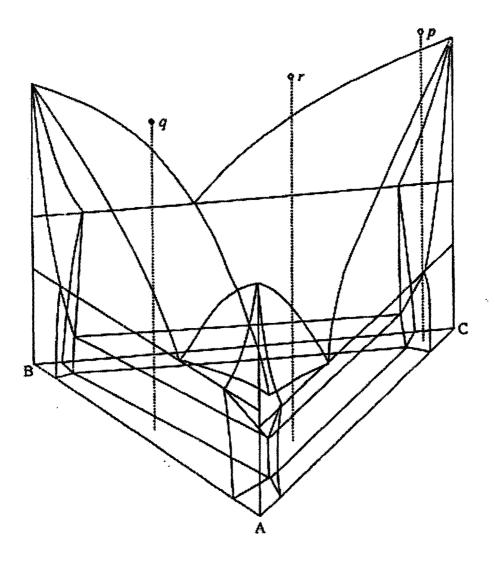
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6. Referring to the figure given below:



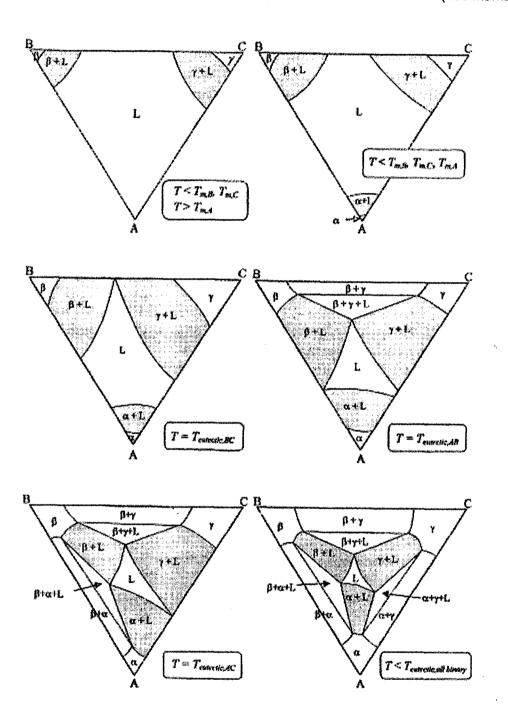
- (a) Determine whether binary compound δ and ternary compounds ϵ and ϕ melt congruently or incongruently.
- (b) Draw all Alkemade lines.
- (c) Determine the crystallization path of a liquid of composition p. (100 marks)

7. ABC ternary system forms three binary eutectics and a ternary eutectics as shown below. Discuss equilibrium cooling paths for the overall compositions p, q and r indicated in the diagram. Discuss also the change in microstructure that should occur during cooling.



Shown in the following are the isothermal sections at a number of different temperatures of the hypothetical system ABC given above :

(100 marks)



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